# Calyxaprenols A-D, New Merohexaprenoid Metabolites from the Marine Sponge *Calyx* sp.

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Abstract – Four new merohexapenoids named calyxaprenols A-D (1 - 4), together with the known compound haliclotriol A (5), have been isolated from the marine sponge *Calyx* sp. which was collected from the southwest islands of Palau. Based on comprehensive spectroscopic analyses, calyxaprenols A (1) and B (2) were determined to be pentacyclic hexaprenoids that are appended to a glycolic acid-substituted phenol moiety, whereas calyxaprenols C (3) and D (4) possess a tricyclic hexaprenoid skeleton joined to a hydroquinone ring. Identification of new merohexaprenoids from a *Calyx* sponge expands the known taxonomic distribution of this sparsely distributed class of marine metabolites and increases the chemical diversity described for this genus of marine sponge.

Keywords – Calyx sp., marine sponge, merohexaprenoid, calyxaprenol

### Introduction

Marine merohexapenoids are a relatively rare family of metabolites derived from mixed terpenoid and polyketide biogenesis which have demonstrated a variety of important biological activities. Since the isolation of the mokupalides,<sup>1</sup> the first sponge-derived head-to-tail type triterpenes, approximately 30 head-to-tail merotriterpenoids which incorporate benzoic subunits appended to a variety of triterpene scaffolds have been identifed from diverse sponge sources.<sup>2</sup> These compounds include the shaagrockols, toxicols, and toxiusol from the Red Sea sponge Toxiclona toxius,<sup>3,4</sup> along with akaterpin, isoakaterpin, ilhabelanol, and ilhabrene from a Callyspongia sp. sponge.5,6 Adociasulfates 1-12 and adociaquinol,7-11 haliclotriols A and B,<sup>12</sup> and halicloic acids A and B,<sup>13</sup> have been reported from Haliclona (aka Adocia) sp. sponges, while adociasulfates 13 and 14 were isolated from the sponge Cladocroce aculeata.14 These merohexapenoid metabolites have been reported to exhibit potent and diverse biological properties

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including inhibiting the enzymatic activities of phosphatidylinositol-specific phospholipase C (PI-PLC),<sup>5</sup> adenosine phosphoribosyl transferase (APRT),<sup>6</sup> indoleamine 2,3dioxygenase,<sup>13</sup> and HIV reverse transcriptase,<sup>15</sup> as well as modulation of the H<sup>+</sup>-ATPase protein pump,<sup>8</sup> and reduction of the microtubule-stimulated ATPase activity of kinesin.<sup>14</sup>

In conjunction with long-term and ongoing natural products discovery efforts at the NCI,<sup>16,17</sup> an extract of the marine sponge Calvx sp. which had been collected around Palau, was selected for detailed chemical study. Prior investigations of *Calyx* sponges have provided a variety of new metabolites including unusual steroids,18-21 fatty acids,<sup>22,23</sup> alkaloids,<sup>24-26</sup> and diphenylbutanoids.<sup>27</sup> Chromatographic separation of the Calyx extract provided four new merohexaprenoid metabolites that were named calyxaprenols A-D (1-4), along with the known compound haliclotriol A (5).12 Based on extensive NMR and MS analyses, calyxaprenols A (1) and B (2) were determined to consist of a core hexacyclic terpenoid ring system attached to a glycolic acid moiety at the benzylic position. The triterpene component of calyxaprenols C (3) and D (4) was established to have a bicyclic rearanged drimane skeleton and a prenyl-substituted cyclohexene ring that are linked via an ethylene bridge, and this C30 scaffold was joined with a hydroquinone moiety (Fig. 1).

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Fig. 1. Structures of calyxaprenols A-D (1 - 4) and haliclotriol A (5) isolated from the marine sponge Calyx sp.

# Experimental

General Experimental Procedures - Optical rotations were measured on a Rudolph research analytical AUTOPOL IV automatic polarimeter, IR spectra were recorded with a Bruker ALPHA II FT-IR spectrometer, and UV spectra were measured with a Thermo Scientific Nanodrop 2000C spectrophotometer. NMR spectra were obtained with a Bruker Avance III NMR spectrometer equipped with a 3 mm cryogenic probe and operating at 600 MHz for <sup>1</sup>H and 150 MHz for <sup>13</sup>C. Spectra were calibrated to residual solvent signals at  $\delta_{\rm H}$  3.31 and  $\delta_{\rm C}$ 49.0 in MeOH- $d_4$ . All 2D NMR experiments were acquired with nonuniform sampling (NUS) set to 50% or 25%. HRESIMS data were acquired on an Agilent Technology 6530 Accurate-mass Q-TOF LC/MS. Highperformance liquid chromatography (HPLC) was performed using a Varian ProStar 215 solvent delivery module equipped with a Varian Prostar 320 UV-Vis detector, operating under Star 6.41 chromatography workstation software.

**Animal material** – Samples of the marine sponge *Calyx* sp. were collected by scuba at a depth of 12 feet, in the southwest islands of Palau in 2008, and kept frozen until extraction. The collection was carried out by the Coral Reef Research Foundation, under contract with the National Products Branch, U.S. National Cancer Institute. A voucher specimen (voucher ID # 0CDN 9731) was deposited at the Smithsonian Institution, Washington, D. C.

**Extraction and isolation** – The frozen sponge sample (127 g wet wt.) was ground and extracted using the standard NCI method for marine samples to provide 4.4 g of organic solvent extract (NSC C029729).<sup>28</sup> A 3.0 g aliquot of the *Calyx* sp. organic extract was subjected to C<sub>18</sub> reversed-phase flash column chromatography using a step gradient elution with 100% hexane (fraction A, 348 mg), 100% CH<sub>2</sub>Cl<sub>2</sub> (fraction B, 217 mg), 100% EtOAc (fraction C, 75 mg), 100% acetone (fraction D, 397 mg), and 100% MeOH (fraction E, 1.1 g). Fraction D was further separated by preparative reversed-phase HPLC

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(Agilent Dynamax C18 column, 21.4 mm × 250 mm; 9.0 mL/min, CH<sub>3</sub>CN-H<sub>2</sub>O gradient from 30:70 to 100:0), yielding 7 peaks rich in secondary metabolites. Purification of sub-fraction 1 was accomplished by semipreparative HPLC (Phenomenex Luna C18 column, 10 mm × 250 mm; 2.0 mL/min, MeCN-H<sub>2</sub>O gradient from 35:65 to 80:20) to yield calyxaprenol C (3, 1.1 mg), as an amorphous solid. Fraction E was separated by preparative reversed-phase HPLC (Agilent Dynamax C18 column, 21.4 mm × 250 mm; 9 mL/min, MeCN-H<sub>2</sub>O gradient from 20:80 to 100:0), yielding 10 peaks rich in secondary metabolites. Further purification of the sub-fraction 1 by semi-preparative HPLC (Phenomenex Luna C18 column,  $10 \text{ mm} \times 250 \text{ mm}$ ; 3.0 mL/min, MeCN-H<sub>2</sub>O gradient from 50:50 to 100:0) afforded calyxaprenol D (4, 0.5 mg), while fractionation of sub-fraction 6 by the same semipreparative HPLC method but with a MeCN-H2O gradient from 55:45 to 100:0 yielded, in order of elution, calyxaprenol A (1, 2.0 mg), calyxaprenol B (2, 1.8 mg), and haliclotriol A (5, 4.2 mg) as amorphous solids.

**Calyxaprenol A (1)** – white, amorphous powder;  $[\alpha]_D^{25}$ +7.5 (*c* 0.11, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 215 (3.86), 271 (2.84) nm; IR (film)  $\nu_{max}$  3412, 2932, 1632, 1485, 1233, 1022 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; HRESIMS *m/z* 625.4108 [M-H]<sup>-</sup> (calcd for C<sub>38</sub>H<sub>57</sub>O<sub>7</sub>, 625.4104).

**Calyxaprenol B (2)** – white, amorphous powder;  $[\alpha]_D^{25}$ +6.9 (*c* 0.15, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 213 (4.19), 272 (3.96) nm; IR (film)  $\nu_{max}$  3339, 2938, 1649, 1483, 1240, 1020 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 2; HRESIMS *m/z* 639.4270 [M-H]<sup>-</sup> (calcd for C<sub>39</sub>H<sub>59</sub>O<sub>7</sub>, 639.4261).

**Calyxaprenol C (3)** – colorless, amorphous solid;  $[\alpha]_D^{25}$  -12.1 (*c* 0.08, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 211 (3.48), 277 (2.81) nm; IR (film)  $\nu_{max}$  3450, 2930, 1642, 1597, 1243, 1040 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 2; HRESIMS *m/z* 575.4075 [M+Na]<sup>+</sup> (calcd for C<sub>36</sub>H<sub>56</sub>O<sub>4</sub>Na, 575.4076).

**Calyxaprenol D (4)** – colorless, amorphous solid;  $[\alpha]_D^{25}$  -10.9 (*c* 0.12, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 210 (3.12), 274 (2.74) nm; IR (film)  $\nu_{max}$  3402, 2935, 1677, 1592, 1242, 1045 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 2; HRESIMS *m/z* 557.3973 [M+Na]<sup>+</sup> (calcd for C<sub>36</sub>H<sub>54</sub>O<sub>3</sub>Na, 557.3971).

#### **Result and Discussion**

The molecular formula of calyxaprenol A (1) was determined by HRESIMS to be  $C_{38}H_{58}O_{7}$ , with 10 degrees of unsaturation. Detailed examination of the <sup>1</sup>H and <sup>13</sup>C



**Fig. 2.** Selected COSY and HMBC correlations for calyxaprenol A (1).

NMR data (Table 1) aided by 2D NMR experiments, revealed this compound possessed a 7/6/6/5/6-hexacyclic fused ring system (rings A-F) and an additional prenylated side chain. The partial structure of the indane ring system (rings E and F) was elucidated from ortho coupled aromatic protons H-3' ( $\delta_{\rm H}$  6.53)/H-4' ( $\delta_{\rm H}$  6.90) and HMBC correlations between H-1 ( $\delta_{\rm H}$  2.61)/C-2, C-3, C-1', C-2', and C-6', H-4' ( $\delta_{\rm H}$  6.90)/C-2', C-6', and C-7', and H-7' ( $\delta_{\rm H}$ 5.52)/C-4', C-5', C-6', and C-8' (Fig. 2). Collectively these correlations supported substitution of a hydroxy group at C-2' ( $\delta_{\rm C}$  154.7) and a glycolic acid residue at C-5' ( $\delta_{\rm C}$ 126.0) in the tetrasubstituted aromatic ring. A partial structure of the decalin system (rings C and D) was established by COSY correlations between H2-8/H2-9, H2-9/H-10, H<sub>2</sub>-4/H<sub>2</sub>-5, and H<sub>2</sub>-5/H-6, and HMBC correlations from H<sub>3</sub>-26 ( $\delta_{\rm H}$  1.09)/C-2, C-6, C-7, and C-8. Fusion of the 5-membered ring (E ring) with the cyclohexane ring (D ring) was supported by HMBC correlations from the singlet methyl H<sub>3</sub>-25 ( $\delta_{\rm H}$  1.16) to C-2, C-6, C-7, and C-8. In the same manner, assignment of the fused oxepane-cyclohexane rings (A and B rings) was confirmed by COSY correlations between H<sub>2</sub>-12/H<sub>2</sub>-13, H<sub>2</sub>-13/H-14, H2-16/H2-17, and H2-17/H-18, and HMBC correlations from H<sub>3</sub>-27 ( $\delta_{\rm H}$  0.92)/C-6, C-10, C-11, and C-12, and H<sub>3</sub>-28 ( $\delta_{\rm H}$  0.84)/C-10, C-14, C-15, and C-16 (Fig. 2). Additional support for assignment of the oxepane ring (A ring) was provided by HMBC correlations from the singlet methyl H<sub>3</sub>-29 ( $\delta_{\rm H}$  1.17) to the C-18 ( $\delta_{\rm C}$  75.8) oxymethine, the C-19 ( $\delta_{\rm C}$  81.6) oxygenated quaternary carbon, and the C-20 ( $\delta_{\rm C}$  43.1) methylene. The ether linkage in ring A was secured by an HMBC cross-peak from H-14 ( $\delta_{\rm H}$  3.53) to C-19 ( $\delta_{\rm C}$  81.6). Thus, the hexacyclic ring system of 1 was established. The terminal isoprene unit of the molecule was defined by COSY correlations between H<sub>2</sub>-20/H<sub>2</sub>-21 and H<sub>2</sub>-21/H<sub>2</sub>-22, as

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Desident	S (torra)		S (town)	$\frac{2}{2}$
Position	$\partial_{\rm C}$ (type)	$\partial_{\rm H}$ mult (J in Hz)	$\partial_{\rm C}$ (type)	$\partial_{\rm H}$ mult (J in Hz)
1	25.4, CH <sub>2</sub>	2.61, dd (14.6, 6.2)	$25.3, CH_2$	2.61, dd (14.4, 6.2)
•		2.42, t (14.6)	(5.5. OU	2.42, t (14.4)
2	65.5, CH	1.70, m	65.5, CH	1.70, m
3	49.8, C	• • •	49.8, C	• • •
4	$40.4, CH_2$	2.39, m	$40.4, CH_2$	2.39, m
-		1.90, m		1.90, m
5	18./, CH <sub>2</sub>	1.63, m	18.7, CH <sub>2</sub>	1.63, m
,		1.54, m		1.58, m
6	62.5, CH	0.94, m	62.5, CH	0.95, m
7	38.4, C	1 51	38.4, C	1.50
8	$43.3, CH_2$	1./I, m	$43.3, CH_2$	1.72, m
0	10.0 CU	1.15, m	10.0 CH	1.14, m
9	19.8, CH <sub>2</sub>	1.72, m	19.8, CH <sub>2</sub>	1.71, m
10	50.0 CH	1.45, m	50.0 CH	1.42, m
10	58.8, CH	0.89, m	58.8, CH	0.90, m
11	38.0, C	1.60	38.0, C	1.50
12	39.9, CH <sub>2</sub>	1.69, m	39.9, CH <sub>2</sub>	1.70, m
12	<b>2</b> 0 1 CU	0.98, m	<b>20 1 CI</b>	0.99, m
13	28.1, $CH_2$	1.57, m	28.1, $CH_2$	1.56, m
1.4	<b>70 0</b> (11	1.37, m	<b>70.0</b> GU	1.38, m
14	78.2, CH	3.53, dd (11.4, 4.6)	78.2, CH	3.53, dd (11.5, 4.6)
15	42.1, C	1.50	42.1, C	1.50
16	36.7, CH <sub>2</sub>	1.50, m	36.7, CH <sub>2</sub>	1.52, m
17	27.0.011	1.43, m	27.0.011	1.44, m
1/	$27.0, CH_2$	1.95, m	$27.0, CH_2$	1.95, m
10	75.0 011	1.66, m	75.0.011	1.66, m
18	/5.8, CH	3./8, d (6./)	/5.9, CH	3.76, d (6.8)
19	81.6, C	1.26	81.5, C	1.26
20	$43.1, CH_2$	1.36, m	$42.9, CH_2$	1.36, m
21	20 4 CH	1.29, m	19.7 CH	1.29, m
21	$20.4, CH_2$	1.45, m	$18.7, CH_2$	1.55, m
22	45 4 CH	1.34, m	41.2 CH	1.42, m
22	$43.4, C\Pi_2$	1.43, III 1.20 m	$41.3, CH_2$	1.40, III 1.42 m
22	71 4 C	1.39, 111	76 A C	1.43, 111
23	71.4, C	1 19 0	70.4, C	1 15 a
24	$29.2, CH_3$	1.16, 8	$23.4, CH_3$	1.15, 8
25	22.4, СП <sub>3</sub> 17.7, СШ	1.10, 8	$22.4, CH_3$	1.10, 8
20	17.7, СП <sub>3</sub>	1.09, 8	$1/./, CH_3$	1.09, 8
27	$10.8, CH_3$	0.92, 8	$10.8, CH_3$	0.92, 8
20	$14.2, CH_3$	0.84, 8	$14.2, CH_3$	0.84, 8
29	$19.4, CH_3$	1.17, 8	$19.4, CH_3$	1.17,8
30 17	$29.2, CH_3$	1.16, 8	$23.4, CH_3$	1.15, 8
1	129.0, C		129.0, C	
2	154.7, C	(52 + 1)(9 - 2)	154.8, C	(52 + 1)(2)
5	114.2, CH	(0.33, 0, (0.3))	114.2, CH	0.33, 0(0.3)
4 51	127.5, CH	0.90, a (8.3)	127.5, CH	0.90, a (8.3)
5 6'	120.0, C		120.0, C	
0	155.2, U	5.50 -	155.2, U	5.50 ~
/ Q'	175 0 C	5.52, 8	175 0 C	5.52, 8
o	1/3.9, C		1/3.9, U	2.10

Table 1. <sup>13</sup>C NMR (150 MHz) and <sup>1</sup>H NMR (600 MHz) Data for Calyxaprenols A (1) and B (2) in MeOH-d<sub>4</sub>



Fig. 3. Selected ROESY correlations for calyxaprenol A (1).

well as HMBC correlations from the *gem*-dimethyl group H<sub>3</sub>-24 and H<sub>3</sub>-30 ( $\delta_{\rm H}$  1.18) to the C-23 ( $\delta_{\rm C}$  71.4) oxygenbearing quaternary carbon. The 6H singlet for these two methyl groups showed HMBC correlations to the C-24/C-30 ( $\delta_{\rm C}$  29.2) carbon signal, which confirmed that they were germinal. Thus, the planar structure of calyxaprenol A (**1**) was established.

The relative configuration of 1 was elucidated from 2D ROESY NMR data (Fig. 3) and comparison with the NMR data for haliclotriol A (5).<sup>12</sup> The all trans fused A-B-C-D-E ring system was established by ROESY crosspeaks between  $H_3$ -28/H-13 $\beta$  and  $H_3$ -27,  $H_3$ -27/H $_3$ -26, and H<sub>3</sub>-26/H-9β, H-5β and H-2 which indicated all ring junction methyl groups were located on the  $\beta$  face of the molecule, with the exception of H<sub>3</sub>-25. Additional ROESY cross-peaks at H-6/H-10 and H-4 $\alpha$ , H-10/H-14, and H-14/ H<sub>3</sub>-29 were supportive of  $\alpha$ -orientations for the A/B/C/D ring junction protons, as well as the H<sub>3</sub>-29 methyl group. Around the D/E ring juncture, the carbon chemical shifts of C-1 ( $\delta_{\rm C}$  25.4), C-2 ( $\delta_{\rm C}$  65.5), C-3 ( $\delta_{\rm C}$  49.8), C-4 ( $\delta_{\rm C}$ 40.4), C-7 ( $\delta_{\rm C}$  38.4), C-25 ( $\delta_{\rm C}$  22.4) and C-26 ( $\delta_{\rm C}$  17.7) were virtually identical to those of haliclotriol A (5).<sup>12</sup> These similarities imply that 1 and 5 have the same boat and envelop conformations in the D and E rings, respectively. This relative relationship was further supported by ROESY correlations between H<sub>3</sub>-26/H-2 and H<sub>3</sub>-25/H-4 $\beta$ . Finally, the  $\beta$ -orientation of hydroxy methine proton H-18 was confirmed by H-18/H-20 and H-17ß ROESY correlations. Thus, the relative configurations were assigned to be 2S\*, 3S\*, 6R\*, 7S\*, 10S\*, 11S\*, 14R\*, 15R\*, 18R\*, and 19S\*. Attempts to obtain additional stereochemical information were unsuccessful as efforts to generate MTPA esters of 1 resulted in decomposition. This completed the structural assignment of calyxaprenol A (1).

The molecular formula of calyxaprenol B (2) was established as  $C_{39}H_{60}O_7$  through HRESIMS analysis, which differed by addition of  $CH_2$  from the molecular formula of 1. Based on the close similarity of its <sup>1</sup>H and

 $^{13}$ C NMR data (Table 1), the planar structure of **2** was concluded to be the same as that of 1, and all of the subsequent 2D NMR data supported this assignment. The presence of a methoxy group in place of a hydroxy group in 2 was apparent from an additional three proton singlet at  $\delta_{\rm H}$  3.18 that showed an HSQC correlation to a new carbon signal at  $\delta_{\rm C}$  49.1. The deshielded chemical shift of the C-23 quaternary carbon ( $\delta_{\rm C}$  76.4 in **2** and  $\delta_{\rm C}$  71.4 in **1**) suggested the methoxy group was substituted at C-23, which was confirmed by HMBC cross-peaks from OCH<sub>3</sub>, H<sub>3</sub>-24, and H<sub>3</sub>-30 to C-23. The ROESY data of 2 exhibited very similar proton-proton cross-peaks to those of 1, which allowed assignment of all trans fused A-B-C-D-E ring junctions and an anti orientation between H<sub>3</sub>-29 and H-18. Thus, the relative configuration of calyxaprenol B (2) was assigned as 2S\*, 3S\*, 6R\*, 7S\*, 10S\*, 11S\*, 14R\*, 15R\*, 18R\*, and 19S\*.

The molecular formula of calyxaprenol C (3) was determined by HRESIMS measurements to be C<sub>36</sub>H<sub>56</sub>O<sub>4</sub>. with 9 degrees of unsaturation. Initial inspection of the <sup>13</sup>C NMR data (Table 2) revealed 10 carbons in the aromatic/olefinic region of the spectrum and 26 aliphatic carbon signals. These data suggested that calyxaprenol C (3) consisted of an aromatic ring joined to a tricyclic hexaprenoid which had two olefinic bonds. Analysis of the <sup>1</sup>H and <sup>13</sup>C NMR data revealed the aromatic portion of the molecule (ring D) had signals appropriate for a monosubstituted hydroquinone with C-4',  $\delta_{\rm C}$  113.2,  $\delta_{\rm H}$ 6.41 (1H, dd, J = 8.4, 3.1 Hz); C-3',  $\delta_{\rm C}$  116.1,  $\delta_{\rm H}$  6.57 (1H, d, J = 8.4 Hz); C-6',  $\delta_{\rm C}$  116.7,  $\delta_{\rm H}$  6.49 (1H, d, J = 3.1 Hz), C-1'  $\delta_{\rm C}$  129.8, C-2'  $\delta_{\rm C}$  148.3, and C-5'  $\delta_{\rm C}$  150.4, respectively. The position of the aromatic hydroxy groups was established by the following HMBC correlations: H-3'/C-1' and C-5', H-4'/C-2' and C-6', and H-6'/C-2' and C-4'.

A decalin ring system (B and C rings) in 3 was assigned from characteristic <sup>1</sup>H and <sup>13</sup>C NMR signals and the COSY and HMBC data (Fig. 4). A connection between the hydroquinone ring and the hexaprenoid scaffold was established by HMBC correlations from the allylic and benzylic methylene proton signals of H<sub>2</sub>-1  $\delta_{\rm H}$  3.33 (1H, d, J=17.5 Hz) and  $\delta_{\rm H}$  3.18 (1H, d, J=17.5 Hz) to three aromatic carbons (C-1', C-2', and C-6'), two olefinic quaternary carbons C-2 ( $\delta_{\rm C}$  139.0) and C-3 ( $\delta_{\rm C}$  129.5), and a fully substituted aliphatic carbon C-7 ( $\delta_{\rm C}$  40.2). Two deshielded signals at  $\delta_{\rm C}$  139.0 and 129.5 were assigned to the C-2/C-3 tetrasubstituted olefin by HMBC correlations from H<sub>3</sub>-25 ( $\delta_{\rm H}$  1.50) to C-2, C-3, and the C-4 methylene signal ( $\delta_{\rm C}$  34.8) and from H<sub>3</sub>-26 ( $\delta_{\rm H}$  1.05) to C-2, C-6, C-7, and C-8. Furthermore, the partial structure of the bicyclic ring system was confirmed by COSY couplings

		3		4
Position	$\delta_{\rm C}$ (type)	$\delta_{\rm H}$ mult (J in Hz)	$\delta_{\rm C}$ (type)	$\delta_{\rm H}$ mult (J in Hz)
1	27.8, CH <sub>2</sub>	3.33, d (17.5)	27.2, CH <sub>2</sub>	2.65, dd (15.3, 9.6)
		3.18, d (17.5)		2.48, br d (15.3)
2	139.0, C		56.3, CH	2.36, br d (9.6)
3	129.5, C		137.2, C	
4	34.8, CH <sub>2</sub>	2.19, m	122.6, CH	5.33, br s
		2.08, m		
5	20.3, CH <sub>2</sub>	1.76, m	24.3, CH <sub>2</sub>	1.96, m
		1.57, m		1.93, m
6	55.4, CH	1.35, m	54.1, CH	1.32, m
7	40.2, C		38.2, C	
8	37.4, CH <sub>2</sub>	1.58, m	40.8, CH <sub>2</sub>	1.95, m
		1.03, m		1.19, m
9	20.1, CH <sub>2</sub>	1.54, m	19.9, CH <sub>2</sub>	1.49, m
		1.32, m		1.43, m
10	38.1, CH <sub>2</sub>	1.77, m	38.1, CH <sub>2</sub>	1.75, m
		0.90, m		0.96, m
11	37.7, C		36.8, C	
12	34.0, CH <sub>2</sub>	1.71, m	32.9, CH <sub>2</sub>	1.78, m
		1.24, m		1.15, m
13	24.2, CH <sub>2</sub>	2.03, m	25.7, CH <sub>2</sub>	1.48, m
		1.80, m		1.25, m
14	137.2, C		49.9, CH	1.79, m
15	128.6, C		137.8, C	
16	31.9, CH <sub>2</sub>	2.04, m	120.1, CH	5.26, br s
	, _	1.99, m	,	,
17	28.0, CH <sub>2</sub>	1.69, m	33.4, CH <sub>2</sub>	2.26, m
	, -	1.67, m	, <u> </u>	1.92, m
18	72.1. CH	3.66, dd (9.6, 4.4)	71.8. CH	3.70, dd (8.8, 6.7)
19	44.9, C		41.4, C	
20	39.1. CH <sub>2</sub>	1.61, m	36.4. CH <sub>2</sub>	1.69. m
	,	1.31. m		1.27. m
21	19.9. CH <sub>2</sub>	1.33, m	23.1. CH <sub>2</sub>	1.99. m
22	45.6. CH <sub>2</sub>	1.40, m	126.4. CH	5.14. t (6.7)
23	71.4. C	,	131.7. C	
24	29.1. CH <sub>3</sub>	1.15. s	25.9. CH <sub>3</sub>	1.69. s
25	,			,
26	20.3. CH <sub>2</sub>	1.50. s	22.5. CH <sub>2</sub>	1.44. s
27	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub>	1.50, s 1.05, s	22.5, CH <sub>3</sub>	1.44, s 0.95, s
	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>2</sub>	1.50, s 1.05, s 0.98, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>2</sub>	1.44, s 0.95, s 0.86, s
28	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub>	1.50, s 1.05, s 0.98, s 1.62, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub>	1.44, s 0.95, s 0.86, s 1.71, s
28 29	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub>	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub>	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s
28 29 30	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub> 29.2, CH <sub>3</sub>	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s 1.15, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub> 17.9, CH <sub>3</sub>	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s 1.65, s
28 29 30 1'	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub> 29.2, CH <sub>3</sub> 129.8, C	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s 1.15, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub> 17.9, CH <sub>3</sub> 132.3, C	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s 1.65, s
28 29 30 1' 2'	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub> 29.2, CH <sub>3</sub> 129.8, C 148.3, C	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s 1.15, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub> 17.9, CH <sub>3</sub> 132.3, C 148.8 C	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s 1.65, s
28 29 30 1' 2' 3'	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub> 29.2, CH <sub>3</sub> 129.8, C 148.3, C 116.1, CH	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s 1.15, s	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub> 17.9, CH <sub>3</sub> 132.3, C 148.8, C 116.7, CH	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s 1.65, s
28 29 30 1' 2' 3' 4'	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub> 29.2, CH <sub>3</sub> 129.8, C 148.3, C 116.1, CH	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s 1.15, s 6.57, d (8.4) 6.41, dd (8.4, 3.1)	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub> 17.9, CH <sub>3</sub> 132.3, C 148.8, C 116.7, CH	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s 1.65, s 6.55, d (8.5) 6.41 dd (8.5, 3.1)
28 29 30 1' 2' 3' 4' 5'	20.3, CH <sub>3</sub> 21.6, CH <sub>3</sub> 29.4, CH <sub>3</sub> 20.1, CH <sub>3</sub> 21.9, CH <sub>3</sub> 29.2, CH <sub>3</sub> 129.8, C 148.3, C 116.1, CH 113.2, CH 150.4, C	1.50, s 1.05, s 0.98, s 1.62, s 0.96, s 1.15, s 6.57, d (8.4) 6.41, dd (8.4, 3.1)	22.5, CH <sub>3</sub> 15.4, CH <sub>3</sub> 29.2, CH <sub>3</sub> 24.1, CH <sub>3</sub> 17.4, CH <sub>3</sub> 17.9, CH <sub>3</sub> 132.3, C 148.8, C 116.7, CH 113.5, CH	1.44, s 0.95, s 0.86, s 1.71, s 0.85, s 1.65, s 6.55, d (8.5) 6.41, dd (8.5, 3.1)

Table 2. <sup>13</sup>C NMR (150 MHz) and <sup>1</sup>H NMR (600 MHz) Data for Calyxaprenols C (3) and D (4) in MeOH-d<sub>4</sub>



Fig. 4. Selected COSY and HMBC correlations for calyxaprenol C (3).

between  $H_2$ -4/ $H_2$ -5,  $H_2$ -5/H-6,  $H_2$ -8/ $H_2$ -9, and  $H_2$ -9/ $H_2$ -10, and additional HMBC correlations between H<sub>3</sub>-27 ( $\delta_{\rm H}$ (0.98)/C-6, C-10, C-11, and a methylene signal at C-12 ( $\delta_{C}$ 34.0). An ethylene bridge connecting ring B and a cyclohexene ring (A ring) was supported by H<sub>2</sub>-12/H<sub>2</sub>-13 COSY cross-peaks and HMBC correlations between H<sub>2</sub>-13 and two quaternary olefinic carbons C-14 ( $\delta_{\rm C}$  137.2)/ C-15 ( $\delta_{\rm C}$  128.6), as well as a fully substituted aliphatic carbon at C-19 ( $\delta_{\rm C}$  44.9). The cyclohexene ring was further defined by HMBC correlations between H<sub>3</sub>-28 ( $\delta_{\rm H}$ 1.62)/C-14, C-15, and a methylene signal at C-16 ( $\delta_{\rm C}$ 31.9), and H<sub>3</sub>-29 ( $\delta_{\rm H}$  0.96)/C-14, an oxymethine carbon at C-18 ( $\delta_{\rm C}$  72.1), C-19, and a methylene carbon at C-20 ( $\delta_{\rm C}$ 39.1). A terminal alkyl chain was apparent from COSY correlations between H2-20/H2-21 and H2-21/H2-22, and HMBC correlations from H<sub>2</sub>-22 as well as a gem-dimethyl group H<sub>3</sub>-24 and H<sub>3</sub>-30 ( $\delta_{\rm H}$  1.15) to the C-23 ( $\delta_{\rm C}$  71.4) oxygen-bonded quaternary carbon. Molecular formula considerations required hydroxy groups at C-18 and C-23, thus the planar structure of calyxaprenol C (3) was established.

The relative stereochemistry of **3** was assigned from 2D ROESY data (Fig. 5). ROESY correlations from H<sub>3</sub>-26 to H-5 $\beta$  ( $\delta_{\rm H}$  1.57), H-1 ( $\delta_{\rm H}$  3.33), and H-12a ( $\delta_{\rm H}$  1.24), together with correlations from H-6 ( $\delta_{\rm H}$  1.35) to H-5 $\alpha$  ( $\delta_{\rm H}$  1.76), H-8 $\alpha$  ( $\delta_{\rm H}$  1.58), and H<sub>3</sub>-27, establish the *trans*-fused decalin ring system and the axial orientation of the ethylene bridge. Additional cross-peaks from H-18  $\delta_{\rm H}$  3.66 (dd, J = 9.6, 4.4 Hz) to H-16 $\alpha$  ( $\delta_{\rm H}$  2.04) and H-20a ( $\delta_{\rm H}$  1.31) were supportive of an  $\alpha$ -orientation of the H<sub>3</sub>-29 ( $\delta_{\rm H}$  0.96) methyl group was confirmed by H<sub>3</sub>-29/H-17 $\beta$ 



Fig. 5. Selected ROESY correlations for calyxaprenol C (3).

 $(\delta_{\rm H} 1.67)$  and H-20b  $(\delta_{\rm H} 1.61)$  ROESY correlations. (Fig. 5). From these data the relative configuration was assigned to be 6*S*\*, 7*S*\*, 11*R*\*, 18*S*\*, and 19*S*\*, which completed the structural elucidation of calyxaprenol C (3).

The molecular formula of calyxaprenol D (4), deduced to be C<sub>36</sub>H<sub>54</sub>O<sub>3</sub> from HRESIMS analysis, represented a loss of H<sub>2</sub>O from the molecular formula of **3**. The <sup>1</sup>H and <sup>13</sup>C NMR data of 4 were similar to those of 3 except for the addition of three proton signals at  $\delta_{\rm H}$  5.33 (1H, br s),  $\delta_{\rm H}$  5.26 (1H, br s), and  $\delta_{\rm H}$  5.14 (1H, t, J = 6.7 Hz) corresponding to three different trisubstituted olefins. The position of H-4 ( $\delta_{\rm H}$  5.33) in the C ring was determined by HMBC correlations from the olefinic methyl signal at H<sub>3</sub>-25 ( $\delta_{\rm H}$  1.44) to carbon signals at C-2 ( $\delta_{\rm C}$  56.3), C-3, and C-4 and the methyl signal at H<sub>3</sub>-26 ( $\delta_{\rm H}$  0.95) to C-2, C-6, C-7, and C-8. These correlations indicated that C-2 was flanked by a trisubstituted olefin on one side and a ring junction bearing a methyl group on the other side. In a similar manner, comparison with the NMR data of 3 revealed a double bond migration in ring A had occurred from C-14/C-15 to C-15/C-16 in 4, and this was supported by HMBC correlations from the methyl signal at H<sub>3</sub>-28 ( $\delta_{\rm H}$  1.71) to an sp<sup>3</sup> methine carbon C-14 ( $\delta_{\rm C}$ 49.9), a fully substituted olefinic carbon C-15 ( $\delta_{\rm C}$  137.8), and a protonated olefin carbon C-16 ( $\delta_{\rm C}$  120.1). The 4methyl-3-pentenyl substructure was elaborated based on COSY correlations between the allylic H<sub>2</sub>-21 protons and the vinyl methine H-22 ( $\delta_{\rm H}$  5.14, t, J = 6.7 Hz), as well as H<sub>2</sub>-20. The partial structure was completed through HMBC correlations from the geminal vinylic methyls H<sub>3</sub>-24 ( $\delta_{\rm H}$  1.69) and H<sub>3</sub>-30 ( $\delta_{\rm H}$  1.69) to C-22, C-23, C-24, and C-30. The ROESY data of 4 showed very similar protonproton cross-peaks to those of 3. ROESY correlations from H<sub>3</sub>-26 to H-5 $\beta$  ( $\delta_{\rm H}$  1.93), H-1 ( $\delta_{\rm H}$  2.48), and H-12a  $(\delta_{\rm H} 1.78)$ , together with correlations from H-6  $(\delta_{\rm H} 1.32)$  to H-2 ( $\delta_{\rm H}$  2.36) and H<sub>3</sub>-27 ( $\delta_{\rm H}$  0.86), established the *trans*fused B/C ring system and the axial position of the ethylene bridge substituted at C-11. Additional crosspeaks between H-18 ( $\delta_{\rm H}$  3.70)/H-14 ( $\delta_{\rm H}$  1.79) and H-20a ( $\delta_{\rm H}$  1.27), and H<sub>3</sub>-29 ( $\delta_{\rm H}$  0.85)/H-17β ( $\delta_{\rm H}$  1.92) and H-20b ( $\delta_{\rm H}$  1.69) were supportive of α-orientations for H-14 and H-18 and a β-orientation for H<sub>3</sub>-29, respectively. The relative configuration was thus assigned to be 2*S*\*, 6*S*\*, 7*S*\*, 11*R*\*, 14*R*\*, 18*S*\*, and 19*S*\* and the structure of calyxaprenol D (4) was established.

The known compound haliclotriol A (5) was also purified from the Calyx sp. extract and it was identified by comparison of its NMR data with published values.<sup>12</sup> Calyxaprenols A (1) and B (2) are related to 5 by addition of H<sub>2</sub>O and MeOH, respectively, to the trisubstituted sidechain olefin in 5. They are more distantly related to halicolic acid A, which is also a glycolic acid-containing merohexaprenoid that incorporates an oxepane ring, but it has a different carbon skeleton than the calyxaprenols.<sup>13</sup> Identification of the calyxaprenol merohexanprenoids from a sponge in the genus Calyx expands the known taxonomic distribution of this important class of marine metabolites. It also increases the chemical diversity of Calyx sponges and suggests that further chemical investigation of these sponges may prove fruitful. Merohexaprenoids such as the calyxaprenols represent hybrid biosynthetic products that incorporate components of both terpenoid and polyketide biosynthesis. While there have been numerous studies of the biosynthesis of hybrid nonribosomal peptide and polyketide metabolites, the biogenesis of hybrid terpenoid and polyketide compounds remains relatively under explored. Considering the range of important biological activities associated with merohexaprenoids, a better understanding of the biosynthetic processes and pathways associated with their production could provide future opportunities for enhanced natural product discovery efforts.

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