# **Research Article**

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# Effect of various bleaching treatments on shear bond strength of different universal adhesives and application modes

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# ABSTRACT

**Objectives:** The aim of this *in vitro* study was to evaluate the bond strength of 2 universal adhesives used in different application modes to bleached enamel.

**Materials and Methods:** Extracted 160 sound human incisors were used for the study. Teeth were divided into 4 treatment groups: No treatment, 35% hydrogen peroxide, 16% carbamid peroxide, 7.5% carbamid peroxide. After bleaching treatments, groups were divided into subgroups according to the adhesive systems used and application modes (*n* = 10): 1) Single Bond Universal, etch and rinse mode; 2) Single Bond Universal, self-etch mode; 3) Gluma Universal, etch and rinse mode; 4) Gluma Universal, self-etch mode. After adhesive procedures nanohybrid composite resin cylinders were bonded to the enamel surfaces. All specimens were subjected to shear bond strength (SBS) test after thermocycling. Data were

analyzed using a 3-way analysis of variance (ANOVA) and Tukey *post hoc* test. **Results:** No significant difference were found among bleaching groups (35% hydrogen peroxide, 16% carbamid peroxide, 7.5% carbamid peroxide, and no treatment groups) in the mean SBS values. There was also no difference in SBS values between Single Bond Universal and Gluma Universal at same application modes, whereas self-etch mode showed significantly lower SBS values than etch and rinse mode (p < 0.05).

**Conclusions:** The bonding performance of the universal adhesives was enhanced with the etch and rinse mode application to bleached enamel and non-bleached enamel.

Keywords: Adhesives; Shear strength; Tooth bleaching

# INTRODUCTION

Interest in aesthetic dentistry in achieving a beautiful smile has prompted the invention of tooth bleaching treatments [1,2]. In today's dentistry, 3 approaches are used for vital tooth bleaching: in-office, at-home, and over-the-counter (OTC) bleaching [3]. During in-office treatments, clinicians lighten teeth rapidly in a single visit, and the results appear to be as good as those with home bleaching treatments [4-6]. Home bleaching treatments are applied via custom-fabricated trays loaded with low levels of bleaching gels for at least 2 weeks (for 2–8 hr/day). In-office treatments are for professional use in clinical applications; they contain high concentrations of hydrogen peroxide (HP; 35%–38%) or carbamide peroxide (CP; 30%–35%), whereas at-home bleaching gels contain low concentrations (3%–16%) of carbamide or HP [7,8].

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### **Conflict of Interest**

No potential conflict of interest relevant to this article was reported.

### **Author Contributions**

Conceptualization: Oz FD, Kutuk ZB; Data curation: Oz FD; Formal analysis: Oz FD; Investigation: Oz FD; Methodology: Oz FD, Kutuk ZB; Project administration: Oz FD; Resources: Oz FD; Software: Oz FD; Supervision: Kutuk ZB; Validation: Kutuk ZB; Visualization: Oz FD, Kutuk ZB; Writing original draft: Oz FD; Writing - review & editing: Oz FD, Kutuk ZB.

### Effect of bleaching treatments on bond strength



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Fatma Dilsad Oz D https://orcid.org/0000-0002-7450-723X Zeynep Bilge Kutuk D https://orcid.org/0000-0003-0562-056X Interest in whitening the teeth for a better aesthetic appearance had led to the use of these bleaching materials. Patients' demands for bleaching procedures and the introduction of new bleaching products have prompted much research concerning the side effects of these materials. Free radicals released during bleaching procedures affect the properties of restorations such as microhardness, surface roughness, color, and microleakage [9].

Tooth mineral loss after bleaching may reduce the strength of the bond between restorative materials and enamel [10,11] because of changes in the mechanical properties of the enamel and dentin [1,12]. The decrease in microhardness or residual oxygen interferes with resin tag formation and inhibits resin polymerization through a free radical mechanism, thus affecting the bond strength [13,14]. To solve these clinical problems related to bleaching procedures, delaying bonding and waiting up to 10 days have been suggested [8,15].

The selection of adhesive type might influence bonding to bleached enamel [12,16]. Recent investigations in adhesive technology have led to the production of multi-mode universal adhesives. Clinicians can use universal adhesives in etch and rinse, self-etch, or selective etching mode depending on the case. Some *in vitro* studies [17-19] have evaluated the bond strength of universal adhesives, but the effects of bleaching procedures on their bonding efficiency in different modes have not been examined. As bleaching procedures have become popular, it is crucial to evaluate the bonding ability of different universal adhesives to bleached enamel. The aim of this *in vitro* study was to assess the shear bond strength (SBS) of 2 different universal adhesives in etch and rinse and self-etch modes after in-office and at-home bleaching treatments. The null hypotheses tested were as follows: 1) bleaching treatments would not affect the bond strength values of the tested universal adhesives, and 2) there would be no difference in bond strength values between the etch and rinse and self-etch modes.

### MATERIALS AND METHODS

This *in vitro* study was approved by the Institutional Ethical Committee of Hacettepe University (ethical approval No. GO 16/711-13). The materials used in this study and the study design are presented in **Table 1** and **Figure 1**.

Extracted sound, human anterior teeth (n = 160) were gathered, and the calculus, plaque, and remaining tissue were removed with scaling instruments and pumice using a rubber cup. The teeth were stored in 0.1% thymol for 1 week at room temperature and transferred to distilled water at 4°C until specimen preparation. A water-cooled diamond disc (Isomet, Buehler, Lake Bluff, IL, USA) was used to separate crowns from roots. After examination under a stereomicroscope (Leica, Meyer Instruments, Houston, TX, USA) for surface structural damage or defects, the teeth were embedded in a block of acrylic resin (Meliodent, Heraeus/Kulzer, Hanau, Germany) with the buccal surface positioned up for surface treatment and composite bonding. Enamel surfaces were polished with 200, 400, and 600 grit silicon carbide papers. The specimens were then randomly divided into four groups (n = 40). The negative control group received no treatment, while the three experimental groups were treated with either 35% HP (Total Blanc Office H35, Nova DFL, Rio de Janeiro, Brazil), 16% CP (Total Blanc Home C16, Nova DFL), or 7.5% CP (Total Blanc Home C7.5, Nova DFL) following the manufacturer's instructions as follows:

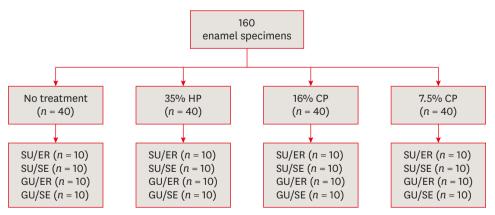
Total Blanc Office H35 (n = 40): A 1-mm layer of Total Blanc H35 was applied to the enamel surface for 20 minutes. All specimens in this group received 3 applications at 3-day intervals.



#### Table 1. Materials used in the study

Material /manufacturer	Batch No.	Composition	Application	
Total Blanc Office H35 /Nova DFL	12050770	35% hydrogen peroxide	A thin layer of gel was applied and left undisturbed for 20 minutes, then removed aith cotton rolls. Afterwards, the gel was applied for 20 more minutes. The enamel surfaces were rinsed with water and air dried.	
Total Blanc Home C16 /Nova DFL	12050720	16% carbamid peroxide	A thin layer of gel was applied and left undisturbed for 8 hours, then removed with cotton rolls. Afterwards, the enamel surfaces were rinsed with water and air dried. The same procedure was applied for 14 days/2 weeks without interruption.	
Total Blanc Home C7.5 /Nova DFL	12050665	7.5% carbamid peroxide	A thin layer of gel was applied and left undisturbed for 8 hours, then removed with cotton rolls. Afterwards, the enamel surfaces were rinsed with water and air dried. The same procedure was applied for 14 days/2 weeks witout interruption	
Single Bond Universal /3M ESPE	585360	MDP phosphate monomers, dimethacrylate resins, HEMA, methacrylate-modified polyalkenoic acid copolymer, fillers, ethanol, water, initiators, silane	Etch and rinse technique Phosphoric acid etching gel (37%) was applied to enamel and left in place for 30 seconds, then rinsed rinse and dried. Bond was applied to the entire cavity wall with the applicator brush and rubbed for 20 seconds. The entire cavity wall was dried sufficiently by blowing mild air for more than 5 seconds until bond does not move. Bond was light-cured with 1,200 mW/cm <sup>2</sup> LED for 10 seconds. Self-etch technique Bond was applied to the entire cavity wall with the applicator brush and rubbed for 20 seconds. The entire cavity wall was dried sufficiently by blowing mild air for more than 5 seconds until bond does not move. Bond was light-cure bond with 1,200 mW/cm <sup>2</sup> LED for 10 seconds.	
Gluma Bond Universal /Heraeus Kulzer GmbH	010022	MDP phosphate monomers, 4-META, dimethacrylate resins, acetone, fillers, initiators, silane	Etch and rinse technique Phosphoric acid etching gel (37%) was applied to enamel and left in place for 30 seconds, then rinsed rinse and dried. Bond was applied to the entire cavity wall with the applicator brush and rubbed for 20 seconds. The entire cavity wall was dried sufficiently by blowing mild air for more than 5 seconds until bond does not move. Bond was light-cured with 1,200 mW/cm <sup>2</sup> LED for 10 seconds. Self-etch technique Bond was applied to the entire cavity wall with the applicator brush and rubbed for 20 seconds. The entire cavity wall was dried sufficiently by blowing mild air for more than 5 seconds until bond does not move. Bond was light-cure bond with 1,200 mW/cm <sup>2</sup> LED for 10 seconds.	
Herculite XRV Ultra Universal Nano-hybrid composite /Kerr Corporation	5139411	3-trimethoxysilylpropyl methacrylate hexamethylene diacrylate 2,2'-ethylenedioxydiethyl dimethacrylate 1,6-hexanediyl bismethacrylate Bis-GMA	Two mm increments were placed and light cured for 20 seconds with 1,200 mW/ cm <sup>2</sup> LED	

MDP, 10-methacryloyloxydecyl dihydrogen phosphate; HEMA, 2-hydroxyethyl methacrylate; LED, light-emitting diode; 4-META, 4-methacryloyloxyethy trimellitate; Bis-GMA, bisphenol A-glycidyl methacrylate.



#### Figure 1. Study design.

SU, Single Bond Universal; GU, Gluma Universal; ER, etch and rinse; SE, self-etch; HP, hydrogen peroxide; CP, carbamide peroxide.

Total Blanc Home C16 (n = 40): A 1-mm layer of Total Blanc C16 was applied to the enamel surface for 6 hours daily for 14 days.



Total Blanc Home C7.5 (n = 40): A 1-mm layer of Total Blanc C7.5 was applied to the enamel surface for 6 hours daily for 14 days.

The enamel specimens were dried using cotton pellets before the bleaching procedure. A soft toothbrush was used to remove the bleaching gel under running tap water, and the samples were stored in distilled water at 37°C. The distilled water was changed daily for 14 days. After 14 days, the specimens were divided into four groups according to the adhesive material and application mode (n = 10) as presented in **Figure 1**. The adhesives Single Bond Universal (SU) and Gluma Universal (GU) were applied according to the manufacturers' instructions (Table 1). The adhesives were cured using a light-emitting diode (LED) light-curing unit (Radii Plus, SDI, Victoria, Australia) in standard mode. The output of the light-curing unit was regularly checked to ensure that it was 1.200 mW/cm<sup>2</sup>. After adhesive application, a nanohybrid universal composite resin (Herculite XRV Ultra Universal, Kerr Corp., Orange, CA, USA) was inserted into a Teflon tube (3 mm wide and 2 mm high) seated perpendicularly against the prepared enamel surfaces and light-cured for 40 s using the LED light-curing unit. After curing, the Teflon tube was removed carefully. The specimens were stored in distilled water at 37°C for 24 hours and thermocycled for 5,000 cycles between 5°C and 55°C with a dwell time of 25 seconds each. The cylinders were then loaded with a chisel-shaped loading metal rod parallel with and close to the bonding interface at 1 mm/min in the shear mode until fracture occurred (LR50K, Lloyd Instruments, Fareham, UK). SBS (MPa) values were calculated by dividing the failure load (N) by the cross-sectional area (mm<sup>2</sup>) of the cylindrical composite. The SBS data exhibited a normal distribution according to the Kolmogorov-Smirnov test; therefore, a three-way analysis of variance and Tukey *post hoc* test at a significance level of 0.05 were used to analyze differences. The failure modes were observed at 40× magnification using a stereomicroscope (SZX7, Olympus, Hamburg, Germany) and classified as either adhesive (between the enamel and bond or composite and bond), cohesive in composite, cohesive within the enamel, or mixed (a combination of adhesive and cohesive failure).

### RESULTS

The mean SBS values and standard deviations of each group are presented in **Table 2**. The specimens in the no treatment group in which the composite was bonded with SU using the etch and rinse mode exhibited the highest mean SBS ( $34.8 \pm 6.98$  MPa). No statistically significant differences in SBS values were observed among the bleaching treatment groups (no treatment, 35% HP, 16% CP, and 7.5% CP; p = 0.346). There were no significant differences in the SBS values between adhesives in the same application modes. However, the self-etch groups showed significantly lower SBS values compared to the etch and rinse groups (p = 0.001). Specimens bleached with 16% CP with SU applied in self-etch mode ( $18.5 \pm 7.35$  MPa) and specimens bleached with 35% HP with SU applied in self-etch mode ( $18.9 \pm 6.75$  MPa) showed the lowest mean SBS values (p < 0.05).

Table 9 Shear bond strongth (SPS)	values of universal adhesives to enamel	according to etch and rinse or self-etch modes
Table 2. Shear bond Scienger (303)	values of universal autresives to enamer	according to etch and thise of self-etch modes

Group	No treatment	16% carbamid peroxide	7.5% carbamid peroxide	35% hydrogen peroxide
SU/ER	$34.8 \pm 7.0^{a}$	$32.4 \pm 6.4^{a}$	$33.2\pm6.6^a$	$30.9 \pm 7.4^{a}$
SU/SE	$20.3 \pm 6.0^{b}$	$18.5 \pm 7.4^{b}$	$20.6 \pm 4.7^{b}$	$18.9 \pm 6.8^{b}$
GU/ER	$33.5 \pm 8.1^{a}$	31.0 ± 10 <sup>a</sup>	$32.6 \pm 8.5^{a}$	$29.8 \pm 6.7^{a}$
GU/SE	$22.3 \pm 7.1^{b}$	$20.4 \pm 7.5^{b}$	$21.0 \pm 7.1^{b}$	$20.7 \pm 7.0^{b}$

The SBS values are shown as means  $\pm$  standard deviations.

<sup>a,b</sup>Different superscript letters show significant differences within the column (p < 0.05).

SU, Single Bond Universal; GU, Gluma Universal; ER, etch and rinse; SE, self-etch.



#### Effect of bleaching treatments on bond strength

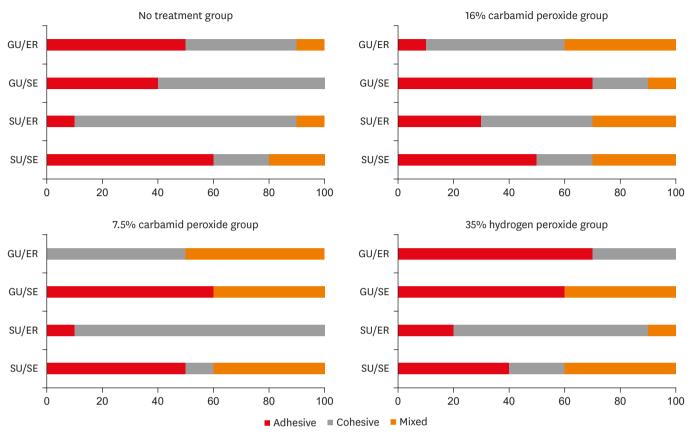


Figure 2. Failure mode distrubutions (%).

SU, Single Bond Universal; GU, Gluma Universal; ER, etch and rinse; SE, self-etch.

A failure mode examination showed that 39.3% of the specimens had adhesive failures, whereas 37.5% had cohesive failures and 23.1% had mixed failures (**Figure 2**).

### DISCUSSION

The present study compared the SBS of a nanohybrid resin composite to human enamel that was bleached with one in-office (35% HP) and two home bleaching materials (16% CP and 7.5% CP) and bonded with 2 universal adhesives in etch and rinse and self-etch modes. Adhesive procedures were conducted at 14 days following the bleaching treatments.

Many types of bleaching systems are currently available with different concentrations of HP or CP as active agents. The vital bleaching agents have direct contact with the enamel surface and might cause chemical and morphological changes in the enamel structure [20-22]. Declines in bond strength may be caused by bleaching treatment or the immediate placement of adhesive restorative materials after bleaching treatment. Furthermore, a number of studies revealed that residual oxygen from the bleaching agent present on the enamel surface might be the cause of failure in adhesive procedures after bleaching [14,20,21]. The adverse effects of bleaching materials on adhesive systems have been investigated widely, and these studies showed that waiting time is crucial for better bond strength values [10,14,15]. *In vitro* studies emphasized that the immediate SBS values of bleached enamel were low, whereas the values obtained at 14–21 days after bleaching were nearly the same as those of untreated



controls [10,14,15]. Similarly, in the present study, after 14 days no significant differences were detected between the bleached and no-treatment groups. In addition, an *in vitro* study [23] compared waiting times after bleaching treatments prior to adhesive application and reported similar SBS values after 7–14 days of waiting.

There was no significant difference in the SBS values of bleached enamel with 16% CP, 7.5% CP, and 35% HP. The first hypothesis is thus supported. Since the different concentrations of home bleaching agents did not affect the bond strength values, we concluded that an increased concentration did not prolong the time needed prior to bonding. The specimens that were not treated with any bleaching gel showed similar results to the bleached specimens. Conversely, an *in vitro* study reported that office bleaching treatment with 40% HP reduced SBS values but home bleaching treatment with 10% CP had no effect [24]. However, the adhesives were applied after 1 day in that study. Similar to these findings, Akın *et al.* [25] showed that 10% CP bleaching did not significantly reduce SBS values but 38% HP bleaching did. On the other hand, Gungor *et al.* [26] suggested that both home and office treatments adversely affected SBS values, but the home bleaching group had lower SBS values than the office bleaching group, unlike in other studies.

Universal adhesives represent the most recent generation of adhesives and can be used in different modes [27,28]. The adhesive systems compared in the present study had different solvents; the ethanol-based adhesive (SU) and acetone-based adhesive showed similar SBS values in the etch and rinse mode. Montalvan *et al.* [16] also reported in their study that after bleaching with 35% HP, the SBS values of an ethanol-based adhesive and an acetone-based adhesive were similar. Additionally, groups that did not receive bleaching treatments and were bonded in etch and rinse mode demonstrated no significant differences. Previous studies [29,30] showed findings similar to ours (i.e., that adhesives with different solvents [ethanol or acetone] showed no difference in SBS tests without bleaching treatments).

Laboratory studies [18,31,32] showed that universal adhesives showed significantly higher SBS values on enamel in the etch and rinse mode than in self-etch mode, as in the present study. Similar to these findings, Vermelho et al. [33] suggested that SU adhesive produced superior SBS values in etch and rinse mode compared to self-etch mode. Additionally, another study comparing bonding efficiency in self-etch and etch and rinse modes of a different universal adhesive (G-Bond Plus, GC Corporation, Tokyo, Japan) clearly indicated that phosphoric acid etching significantly increased the strength of bonding to enamel [34]. Also, Suzuki et al. [19] compared the enamel bond durability of several universal adhesives in different etching modes and found results similar to ours. McLean et al. [27] also reported that the etching procedure increases SBS values, whereas a self-etch adhesive (Clearfil SE, Kuraray Dental, New York, NY, USA) showed significantly higher SBS values than SU. The results of the current research show the superiority of etch and rinse mode compared to self-etch mode for the 2 tested universal adhesive systems; therefore, the second hypothesis is rejected. Therefore, the treatment of dental tissues prior to adhesive application is an important step for the clinical success of restorations. Two recent clinical studies [35,36] on the performance of universal adhesives mentioned that no significant differences were seen between the etch and rinse and self-etch modes in any criteria evaluated, but long-term observations are needed.

The universal adhesives used in the present study contain 10-methacryloyloxydecyl dihydrogen phosphate (MDP monomer), which provides chemical bonding to



hydroxyapatite. At the etched enamel substrate, chemical bonding of MDP monomers could significantly increase long-term enamel bonding [37]. The pH value of SU (pH 2.7) is considered ultramild, whereas the pH value of GU (pH 1.8) is mild. Therefore, the additional etching step clearly improved micromechanical bonds between the composite resin and the highly mineralized enamel substrate. Nonetheless, neither the acidity of the adhesive agent nor the thickness of the hybrid layer nor the length of the resin tags is solely responsible for bonding effectiveness and stability for all adhesives. Although the universal adhesives in this study had different pH values, the SBS values were not affected by acidity.

The failure modes were mostly cohesive for SU in etch and rinse mode, whereas they were adhesive in self-etch mode. Mixed mode failures were observed mostly with GU in the etch and rinse mode after 7.5% CP application. These adhesive failures demonstrate the weak link between the composite resin and enamel surface due to the absence of phosphoric acid etching treatment prior to adhesive applications. Cohesive and mixed failures were found at a high frequency on specimens treated using the etch and rinse mode.

# CONCLUSIONS

Within the limitations of this *in vitro* study, we conclude that the tested universal adhesives showed significantly higher bond strength in etch and rinse mode than in self-etch mode regardless of the bleaching treatment.

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