# Synthesis of Novel Mercaptophenyl Carbocyclic C-Nucleoside Analogue Using Sequential [3,3]-Sigmatropic Rearrangement and Ring-closing Metathesis 

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Novel mercaptophenyl carbocyclic $C$-nucleoside analogue was synthesized wia a cyclopentenol intermediate 10, which was prepared using a sequential [3,3]-sigmatropic rearrangement and ring-closing metathesis (RCM). Friedel-Crafts alkylation was then used to couple the thiophenol.
Key Words : Carbocyclic C-nucleoside. [3,3]-Sigmatropic rearrangement. Ring-closing metathesis, FriedelCrafts alkylation

## Introduction

Recently, aromatic C-nucleosides ${ }^{1}$ for both information storage and retrieval of DNA were synthesized. which generally bind DNA quite nonselectively at almost any sequence. This has paved the way for the generation of new functional materials. a new genetic code. or novel antisense nucleotides. ${ }^{\text {- The replacement of natural DNA bases with aromatic }}$ analogues can provide functions towards the expansion of the genetic alphabet and be used to align metal ions along a helical axis inside the duplex. ${ }^{3}$

Synthesis of natural carbocyclic and $C$-nucleoside analogues have been inspired by their interesting biological activities as well as their chemical and enzymatic stability. ${ }^{4}$ $C$-nucleosides are a unique class of nucleosides in which the heterocyclic is comnected to sugar moiety by a $C-C$ bond instead of the $C-N$ bond of natural nucleosides. ${ }^{5}$ As a result. they are resistant to the chemical and enzy'matic hydrolytic cleavage of the gly'cosidic bond.

Therefore as a part of our drug discovering program, a novel mercaptophenyl carbocyclic C-nucleoside analogue. which combine the properties of the enzyme resistant carbocyclic carbohydrate moiety ${ }^{6}$ and the redox-active nucleosidic base. were synthesized using thiophenol. This paper reports the synthetic route employing a versatile reaction sequence ([3.3]-sigmatropic rearrangement, ring-closing metathesis. and Friedel-Crafts alkylation) from a simple acyclic starting material.

## Results and Discussion

As shown in Scheme 1. the $\alpha \cdot \beta$-unsaturated ethylester 3 was synthesized from acetol 1 and reduced by DIBALH at $-78^{\circ} \mathrm{C}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give the allylic alcohol 4 . The ally lic alcohol was subjected to a [3,3]-sigmatropic rearrangement using triethylorthoacetate to produce the $\gamma \delta$-unsaturated ester $5{ }^{7}$ The ester derivative 5 was reduced to alcohol 6 by DIBALH. which was then oxidized to the aldehyde 7 using PCC. The slow addition of viny lmagnesium bromide to a solution of the aldehyde 7 in THF at $-78^{\circ} \mathrm{C}$ could fumish the divinyl intermediate 8 .

Using the divinyl 8. the fomation of five-membered carbocycle was examined. $\mathrm{RCM}^{\S}$ now stands as one of the most powerful tools for preparing medium to large ring systems through $C-C$ bond formation. This powerful procedure was successfully adopted for the elaboration of our desired five member carbocyclic moiety. Therefore. the divinyl 8 was subjected to ring-closing metathesis conditions using the $2^{\text {nd }}$ generation Grubbs catalyst [ $(\mathrm{Im}) \mathrm{Cl}_{2} \mathrm{PCy}_{3} \mathrm{Ru}-$ CHPh ] to afford the cyclopentenol $9(45 \%)$ and $10(46 \%)$. respectively. ${ }^{\circ}$ The relative stereochemistry of the cyclopentenol products ( 9 and 10 ) was determined by employing the NOE experiment between the proximal hydrogen atoms. Upon the irradiation of $C_{1}-\mathrm{H}$. a relatively strong NOE was observed at the methyl protons of compound $9\left[C_{4}-\mathrm{H}\right.$ $(0.7 \%)]$, but not at the methyl protons of compound $\mathbf{1 0}\left[C_{4}-\right.$ $\mathrm{H}(0.2 \%)$ ] (Figure 1).
To our surprise. the Friedel-Craft approach wia electrophilic aromatic disulfide was applied successfully to the allylic position of the carbocyclic system. which is usually


Scheme 1. Synthesis of cyclopentenol intermediates. Reagents: i) TBDMSCI, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, imidazole; ii) Triethylphosphonoacetate, $\mathrm{NaH}, \mathrm{THF}$; iii) DIBALH, $\mathrm{CH}_{2} \mathrm{Cl}_{2},-20^{\circ} \mathrm{C}$ : iv) Triethylorthoacetate, propionc acid, overnight, $130-135^{\circ} \mathrm{C}$; v) DIBALH, toluene, -78 ${ }^{\circ} \mathrm{C}$; vij $\mathrm{PCC}, 4 \mathrm{MS}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; vii) $\mathrm{CH}_{2}=\mathrm{CHMgBr}, \mathrm{THF},-78^{\circ} \mathrm{C}$; viii) Grubbs' catalyst $\mathrm{II}_{9} \mathrm{CH}_{2} \mathrm{Cl}_{2}$, reflux, ovemight.


NOE (0.7\%) 9


NOE (0.2\%) $\quad 10$

Figure 1. NOE comparisons of compound 9 and 10 .
employed in the furanose system. ${ }^{10}$ The activation of the hydroxyl group of compound $\mathbf{1 0}$ to the acetoxy group $\mathbf{1 1}$ was made using acetic anhydride. which was coupled to phenyl disulfide in the condition of $\mathrm{SnCl}_{4}$ as a Lewis acid to give the desired disulfide derivative 12. The disulfide linkage of compound 12 was reduced successfully by $\mathrm{LiAlH}_{4}$ to provide the thiophenol analogue 13 . and which was readily desilyated by a treatment with tetrabutylammonium fluoride (TBAF) to provide compound target aromatic (-nucleoside analogue 14. Although the structure of compound synthesized analogue is quite different from the classical nucleosides, its antiviral activity was evaluated against a wide variety of DNA and RNA viruses [herpes simplex virus type 1 (HSV-1 strain KOS) and type-2, starin G]. vaccinia virus. vesicular stomatitis virus, thymidine kinase-deficient (TK HSV-1 stain KOS). varicella-zoster virus (TK VZV stain Oka and TK strain 07/L). cytomegalovirus (stain AD-169 and Davis). Coxachie B4 virus]. However. this structure did not show any significant antiviral activity against several viruses.
The synthetic information obtained in the present study will be useful for the synthesis of novel aromatic C-nucleosides and their derivatives to clarify the mechanism of the disulfide base-pairing for base-to-base formation during the incorporation of synthetic aromatic (-nucleosides into DNA.
In sunmary. a synthetic procedure of novel non-classical nucleoside analogue was developed using sequential Johnson's Claisen rearrangement, ring-closing metathesis. and Friedel-Crafts alkylation. In our laboratory. this reiterative three-step sequences (i.e. [3.3]-sigmatropic rearrangement. ring-closing metathesis, and Friedel-Crafts reaction) have been widely used for the synthesis of a variety of nonclassical aromatic nucleoside analogues and their phosphramidite derivatives.

## Experiments

The melting points were determined on a Mel-temp II laboratory device and were uncorrected. The NMR spectra
were recorded on a Bruker 300 Fourier transform spectrometer; the chemical shifts are reported in parts per million ( $\delta$ ) and the signals are quoted as s (singlet). d (doublet). t (triplet), q (quartet), m (multiplet) and dd (doublet of doublets). The UV spectra were obtained on a Beckman DU-7 spectrophotometer. The elemental analyses were performed using an Elemental Analyzer System (EAlll2). The TLC was performed on Uniplates (silica gel) purchased from Analtech Co. Unless stated otherwise, all the reactions were carried out in a $\mathrm{N}_{2}$ atmosphere. Dry dichloromethane, benzene and pyridine were obtained by distillation from $\mathrm{CaH}_{2}$. Dry THF was obtained by distillation from Na and benzophenone inmediately before use.

2-(t-Butyldimethylsilyloxy)-acetone (2): TBDMSCl (44 g. 0.297 mol ) was added slowly to a solution of acetol 1 ( 20 g. 0.27 mol ) and imidazole ( $27 \mathrm{~g}, 0.405 \mathrm{~mol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(300$ mL ) at $0^{\circ} \mathrm{C}$. and stirred for 5 h at the same temperature. The solvent was evaporated under reduced pressure. The residue was extracted twice with diethyl ether and water. The combined organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered. and concentrated under reduced pressure. The residue was purified by silica gel column chromatography ( $\mathrm{EtOAc} / \mathrm{hexane}, 1: 10$ ) to give compound $2(39.6 \mathrm{~g} .78 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3} .300 \mathrm{MHz}\right) \delta 4.05(\mathrm{~s}, 2 \mathrm{H})$, $2.07(\mathrm{~s} .3 \mathrm{H}), 0.84(\mathrm{~s} .9 \mathrm{H}), 0.07(\mathrm{~s} .6 \mathrm{H})$.
(E) and (Z)-4-(t-Butyldimethylsilyloxy)-3-methyl-but-2-enoic acid ethyl ester (3): Triethyl phosphonoacetate ( 1.405 mL .9 .25 mmol ) was added drop wise to a suspension of sodium hydride ( $60 \%$ in mineral oil. $0.37 \mathrm{~g}, 9.25 \mathrm{mmol}$ ) in distilled THF at $0^{\circ} \mathrm{C}$, and stirred constant at room temperature for 1 h . The ketone $2(1.74 \mathrm{~g}, 9.25 \mathrm{mmol})$ was added to this mixture and stirred for 1 h . The solution was neutralized with AcOH , and extracted with EtOAc. The organic layer was washed with brine, dried over auhydrous $\mathrm{MgSO}_{4}$. filtered and evaporated. The residue was purified by silica gel colunn chromatography (EtOAc/hexane, 1:15) to give compound $3(2.29 \mathrm{~g}, 96 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3} .300 \mathrm{MHz}\right) \delta 5.99(\mathrm{~s} .1 \mathrm{H}), 4.17(\mathrm{q} . J=6.9 \mathrm{~Hz} .2 \mathrm{H})$, 4.13 (s. 2 H ). 2.05 .1 .96 (s. s. 3 H ). 1.22 (t. $J=6.7 \mathrm{~Hz}, 3 \mathrm{H}$ ). 0.93 .0 .90 (s. s. 9 H ). 0.09 .0 .08 (s. s. 6 H ).
( $E, Z$ )-+-( $t$-Butyldimethylsilyloxymethyl)-3-methyl-but-2-en-1-ol (4): DIBALH ( 53.9 mL .1 .0 M solution in hexane) was added slowly to a solution of compound $3(6.64 \mathrm{~g} .25 .7$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(150 \mathrm{~mL})$ at $-20^{\circ} \mathrm{C}$, and stirred for I h at the same temperature. Methanol ( 53 mL ) was then added to the resulting mixture. The mixture was stirred at room temperature for 3 h , and the resulting solid was filtered through a Celite pad. The filtrate was concentrated under


Scheme 2. Synthesis of mercaptophenyl carbocyclic $C$-nucleoside.
vacuum. and the residue was purified by silica gel colunn chromatography (EtOAc/hexane, 1:5) to give the allylic alcohol $4(4.95 \mathrm{~g} .89 \%)$ as a colorless oil: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right.$. $300 \mathrm{MHz}) \delta 5.68(\mathrm{br} \mathrm{s}, \mathrm{IH}), 4.2 \mathrm{I}(\mathrm{d} . J=6.6 \mathrm{~Hz} .2 \mathrm{H}) .4 .03(\mathrm{~s}$. $2 \mathrm{H}), 1.77 .1 .64(\mathrm{~s}, \mathrm{~s} .3 \mathrm{H}), 0.91(\mathrm{~s} .9 \mathrm{H}), 0.07(\mathrm{~s} .6 \mathrm{H})$.
( $\pm$ )-3-( $t$-Butyldimethylsilyloxymethyl)-3-methyl-pent-+enoic acid ethyl ester (5): Propionic acid ( 0.5 mL ) was added to a solution of allylic alcohol $4(6 \mathrm{~g}, 27.73 \mathrm{mmol})$ in tiethyl orthoacetate ( 110 mL ), and the mixture was heated at $130-135^{\circ} \mathrm{C}$ overnight with constant stirring under conditions suitable for the removal of ethanol by distillation. The excess triethyl orthoaceate was distilled off and the residue was purified by silica gel column chromatography (EtOAc/ hexane. 1:20) to give compound $5(6.59 \mathrm{~g} .83 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3} .300 \mathrm{MHz}\right) \delta 5.91(\mathrm{~d}, J=10.8$ $\mathrm{Hz}, 1 \mathrm{H}), 5.89(\mathrm{~d} . J=11.4 \mathrm{~Hz}, \mathrm{IH}) .5 .05(\mathrm{~d}, J=1.2 \mathrm{~Hz} . \mathrm{IH})$. $5.02(\mathrm{~d}, J=7.5 \mathrm{~Hz} .1 \mathrm{H}) .401(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.46(\mathrm{~d}, J$ $=9.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~d}, J=9.3 \mathrm{~Hz} .1 \mathrm{H}), 2.40(\mathrm{~d}, J=3.3 \mathrm{~Hz}$. $2 \mathrm{H}), 1.23(\mathrm{t} . J=7.5 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{~s}, 9 \mathrm{H}) .0 .02(\mathrm{~s} .6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.\mathrm{CDCl}_{3} .75 \mathrm{MHz}\right) \delta 171.94,143.11,112.99,69.91$. 59.83. 41.33, 25.81, 22.60, 20.70, 18.20, 14.26, -5.58.
( $\pm$ )-3-( $t$-Butyldimethylsilyloxymethyl)-3-methyl-pent-t-en-1-01 (6) DIBALH ( $33.98 \mathrm{~mL}, 1.0 \mathrm{M}$ solution in hexane) was added slowly to a solution of compound $5(7.5 \mathrm{~g} .16 .18$ mmol) in $\mathrm{CH}_{2} \mathrm{Cl}_{-}(200 \mathrm{~mL})$ at $-20^{\circ} \mathrm{C}$, and stirred for 1 h . Methanol ( 34 mL ) was then added to the resulting mixture. The mixture was stirred at room temperature for 2 h , and the resulting solid was filtered through a Celite pad. The filtrate was concentrated under vacuum and the residue was purified by silica gel colunn chromatography ( EtOAc /hexane. 1:6) to give the alcohol $6(3.56 \mathrm{~g}, 90 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ $\mathrm{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 5.89(\mathrm{dd}, J=17.7,11.4 \mathrm{~Hz}, \mathrm{IH})$. $5.06(\mathrm{dd}, J=4.5 .1 .2 \mathrm{~Hz} .1 \mathrm{H}), 5.02(\mathrm{dd} . J=10.5 .0 .9 \mathrm{~Hz}$. $1 \mathrm{H}), 3.67$ (dd. $J=11.7 .6 .3 \mathrm{~Hz}, 1 \mathrm{H}), 3.46(\mathrm{~d}, J=9.6 \mathrm{~Hz}$. $1 \mathrm{H}), 3.38$ (d. $J=9.6 \mathrm{~Hz}, 1 \mathrm{H}) .1 .69(\mathrm{t} . J=5.7 \mathrm{~Hz}, 2 \mathrm{H}) .1 .00$ $(\mathrm{s} .3 \mathrm{H}) .0 .90(\mathrm{~s} .9 \mathrm{H}) .0 .06(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75\right.$ $\mathrm{MHz}) \delta 144.45 .112 .80 .70 .81 .59 .34,41.18,25.84,21.04$, 18.29. -5.52: Anal calc for $\mathrm{C}_{13} \mathrm{H}_{28} \mathrm{O}_{2}$ Si: C. 63.87 ; H. 11.55. Found: C. 63.74 : H. 11.61.
( $\pm$ )-3-(tert-Butyl-dimethylsilyloxymethyl)-3-methyl-pent-4-enal (7): $4 \AA$ molecular sieves ( 4.2 g ) and PCC ( 3.87 g. 18.13 mmol ) were added slowly to a solution of compound $6(1.76 \mathrm{~g} .7 .2 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. and stirred overnight at rt . An excess of diethyl ether ( 200 mL ) was then added to the mixture. The misture was stirred vigorously for 2 h at the same temperature. and the resulting solid was filtered through a short silica gel colunn. The filtrate was concentrated under vacuum and purified by silica gel column chromatography (EtOAc/hexane. 1:20) to give compound $7(1.39 \mathrm{~g} .80 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 9.75(\mathrm{t} . J=3.3 \mathrm{~Hz}, \mathrm{IH}) .5 .96(\mathrm{~d}, J=$ $11.1 \mathrm{~Hz}, 1 \mathrm{H}) .5 .90(\mathrm{~d} . J=10.8 \mathrm{~Hz}, 1 \mathrm{H}) .5 .14(\mathrm{~d} . J=8.4 \mathrm{~Hz}$. $1 \mathrm{H}) .5 .11(\mathrm{~d} . J=16.2 \mathrm{~Hz}, 1 \mathrm{H}) .3 .49(\mathrm{~d} . J=9.6 \mathrm{~Hz}, 1 \mathrm{H}) .3 .40$ (d. $J=9.3 \mathrm{~Hz} .1 \mathrm{H}$ ). 2.41 (t. $J=3.0 \mathrm{~Hz} .2 \mathrm{H}$ ). 1.12 (s. 3 H ). 0.88 (s. 9H) .0 .03 (s. 6 H ): ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta$ $203.08,142.61,113.93,70.53 .50 .70 .41 .60 .25 .80 .21 .30$. -5.61 .
(rel)-(3R and 3S,5S)-5-(tert-Butyl-dimethylsilyloxy-methyl)-5-methyl-hepta-4-1,6-diene-3-ol (8): Vinylmagnesium bromide ( 9.8 mL . 1 M solution in THF) was added slowly to a solution of compound $7(2.83 \mathrm{~g} .8 .89 \mathrm{nmmol})$ in anlydrous THF at $-78^{\circ} \mathrm{C}$ and stirred for 2 h at the same temperature. The mixture was quenched with a saturated ammonium chloride solution ( 10 mL ) and elevated to room temperature. The mixture was extracted with EtOAc and water. and the organic layer was washed with brine. dried over anhydrous $\mathrm{MgSO}_{4}$, and then filtered. The filtrate was concentrated under reduced pressure, and purified by silica gel column chromatography (EtOAc/hexane, 1:10) to give a mixture of compound 8 ( $2.19 \mathrm{~g} .91 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $\left.\mathrm{CDCl}_{3 .} .300 \mathrm{MHz}\right) \delta 6.02-5.74(\mathrm{~m} .2 \mathrm{H}) .5 .27-4.99(\mathrm{~m}$, $4 \mathrm{H}) .4 .25($ br s, 1 H$) .3 .56-3.40(\mathrm{~m}, 2 \mathrm{H}) .1 .70-1.56(\mathrm{~m}, 2 \mathrm{H})$, 1.08. $1.02(\mathrm{~s}, \mathrm{~s}, 3 \mathrm{H}) .0 .91(\mathrm{~s}, 9 \mathrm{H}), 0.08(\mathrm{~s} .6 \mathrm{H})$.
(rel)-( $1 R, 4 S$ )-1-(tert-Butyldimethylsilanyloxymethyl)-4-methyl-cyclopenten-2-ol (9); and (rel)-(1S,4S)-4-(tert-butyldimethylsilanyloxymethyl)-4-methyl-cyclopenten-2-0l (10): A $2^{\text {nil }}$ generation Grubbs* catalyst ( $22 \mathrm{mg}, 0.025$ mmol ) was added to a solution of compound $8(935 \mathrm{mg}, 3.46$ mmol) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. The reaction misture was refluxed ovemight. and concentrated. The residue was purified by silica gel columu chromatography (EtOAc/hexane, 1:5) to give the cyclopentenol $9(377 \mathrm{mg}, 45 \%)$ and compound 10 ( $386 \mathrm{mg} .46 \%$ ); compound 9. Spectra for 9 : ${ }^{1} \mathrm{H}$ NMR (CDCl $\left.{ }_{3}, 300 \mathrm{MHz}\right) \delta 5.81$ (dd. $\left.J=5.4 .2 .1 \mathrm{~Hz} .1 \mathrm{H}\right)$, $5.45(\mathrm{~d} . J=5.7 \mathrm{~Hz}, \mathrm{IH}) .4 .50(\mathrm{t} . J=7.8 \mathrm{~Hz}, 1 \mathrm{H}) .3 .32(\mathrm{~s}$, $2 \mathrm{H}) .1 .83(\mathrm{dd}, J=14.1,6.9 \mathrm{~Hz}, 1 \mathrm{H}) .1 .67(\mathrm{~d} . J=14.1 .1 \mathrm{H})$, $0.95(\mathrm{~s}, 3 \mathrm{H}), 0.82(\mathrm{~s}, 9 \mathrm{H}), 0.01(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3} .75\right.$ $\mathrm{MHz}) \delta 140.34,133.89 .76 .14$. 69.35. 50.04. 45.20. 25.99, 23.32. $18.57,-5.52$; Anal calc for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Si}$ : C. $64.41: \mathrm{H}$, 10.81. Found: C, 64.32: H. 10.98. Spectra for 10 : ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3} .300 \mathrm{MHz}\right) \delta 5.73(\mathrm{~m} .2 \mathrm{H}) .4 .84(\mathrm{t}, J=6.0 \mathrm{~Hz} .1 \mathrm{H})$. $3.32(\mathrm{~s}, 2 \mathrm{H}), 2.29$ (dd. $J=13.5,7.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.37$ (dd. $J=$ $13.5,4.2 \mathrm{~Hz}, \mathrm{lH})$. $1.11(\mathrm{~s} .3 \mathrm{H}) .0 .87(\mathrm{~s}, 9 \mathrm{H}), 0.01(\mathrm{~s} .6 \mathrm{H})$ : ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3} .75 \mathrm{MHz}\right) \delta 141.50,133.02,77.61 .70 .64$. 50.94. 44.83, 25.83. 24.67. 18.23. -5.50: Anal calc for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Si}: \mathrm{C}, 64.41$ : H. 10.81 . Found: C, 64.56 : H, 10.76.
(rel)-( $1 S, 4 S$ )-1-Acetoxy-4-(tert-butyldimethylsilanyloxy-methyl)-4-methyl-cyclopenten-2-ene (11): Acetic anhydride ( 1.64 g .16 .08 mmol ) and dimethy laminopyridine ( 131 mg .1 .072 mmol ) were added to a solution of compound $\mathbf{1 0}$ $(2.6 \mathrm{g} 10.72 \mathrm{mmol}$.$) in anhydrous pyridine (15 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. The mixture was then stirred ovenight at room temperature. A sat. $\mathrm{NaHCO}_{3}(5 \mathrm{~mL})$ solution was added to the mixture and concentrated under reduced pressure. The residue was extracted with EtOAc/water, and the organic layer was washed with brine. dried over anhydrous $\mathrm{MgSO}_{4}$, and then filtered. The residue was purified by silica gel column chromatography ( $\mathrm{EtOA} /$ /hexane $1: 10$ ) to give compound 11 ( $2.59 \mathrm{mg} .85 \%$ ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}, 300$ $\mathrm{MHz}) \delta 5.71(\mathrm{~m}, \mathrm{IH}) .5 .59(\mathrm{~m} .1 \mathrm{H}), 4.29(\mathrm{dd} . J=12.6,6.8$ $\mathrm{Hz} .1 \mathrm{H}), 3.36(\mathrm{~s} .2 \mathrm{H}) .2 .23$ (dd, $J=13.4,7.4 \mathrm{~Hz} .1 \mathrm{H}) .2 .07$ (s. 3H). 1.53 (dd. $J=13.4 .4 .4 \mathrm{~Hz}, 1 \mathrm{H}) .1 .10(\mathrm{~s} .3 \mathrm{H}) .0 .86$ (s. $9 \mathrm{H}) .0 .02(\mathrm{~s} .6 \mathrm{H}):{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right) \delta 171.78$, $141.54,133.32,77.54,75.34 .51 .02,44.65,25.73 .24 .43$.
18.54. 17.14, -5.55: Anal calc for $\mathrm{C}_{15} \mathrm{H}_{28} \mathrm{O}_{3} \mathrm{Si}$ : C, 63.33 ; H . 9.92. Found: C, 63.12: H, 10.09 .
(rel)-(1S, 4 S)-t-Butyldimethyl-[1-methyl-4-(4-phenyldi-sulfanylphenyl)-cyclopent-2-enylmethoxyl-silane (12): Phenyldisulfide ( $2.13 \mathrm{~g}, 9.75 \mathrm{mmol}$ ) was added to a solution of compound $11(1.85 \mathrm{~g}, 6.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(7 \mathrm{~mL})$ at -15 ${ }^{\circ} \mathrm{C} . \mathrm{SnCl}_{4}\left(13 \mathrm{mLL} .1 \mathrm{M}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ was then added slowly to the mixture at the same temperature. The mixture was stirred for 5 h at $0^{\circ} \mathrm{C}$ and quenched by adding a saturated $\mathrm{NaHCO}_{3}$ solution ( 25 mLL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed with brine, dried over anhydrous $\mathrm{MgSO}_{4}$. and filtered. The residue was purified by silica gel column chromatography (EtOAc/hexane. 1:50) to give compound $12(2.33 \mathrm{~g}, 81 \%)$ as a colorless oil. ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}, 300$ $\mathrm{MHz}) \delta 7.37-7.18(\mathrm{~m}, 9 \mathrm{H}) .5 .71$ (dd. $J=5.4,2.1 \mathrm{~Hz}, \mathrm{IH})$. $5.60(\mathrm{dd}, J=5 . \mathrm{I} .1 .5 \mathrm{~Hz} . \mathrm{lH}), 4.29(\mathrm{dd} . J=12.6 .4 .8 \mathrm{~Hz}$. $1 \mathrm{H}) .3 .3 \mathrm{I}(\mathrm{d}, J=10.5 \mathrm{~Hz}, 2 \mathrm{H}) .2 .36(\mathrm{dd}, J=13.8 .8 .4 \mathrm{~Hz}$. $1 \mathrm{H}), 1.61(\mathrm{dd}, J=13.8,4.8 \mathrm{~Hz} .1 \mathrm{H}) .1 .13(\mathrm{~s}, 3 \mathrm{H}) .0 .85(\mathrm{~s}$. 9 H ), 0.01 (s. 6 H ): ${ }^{13} \mathrm{C}$ NMR (CDCl 3.75 MHz$) \delta 141.29$. 140.57. 140.21, 136.32. 133.56. 130.73, 129.36. 128.43. 126.62. 76.98. $74.32 .51 .54,43.48,25.73,24.43$. 18.54. -5.55 ; Anal calc for $\mathrm{C}_{25} \mathrm{H}_{34} \mathrm{OS}_{2} \mathrm{Si}$ : $\mathrm{C}, 67.82$ : H. 7.74 ; S . 14.48. Found: C. 67.99; H. 7.87: S, 14.31.
(rel)-( $\mathbf{1 S}, \mathbf{4 S}$ )-1-[t-(tert-Butyldimethylsilanyloxymethyl)-4-methyl-cyclopent-2-enyl]-benzenethiol (13): To a solution of compound 12 ( 1.83 g .4 .14 mmol ) in anhydrous THF $(40 \mathrm{~mL})$ was slowly added $\mathrm{LiAlH}_{4}(786 \mathrm{mg} .20 .7 \mathrm{mmol})$ at 0 ${ }^{\circ} \mathrm{C}$. The mixture was stirred overnight at room temperature and quenched by adding $1 \mathrm{~N} \mathrm{H}_{2} \mathrm{SO}_{4}(4.5 \mathrm{~mL})$. and the stirred for a further 10 min . The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{H}_{2} \mathrm{O}$. and the organic layer was washed with brine. dried over anhydrous $\mathrm{MgSO}_{4}$. and then filtered. The residue was purified by silica gel column chromatography (EtOAc/ hexane. $1: 20$ ) to give compound 13 ( $429 \mathrm{mg} .31 \%$ ) as a colorless oil. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.35-7.17$ (m. $4 \mathrm{H}), 5.85(\mathrm{dd}, J=5.7,2.4 \mathrm{~Hz}, \mathrm{IH}) .5 .73(\mathrm{dd} . J=4.5 .2 .4 \mathrm{~Hz}$. 1H). 4.29 (m. IH). 3.55 (d. $J=2.7 \mathrm{~Hz} . \mathrm{IH}) .3 .40(\mathrm{~s} . \mathrm{IH})$. $3.29(\mathrm{~d} . J=10.5 \mathrm{~Hz} .2 \mathrm{H}) .2 .22(\mathrm{~m} .1 \mathrm{H}) .1 .70(\mathrm{~m}, 1 \mathrm{H}) .1 .11$ (s. 3 H ). 0.86 (s. 9 H ). 0.02 (s. 6 H ): ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75\right.$ $\mathrm{MHz}) \delta 135.60 .135 .28$. 133.44. 131.04. 128.83. 126.62. 76.48. 75.07. 52.43. 44.43, 25.56. 24.21. 18.32, -5.58: Anal calc for $\mathrm{C}_{10} \mathrm{H}_{30} \mathrm{OSSi}$ : C. $68.20, \mathrm{H}, 9.04$; S. 9.58 . Found: C. 68.36; H, 8.90; S, 9.45.
(rel)-(1S,4S)-1-[+-(Hydoxymethyl)-4-methyl-cyclopent-2-enyl)-benzenethiol (14): TBAF ( 2.14 mL .1 .0 M solution in THF) was added to a solution of compound 13 ( 480 mg . 1.43 mmol ) in THF ( 15 mL ) at $0^{\circ} \mathrm{C}$. The mixture was stirred
for 5 h at rt and concentrated. The residue was purified by silica gel column chromatography (EtOAc/hexane. 2:1) to give compound $14(251 \mathrm{mg}, 80 \%)$ : ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}, 300$ $\mathrm{MHz}) \delta 7.55-7.19(\mathrm{~m} .4 \mathrm{H}), 5.89(\mathrm{dd} . J=5.4 .2 .1 \mathrm{~Hz} .1 \mathrm{H})$, $5.74(\mathrm{dd}, J=5.7 .1 .2 \mathrm{~Hz}, 1 \mathrm{H}) .4 .32(\mathrm{~m}, 1 \mathrm{H}), 3.62(\mathrm{~s} .2 \mathrm{H})$, $3.38(\mathrm{~s}, \mathrm{IH}), 2.28$ (dd. $J=14.4,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.84$ (dd. $J=$ 14.4. $4.2 \mathrm{~Hz} . \mathrm{IH}) .1 .12(\mathrm{~s} .3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}, 75 \mathrm{MHz}\right)$ $\delta 135.38,135.00$. 133.84. $130.96,128.85,126.68$. 69.48, 60.07. 51.43. 39.43. 24.52; Anal calc for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{OS}: \mathrm{C}$, 70.87: H. 7.32; S. 14.55. Found: C, 70.77: H. 7.20; S. 14.68.

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