Synthesis of Cyclic Compounds Having *exo*-Methylene Groups through the Diels-Alder Reactions of Vinyl Allenes Obtained from Propargyl Bromide and Indium

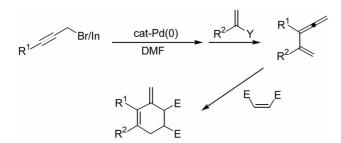
Kooyeon Lee and Phil Ho Lee°

Department of Chemistry and Institute for Basic Science, Kangwon National University, Chunchon 200-701, Korea E-mail: phlee@kangwon.ac.kr Received August 10, 2007

Key Words : Diels-Alder reaction, Indium, exo-Methylene compound, Propargyl bromide, Vinyl allene

Because allene is a very interesting compound having a hybrid character of C-C double and triple bond, vinyl allenes have been recognized as versatile building blocks in organic synthesis.¹ In particular, vinvl allenes take part in not only the Diels-Alder reaction² as the 1.3-diene moiety but also transition metal-catalyzed organic reactions,³ affording efficient synthetic methods for complex ring compounds. However, because it is not easy to effectively prepare a variety of vinyl allenes, its application to organic reactions has been limited despite the potential of vinyl allenes in organic synthesis. Although vinyl allenes were used in the Diels-Alder reactions, development of synthetic method of cyclic compounds having exo-methylene group is still required due to its utility in synthesis of natural products with biological activities.⁴ Recently, we have demonstrated that allenvlindiums generated in situ from indium and propargyl bromides are effective cross-coupling partners in palladiumcatalyzed cross-coupling reactions to produce substituted allenes in excellent vields.⁵ In continuation of our studies directed toward preparative method of vinyl allenes with allenvindium, we describe herein the Diels-Alder reaction of vinyl allenes possessing 3.4- and 4,5-disubstituents and ketone group with a variety of dienophiles to give cyclic compounds having exo-methylene group (Scheme 1).

First, 3-methyl-4-phenyl-1,2.4-pentatriene (1a) as vinyl allene was prepared from the reaction of α -bromostyrene with allenylindium obtained from indium and 1-bromo-2-butyne and then, the Diels-Alder reactions with dienophiles were examined to obtain cyclic compounds having *exo*-methylene group. The results are summarized in Table 1. Reaction of 1a with maleic anhydride (2a) produced the Diels-Alder adduct 3a in 96% yield in toluene at 100 °C for



Scheme 1. Synthesis of cyclic compounds having *exo*-methylene group from vinyl allenes and dienophiles.

3 h (entry 1). Also, 3-ethyl-4-phenyl-1.2,4-pentatriene (1b) reacted with a variety of dienophiles such as 2a, N-phenyl maleimide (2b). dimethyl maleate (2c). and 1,4-naphthoquinone (2d), producing the desired products (3b, 3c, 3d, and 3e) in good to excellent yields in toluene (entries 2-5). Stimulated by these results, tandem cross-coupling reaction of α -bromostyrene with 1-bromo-2-pentyne and indium (1 equiv.) in the presence of Pd(PPh₃)₄ (4 mol%) in DMF followed by Diels-Alder reaction with 2a was attempted to obtain 3b in one-pot procedure. However, the desired product 3b was produced in 7% yield in DMF. Although the cross-coupling product 1b was produced smoothly, the following Diels-Alder reaction did not proceed effectively. Treatment of 1b with 2a did not proceed in THF and DMF. In addition, hetero Diels-Alder reactions were tested. Subjecting 1b to ethyl glyoxylate (2e) provided exclusively 3.6-dihydro-2H-pyran (3f) having exo-methylene group in 80% yield (entry 6). No constitutional isomeric product is formed in this reaction, indicating that the electron-rich central carbon of vinyl allene preferentially adds to the more electron-deficient carbon of dienophile. Although compound 1b reacted with ethyl acrylate (2f) to afford mixture of

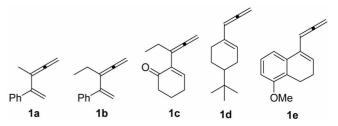


Figure 1. Vinyl allenes.

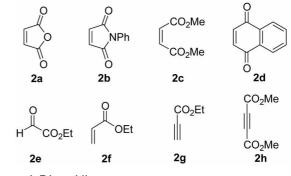




Table 1. [4+2] Cycloaddition reaction of vinyl allenes with dienophiles"

Entry	Reactants	Time (h)	Products		Yield $(\%)^b$
1	1a / 2a	3	Ph O	3a	96
2	1b / 2a	3	Ph O	3b	80
3	1b / 2b	4	Ph O	3c	85
4	1b / 2c	18	Ph CO ₂ Me CO ₂ Me	3d	30 70 ^c
5	1b / 2d	5	Ph O	3e	43 87 ^c
6	1b / 2e	20	Ph CO ₂ Et	3f	80 ^d
7	1b / 2f	20	Ph O	3g	76 ^e (1:2) ^f
8	1c / 2a	13		3h	20 ^g 77
9	1c / 2b	13	O O O O O O O O O O O O O O O O O O O	3i	40 ^g 83
10	1c / 2e	16	CO2Et	3j	30 ^h 87 ⁱ
11	1d / 2b	12	N-Ph O	3k	92(1:1.5) ^{cj}
12	1e / 2b	16	M-Ph OMe	31	74

"Reactions were carried out with vinyl allene (1 equiv.) and dienophile (1 equiv.) unless otherwise noted. Reaction conditions: Toluene (100 °C) for entries 1-7 and 11-12. CH_2CI_2 (25 °C) for entries 8 and 9. CH_2CI_2 (65 °C) for entry 10. ^hIsolated yields. 'Dienophile (3 equiv.) was used. 'Dienophile (30 equiv.) was used. 'Dienophile (30 equiv.) was used. 'Dienophile (5 equiv.) was used. 'Regioisomeric ratio. Ethyl 4-ethyl-5-methylene-3-phenylcyclohex-3-enecarboxylate was produced in major. ^gToluene (25 °C) was used. ^hCH₂CI₂ (25 °C) was used. 'Dienophile (2 equiv.) was used. ^hDienophile (2 equiv.) was used. ^hCH₂CI₂ (25 °C) was

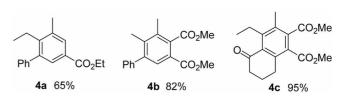


Figure 3. Benzene derivatives from Diels-Alder reaction and aromatization.

regioisomeric products in 76% vield, ethyl 4-ethyl-5-methylene-3-phenylcyclohex-3-enecarboxylate was produced in major (entry 7). Although exposure of vinyl allene 1c to 2a gave exo-methylenecyclohexene derivative 3h in 20% yield in toluene, adduct 3h was produced in 77% yield in CH₂Cl₂ at room temperature for 13 h (entry 8). In the case of 2b, [4+2] cycloaddition adduct **3i** was obtained in 83% yield in CH2Cl2 at room temperature for 13 h (entry 9). Reaction of 1c with 2e furnished pyran 3j having exo-methylene group in 87% vield with complete regioselectivity (entry 10). Vinyl allenes (1d and 1e) obtained from vinvl triflates and allenvlindium were treated with 2b to produce the desired products (3k and 3l) in 92% (dr = 1:1.5) and 74% yields, respectively (entries 11 and 12). Dienophiles having triple bond such as ethyl propiolate (2g) and DMAD (2h) were used to Diels-Alder reaction to give rise to multi-substituted benzene derivatives. Reaction of 1b with 2g produced selectively ethyl 4-ethyl-3-methyl-5-phenyl benzoate (4a) in 65% yield in toluene at 100 °C for 20 h. Compound 1a and 1c were treated with DMAD to afford 4b and 4c in 82% (toluene, 100 °C. 4 h) and 95% (CH₂Cl₂, 65 °C. 15 h) yields, respectively, through Diels-Alder reaction followed by aromatization.

In summary, we have shown that cyclic compounds having *exo*-methylene group were selectively produced through the Diels-Alder reaction of vinyl allenes obtained from propargyl bromide and indium with a variety of symmetric and unsymmetric dienophiles in good to excellent yields. Also, dienophiles having triple bond gave multi-substituted benzene derivatives through Diels-Alder reaction followed by aromatization.

Experimental Section

5-Methyl-4-methylene-6-phenyl-3a,4,7,7a-tetrahydroisobenzofuran-1,3-dione (3a): A mixture of 3-methyl-4phenyl-1,2,4-pentatriene (1a) (46.8 mg, 0.3 mmol) and maleic anhydride (29.4 mg, 0.3 mmol) in toluene (1.0 mL) was heated at 100 °C for 4 h. The reaction mixture was cooling to room temperature and then quenched with NaHCO₃ (sat. aq.). The aqueous layer was extracted with CH₂Cl₂ (20 mL × 3) and the combined organics were washed with brine. dried with MgSO₄ and concentrated. Recrystallization using methylene chloride and *n*-hexane gave **3a** (74.0 mg, 96%) as a white solid. ¹H NMR (300 MHz, CDCl₃. 25 °C, TMS) δ 7.40-7.28 (m. 3H), 7.18-7.15 (m, 2H). 5.52 (s. 1H), 5.47(s, 1H), 4.06 (d. *J* = 9.34 Hz. 1H), 3.56 (ddd, *J* = 9.58, 6.66, 3.19 Hz, 1H), 2.94 (dd, *J* = 16.92,

Notes

135.4. 129.6. 128.4. 128.1, 127.4. 116.2, 46.2, 40.0, 29.4. 17.0; IR (film) 2951. 1731. 1434. 1241. 1060. 912. 818, 759 cm⁻¹; m.p. 108-109 °C; HRMS (EI) calcd for $C_{16}H_{14}O_3$ M⁺ 254.0943, found 254.0945.

5-Ethyl-4-methylene-6-phenyl-3a,4,7,7a-tetrahydroisobenzofuran-1,3-dione (3b): ¹H NMR (400 MHz, CDCl₃. 25 °C. TMS) δ 7.38-7.34 (m, 2H), 7.31-7.28 (m, 1H). 7.14-7.11 (m, 1H). 5.50 (d, J = 0.7 Hz. 1H), 5.45 (s. 1H), 4.04 (d, J = 9.51 Hz. 1H). 3.56 (ddd. J = 9.39, 6.56, 2.70 Hz, 1H), 2.87 (dd. J = 16.24, 2.81 Hz. 1H). 2.67 (ddd. J = 16.25, 6.56, 2.23 Hz, 1H), 2.40-2.31 (m. 1H), 2.23-2.14 (m. 1H). 0.88 (t, J = 7.40 Hz, 3H): ¹³C NMR (100 MHz. CDCl₃): δ 173.5. 171.3. 141.3. 137.0, 135.0. 134.4. 128.5, 127.6, 127.3. 116.0, 47.7. 40.3, 29.8. 22.9. 13.3: IR (film) 1844, 1781, 1700. 904 cm⁻¹: m.p. 135-136 °C: MS (EI) *m*: 2 268 (M⁻).

5-Ethyl-4-methylene-2,6-diphenyl-3a,4,7,7a-tetrahydroisoindole-1,3-dione (3c): ¹H NMR (400 MHz. CDCl₃, 25 °C. TMS) δ 7.48-7.45 (m, 2H). 7.40-7.32 (m. 3H), 7.27-7.24 (m, 3H). 7.17-7.14 (m. 2H), 5.50 (s. 1H), 5.40 (s. 1H), 3.93 (d, *J* = 8.87 Hz. 1H). 3.45-3.41 (m. 1H). 3.00 (dd, *J* = 15.73. 2.27 Hz. 1H). 2.67 (ddd, *J* = 15.70. 6.39. 2.37 Hz, 1H). 2.48-2.39 (m. 1H), 2.26-2.17 (m. 1H). 0.85 (t, *J* = 7.25 Hz, 3H): ¹³C NMR (100 MHz. CDCl₃) δ 178.5, 176.3. 141.7, 137.2. 137.1. 134.7. 132.1, 129.2, 128.6. 128.3, 127.9, 127.0. 126.3. 115.1. 48.4. 39.9. 30.4. 23.0, 13.6: IR (film) 1781. 1715. 1381. 1175 cm⁻¹: m.p. 141-142 °C; MS (EI) *m*:*z* 343 (M⁺).

Dimethyl 4-ethyl-3-methylene-5-phenyl-cyclohex-4-ene-1,2-dicarboxlate (3d): ¹H NMR (400 MHz, CDCl₃. 25 °C. TMS) δ 7.36-7.31 (m, 2H). 7.28-7.24 (m, 1H), 7.15-7.12 (m. 2H), 5.30 (s. 1H). 5.23 (s. 1H), 3.93 (d, J = 3.56 Hz. 1H). 3.72 (s, 3H). 3.69 (s, 3H). 3.05-2.94 (m. 2H), 2.69-2.58 (m. 1H), 2.21-2.05 (m. 2H). 0.91 (t. J = 7.43 Hz, 3H): ¹³C NMR (100 MHz, CDCl₃) δ 173.5. 172.1. 142.9, 139.5, 137.2. 133.0. 128.3. 128.2. 127.7, 126.7. 113.9, 52.0, 48.6. 41.1. 32.3, 22.1. 13.7; IR (film) 2952, 1739, 1205 cm⁻¹; MS (EI) *m*:*z* 314 (M⁻).

2-Ethyl-1-methylene-3-phenyl-1,4,4a,9a-tetrahydroanthraquinone (3e): ¹H NMR (400 MHz. CDCl₃, 25 °C. TMS) δ 8.10-8.07 (m, 1H). 8.05-8.03 (m. 1H), 7.78-7.72 (m. 2H), 7.38-7.33 (m. 2H), 7.29-7.25 (m. 3H), 5.31 (s, 1H). 4.88 (s. 1H). 4.11 (dt. *J* = 3.61, 1.41 Hz, 1H). 3.58 (q. *J* = 5.56 Hz, 1H). 2.95 (dd. *J* = 18.08, 5.97 Hz, 1H) 2.61 (dd. *J* = 18.06, 5.45 Hz, 1H), 2.17-2.11 (m. 2H), 0.92 (t. *J* = 7.45 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 197.0, 196.9, 142.9, 137.1, 136.4, 135.1, 134.4, 134.3, 134.1, 133.8, 128.3, 127.7, 126.9, 126.8, 127.7, 113.7, 54.8, 47.8, 32.8, 22.1, 14.0; IR (film) 3060, 1695, 1594, 703 cm⁻¹; m.p. 101-102 °C: MS (EI) *m*z 328 (M⁺).

Ethyl 4-ethyl-3-methylene-5-phenyl-3,6-dihydro-2*H*pyran-2-carboxylate (3f): ¹H NMR (400 MHz, CDCl₃, 25 °C. TMS) δ 7.38-7.28 (m, 3H). 7.18-7.15 (m, 2H). 5.32 (s. 1H), 5.15 (s. 1H), 4.88 (s. 1H). 4.71 (d. *J* = 17.13 Hz. 1H). 4.37 (d. *J* = 17.13 Hz. 1H). 4.28 (q. *J* = 7.11 Hz. 2H). 2.19 (q. *J* = 7.29 Hz. 2H), 1.32 (t. *J* = 7.17 Hz. 3H). 0.97 (t. *J* = 7.47 Hz, 3H); ¹³C NMR (100 MHz. CDCl₃) δ 170.5, 138.1, 136.0, 135.6, 131.3, 128.4, 128.3, 127.5, 111.1, 68.0, 61.2, 21.2, 14.3, 13.8; IR (film) 2975, 1739, 1609, 1465, 1130 cm⁻¹; MS (EI) *m*/*z* 199 (M⁺-CO₂Et).

Ethyl 4-ethyl-5-methylene-3-phenylcyclohex-3-enecarboxylate (3g, major compound): ¹H NMR (400 MHz, CDCl₃. 25 °C, TMS) δ7.35 (m, 2H), 7.27 (m. 1H), 7.16 (d. *J* = 6.83 Hz. 2H), 5.09 (s, 1H). 4.97 (s, 1H). 4.14 (q, *J* = 7.11 Hz. 2H). 2.83-2.77 (m, 1H). 2.71 (dd. *J* = 14.1, 3.46 Hz, 1H), 2.65-2.54 (m, 3H), 2.18-2.07 (m. 2H). 1.25 (t. *J* = 7.16 Hz, 3H). 0.93 (t, *J* = 7.4 Hz. 3H); ¹³C NMR (100 MHz. CDCl₃) δ 174.8, 143.4, 141.0. 136.6. 134.6, 128.2, 127.7. 126.6. 110.1, 60.4. 40.2, 35.8. 35.1. 21.9, 14.3. 14.2: IR (film) 2932. 1732, 1442. 1178, 1038, 884, 761, 701 cm⁻¹; MS (EI) *m*: 270 (M⁻).

Ethyl 3-ethyl-2-methylene-4-phenylcyclohex-3-enecarboxylate (3g, minor compound): ¹H NMR (400 MHz, CDCl₃. 25 °C, TMS) δ 7.33 (t. J = 7.38 Hz. 2H), 7.24 (m, 1H), 7.12 (m, 2H), 5.22 (s, 1H), 4.98 (s, 1H), 4.19 (m. 2H), 3.40 (t. J = 5.18 Hz. 1H). 2.49 (m, 1H), 2.33 (dt, J = 18.4, 5.20 Hz, 1H), 2.25-2.12 (m. 3H), 1.98 (m. 1H). 1.28 (t, J = 7.10 Hz. 3H), 0.95 (t. J = 7.44 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 173.9, 143.8. 140.5, 137.9. 133.7, 128.1, 127.6, 126.4, 111.1. 60.4, 47.5, 31.1, 25.7, 22.2. 14.3. 13.9; IR (film) 2931, 1732. 1605. 1443, 1373. 1308, 1258, 1155, 1040. 759 cm⁻¹: MS (EI) *m* iz 270 (M⁺).

5-Ethyl-4-methylene-3a,7,8,9,9a,9b-hexahydro-4*H*naphtho[1,2-c]furan-1,3,6-trione (3h): ¹H NMR (400 MHz, CDCl₃, 25 °C, TMS) δ 5.67 (s. 1H), 5.63 (s. 1H). 4.02 (d. *J* = 9.52 Hz, 1H). 3.46 (dd, *J* = 9.50, 5.15 Hz, 1H). 3.15 (octet, *J* = 7.27, 12.02, 14.68 Hz, 1H). 2.68-2.62 (m. 1H), 2.57-2.48 (m, 1H), 2.46-2.35 (m, 1H), 2.33-2.03 (m, 2H), 2.09-1.97 (m, 2H), 1.76-1.63 (m, 1H). 1.00 (t. *J* = 7.39 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 199.2, 171.0, 170.4, 153.3, 137.6, 131.2, 121.0, 49.7, 44.7, 40.0, 37.2, 25.8, 25.3, 21.2, 13.1: IR (film) 1885, 1780, 1662, 1549, 1200 cm⁻¹: m.p. 156 °C; MS (EI) *m*/z 260 (M⁺).

5-Ethyl-4-methylene-2-phenyl-3a,7,8,9,9a,9b-hexahydro-4H-benzo[e]isoindole-1,3,6-trione (3i): ¹H NMR (400 MHz, CDCl₃. 25 °C, TMS) δ 7.65-7.32 (m, 3H). 7.15-7.13 (m, 2H). 5.63 (s, 1H). 5.62 (s, 1H), 3.90 (d, J = 9.04 Hz, 1H). 3.46 (dd, J = 4.92, 8.87 Hz, 1H). 3.19 (octet, J = 7.29, 11.72, 14.63 Hz, 1H). 2.72-2.67 (m, 1H), 2.54-2.33 (m, 3H), 2.28-2.17 (m, 1H), 2.08-1.99 (m, 2H), 1.76-1.62 (m, 1H), 1.00 (t, J = 7.33 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃); δ 199.9, 176.4, 175.0, 153.5, 139.7, 131.7, 131.1, 129.2, 128.7, 126.4, 120.0, 49.9, 43.7, 40.0, 38.2, 25.8, 25.5, 21.4, 13.4; IR (film) 1713, 1668, 1498 cm⁻¹; m.p. 160-161 °C; MS (EI) *m*: 2 335 (M⁻).

Ethyl 4-ethyl-3-methylene-5-oxo-3,5,6,7,8,8a-hexahydro-2*H*-chromene-2-carboxylate (3j): ¹H NMR (400 MHz, CDCl₃. 25 °C. TMS) δ 5.56 (s. 1H), 5.27 (s. 1H), 4.74 (t. *J* = 1.53 Hz, 1H). 4.42-4.30 (m. 3H). 2.59-2.28 (m. 5H). 2.07-2.00 (m. 1H). 1.92-1.82 (m. 1H). 1.75-1.63 (m. 1H). 1.35 (t. *J* = 7.10 Hz, 3H). 1.13 (t. *J* = 7.45 Hz, 3H): ¹³C NMR (100 MHz, CDCl₃) δ 201.8, 169.2, 142.6. 137.3, 135.4, 114.1, 76.8, 61.5, 42.7, 31.5. 21.8. 19.2. 14.4. 14.2: IR (film) 2940, 490 Bull. Korean Chem. Soc. 2008, Vol. 29, No. 2

1735, 1685, 1457, 1192 cm⁻¹; MS (EI) $m^2 z 264$ (M⁺).

8-*tert*-**Butyl-4**-methylene-2-phenyl-3a,4,6,7,8,9,9a,9boctahydrobenzo[e]isoindole-1,3-dione (3k, major compound): ¹H NMR (400 MHz, CDCl₃. 25 °C, TMS) δ 7.44 (t. J = 7.70 Hz, 2H). 7.36 (t, J = 7.47 Hz. 1H), 7.23 (d. J = 7.56 Hz, 2H). 6.04 (s. 1H). 5.30 (s. 1H), 5.17 (s. 1H), 3.80 (d, J = 8.30 Hz, 1H). 3.36 (dd, J = 8.01. 6.86 Hz, 1H), 2.77 (m, 1H). 2.37 (m, 1H). 2.25 (m, 1H). 2.02 (td. J = 12.65, 8.87 Hz. 1H), 1.73 (m. 2H). 1.52 (m, 1H), 1.32 (qd, J = 12.48. 5.41 Hz, 1H): ¹³C NMR (100 MHz, CDCl₃) δ 175.8, 175.0. 142.1. 133.0, 130.9. 128.0. 127.4, 125.3, 121.1. 113.5, 43.5. 42.2, 41.4, 32.3. 31.1. 30.1. 26.0, 24.8, 21.5; IR (film) 3463. 2959. 2867, 2249, 1775, 1709 cm⁻¹; MS (EI) *m*:z 349 (M⁺).

4-Methoxy-12-methylene-16-phenyl-6,7,8,12,13,14-hexahydro-16-aza-cyclopenta[a]phenanthrene-15,17-dione (3l): ¹H NMR (400 MHz, CDCl₃, 25 °C, TMS) δ 7.41 (m, 2H). 7.33 (m, 1H). 7.23 (m, 3H), 7.14 (t. J = 8.05 Hz, 1H), 6.85 (d, J = 2.25 Hz, 1H). 6.75 (d. J = 8.02 Hz, 1H), 5.52 (s. 1H). 5.35 (s. 1H), 3.92 (d. J = 8.09 Hz, 1H), 3.83 (s, 3H), 3.50 (dd. J = 8.07, 5.44 Hz, 1H). 3.27 (td, J = 16.33, 3.32 Hz, 1H), 2.88 (m, 1H). 2.55 (qd, J = 13.06, 3.81 Hz, 1H). 2.33 (m, 1H), 2.13 (m. 1H): ¹³C NMR (100 MHz, CDCl₃) δ 175.0, 174.8, 155.8, 135.6, 134.8, 133.0, 130.9, 127.9, 127.3, 126.4, 125.5, 125.3, 121.1, 116.1, 114.7, 108.1, 54.5, 45.0, 43.3, 34.3, 23.0, 21.8. IR (film) 2834, 1710, 1572, 1499, 1387, 1261 cm⁻¹; m.p. 189-190 °C; MS (EI) *m*/z 371 (M⁺).

Ethyl 4-ethyl-5-methyl-3-phenyl benzoate (4a): ¹H NMR (400 MHz, CDCl₃, 25 °C, TMS) δ 7.84 (d, J = 1.33 Hz, 1H), 7.70 (d, J = 1.69 Hz, 1H), 7.46-7.37 (m, 3H), 7.30-7.27 (m, 2H), 4.35 (q, J = 7.09 Hz, 2H), 2.60 (q, J = 7.52 Hz, 2H), 2.44 (s, 3H), 1.36 (t, J = 7.18 Hz, 3H), 0.99 (t, J = 7.51 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.8, 145.6, 142.4, 141.8, 136.6, 130.5, 129.2, 129.0, 128.0, 127.3, 127.0, 60.8, 23.3, 19.7, 14.4, 14.1; IR (film) 2971, 2930, 1717, 1223 cm⁻¹; MS (EI) *m*/z 268 (M⁺).

Dimethyl 3,4-dimethyl-5-phenylphthalate (4b): A mixture of 3-methyl-4-phenyl-1.2.4-pentatriene (1a) (46.86 mg, 0.3 mmol) and DMAD (42.63 mg, 0.3 mmol) was heated in toluene (1.0 mL) at 100 °C for 4 h. The reaction mixture was quenched with NaHCO₃ (sat. aq.). The aqueous layer was extracted with CH₂Cl₂ (20 mL × 3) and the combined organics were washed with water and brine, dried over MgSO₄, filtered, and concentrated under reduced pressure. It was purified by recrystallization (CH₂Cl₂:Hex) to give **4b** (73.0 mg, 82%). ¹H NMR (300 MHz, CDCl₃, 25 °C. TMS) δ 7.78 (s, 1H). 7.47-7.36 (m, 3H), 7.28-7.25 (m, 2H), 3.99 (s, 3H), 3.86 (s, 3H), 2.30 (s, 3H), 2.21 (s, 3H): ¹³C NMR (100 MHz, CDCl₃) δ 170.5, 166.1, 143.1, 141.0, 140.4, 134.6, 134.4, 129.2, 129.1, 128.2, 127.3, 124.3, 52.6, 52.3, 17.8.

17.1; IR (film) 1730, 1432, 1323 cm⁻¹; m.p. 107-108 °C; MS (EI) *m/z* 298 (M⁺).

Dimethyl 4-ethyl-3-methyl-5-oxo-5,6,7,8-tetrahydronaphthalene-1,2-dicarboxylate (4c): ¹H NMR (400 MHz, CDCl₃. 25 °C, TMS) δ 3.90 (s. 3H), 3.87 (s, 3H). 2.99 (q, J =7.34 Hz, 2H). 2.96 (t. J = 6.26 Hz. 2H). 2.67 (t, J = 6.75 Hz, 2H). 2.33 (s, 3H), 2.04 (quintet, J = 6.50 Hz, 2H), 1.20 (t. J =7.33 Hz. 3H): ¹³C NMR (100 MHz. CDCl₃) δ 200.1, 169.0, 168.1. 148.4. 141.1. 136.4. 133.7. 133.4, 128.5, 52.6. 52.5. 40.7. 28.1, 23.9. 22.3. 15.6, 14.1; IR (film) 2953, 2359. 1739. 1690. 1436 cm⁻¹: MS (EI) *m*:*z* 304 (M⁻).

Acknowledgment. This work was supported by KOSEF through the National Research Lab. Program funded by the Ministry of Science and Technology (No. M10600000203-06J0000-20310). by the Korea Science and Engineering Foundation (KOSEF, R01-2006-000-11283-0), and by the CMDS at KAIST. The NMR and mass data were obtained from the central instrumental facility in Kangwon National University.

References

- (a) Patai, S., Ed. The Chemistry of Ketenes, Allenes and Related Compounds, Wiley: New York, 1980. (b) The Chemistry of the Allenes, Landor, S. R., Ed.; Academic: London, 1982. (c) Smadja, W. Chem. Rev. 1983, 83, 263. (d) Schuster, H. F.; Coppola, G. M. Allenes in Organic Synthesis; Wiley: New York, 1984. (e) Zimmer, R.; Dinesh, C. U.; Nandanan, E.; Khan, F. A. Chem. Rev. 2000. 100, 3067. (f) Modern Allene Chemistry, Krause, N., Hashmi, A. S. K., Eds.; Wiley-VCH: Weinheim, 2004.
- (a) Angelov, C. M.; Mondeshka, D. M.; Tancheva, T. N. Chem. Commun. 1985, 647. (b) Reich, H. J.; Eisenhart, E. K.; Whipple, W. L.; Kelly, M. J. J. Am. Chem. Soc. 1988, 110, 6432. (c) Dulcere, J. P.; Agati, V.; Faure, R. Chem. Commun. 1993, 270. (d) Spino, C.; Thibault, C.; Gingras, S. J. Org. Chem. 1998, 63, 5283. (e) Regás, D.; Afonso, M. M.; Rodríguez, M. L.; Palenzuela, J. A. J. Org. Chem. 2003, 68, 7845. (f) Regás, D.; Ruiz, J. M.; Afonso, M. M.; Palenzuela, J. A. J. Org. Chem. 2006, 71, 9153.
- (a) Murakami, M.: Itami, K.: İto, Y. Angew. Chem. Int. Ed. 1995.
 34. 2691. (b) Murakami, M.: Itami, K.; Ito, Y. J. Am. Chem. Soc. 1997. 119, 7163. (c) Lee, P. H.; Lee, K. Angew. Chem. Int. Ed. 2005. 44, 3253. (d) Lee, P. H.; Lee, K.; Kang, Y. J. Am. Chem. Soc. 2006. 128, 1139. (e) Funami, H.; Kusama, H.; Iwasawa, N. Angew. Chem. Int. Ed. 2007, 46, 909.
- (a) Cavallito, C. J.; Fruehauf, D. M.; Bailey, J. H. J. Am. Chem. Soc. 1948, 70, 3724. (b) Lee, K. H.; Hall, I. H.; Mar, E. C.; Starnes, C. O.; El Gebaly, S. A.; Waddell, T. G.; Hadgraft, R. I.; Ruffner, C. G.; Weidner, I. Science 1977, 196, 533. (c) Spring, O.; Albert, K.; Gradmann, W. Phytochemistry 1981, 20, 1883. (d) Park, B. K.; Nakagawa, M.; Hirota, A.; Nakayama, M. J. Antibiot. 1988, 41, 751. (e) Liu, S.; Liu, H.; Yan, W.; Zhang, L.; Bai, N.; Ho, C.-T. Bioorg, Med. Chem. Lett. 2004, 14, 1101.
- Lee, K.; Seomoon, D.; Lee, P. H. Angew. Chem. Int. Ed. 2002, 41, 3901.