Identification of New Compounds from *Catharanthus roseus* Hairy Root Cultures

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Two new compounds lanast-5,8-dien-3 β -ol-27-oic acid-3 β -D-glucopyranosyl (4'-1")-10",11"-dimethoxy anthracene (1), 2-methoxy-6-(*n*-nonacontan-5",6"-dionyl)-11-hydroxy-13-methyl-11 β -D-rhamnopyranoside anthracene (2) have been isolated from the hairy root cultures of *Catharanthus roseus*. Their structures have been elucidated with the help of 500 MHz NMR using one- and two-dimensional NMR in combination with IR, EI/MS, FAB/MS and HRFABMS spectroscopy.

Key Words : Catharanthus roseus, Apocynaceae, Cultures hairy roots, New constituents.

Introduction

The periwinkle, Catharanthus roseus (Apocynaceae), is a widely used ornamental and medicinal plant. C. roseus is an herbaceous shrub¹ and has been extensively studied due to its production of two valuable alkaloids, vincristine and vinblastine which are used in the treatment of human neoplasm, and an alkaloid from the root, ajmalicine which is used in the treatment of circulatory disorders and hypertension. Biologically indole alkaloids produced by plants are believed to play a role as antimicrobial and antifeeding compounds,²³ This madagasean periwinkle produces numerous indole alkaloids which have important therapeutic activities.⁴ Only few phenolic compounds have been reported in this genus.5.6 Recently, two flavonols trisaccharides of kaempferol and quercetin have been isolated and identified.⁷ Several indole alkaloids have been isolated from the C. roseus cell suspension cultures.8.9 However, the production of the most valuable compounds reported from this plant. vincristine and vinblastine that are terpenoid indole alkaloids,10 has not yet been achieved in these cultures. Besides indole alkaloids, the presence of anthocyanidins.¹¹ phenolics,^{9,12} and terpenoid compounds^{8,9} in the cultures of C. roseus has been reported. As part of its secondary metabolism this plant produces pharmaceutically valuable terpenoid indole alkaloids such as vincristine and vinblastine which are used as anticancer drugs. A very low yield of these compounds is a major motivation of the research interest in this plant. Although the hairy root cultures do not produce these two bisindole alkaloids that consist of catharanthine and vindoline, they have been shown to produce catharanthine and tabersonine. This paper deals with the isolation and structural elucidation of two new compounds (1) and (2) on the basis of spectral data and chemical reactions and also reported one new and three known compounds, β -sitosterol, 3-epibetulinic acid, *n*-pentadecanyl octa-dec-19-en-oate.¹³ Due to high significance of medicinal natural products of these plants roots the work such as terpenoids, alkaloids analysis has extensively been conducted. The aim of the present investigation is to report some of the new findings in the form of natural product from culture roots of C, *roseus*.

Results and Discussion

Compound I, was obtained as a dark yellow crystalline mass and its molecular formula $(C_{52}II_{70}O_{10})$ was deduced from HR-FABMS. It gave positive tests for triterpenic glycosides and produced effervescences with sodium bicarbonate indicating the presence of carboxylic acid in the molecule. Its IR spectrum showed characteristic absorption



Figure 1. Key HMBC correlations of compound 1 and 2.



Figure 2. Fragmentation pattern of compound 1.

bands for hydroxyl groups $(3577, 3415, 3180 \text{ cm}^{-1})$, carboxylic group (3381, 1695 cm^{-1}) and unsaturation (1613 cm⁻¹). The positive ion FAB-MS of 1 displayed a molecular ion peak at m/z 855 corresponding to a tetracyclic triterpenic glycoside linked with an anthracene moiety C₅₂H₇₀O₁₀. The prominent ion peaks generated at m/z in the EI-MS 455 [M- $C_6H_{10}O_5-C_{16}H_{13}O_2$, 438 [M-C₆H₁₂O₆-C₁₆H₁₃O₂]⁺, indicated that a lanostene-type triterpene carbon framework was present in the molecule. The ion fragments arose at m/z410 [455-COOH]⁺, 395 [410-Me]⁻, 365 [395-2xMe]⁺, 383 [455-CH(CH₃)COOH]⁺ and 350 [365-Me]⁺ suggested that the carboxylic group was present in the side chain. The ion fragments produced at m/z 265 [C₈H₁₄-C_{11,12} fission]⁻, 184 $[C_{14,15}-C_{13,17}$ fission], 169 $[184-Me]^+$, 139 $[184-Me]^+$ $COOH]^-$, 143 $[C_8H_{14}O_2 \text{ side chain}]^-$, 294 $[455-\text{side chain}]^-$ H₂O]⁻ and 264 [294–2 x Me]⁻ supported the presence of the hydroxyl group in ring A, placed at C-3 on the basis of biogenetic consideration, two vinylic linkages in ring B and saturated nature of rings C and D. The ion peaks at m/z 237 was formed due to elimination of the anthracene moiety $[C_{14}H_7 (OCH_3)_2]^-$ which yielded the ion peaks at m/z 222 $[237-Me]^{-}$ and 175 $[237-2 \times OMe]^{+}$. The fragmentation

pattern of compound 1 is shown in Figure 2.

The 'H NMR spectrum of 1 two one-proton ortho-, metacoupled protons at δ 7.12 (J = 7.9, 3.0 Hz) and 7.08 (J = 7.25, 3.0 Hz) assigned to H-2" and H-4" respectively. Two one-proton ortho-coupled doublets at δ 7.28 (J = 7.5 Hz) and 7.46 (J = 7.5 Hz) were attributed to H-8" and H-9" respectively. A one-proton ortho-, coupled at δ 7.10 (J=7.9, 7.25 Hz) and two one-proton broad signals at δ 7.52 and 8.14 were ascribed to H-3", H-6" and H-13" respectively. A one-proton doublet at δ 5.32 (J = 2.5 Hz) was accounted to vinylic H–6. Three one-proton doublet at δ 4.42 (J = 7.2 Hz), 3.12 (J = 10.5 Hz) and 3.08 (J = 10.5 Hz) were assigned correspondingly to anomeric H-1' and hydroxyl methylene protons H_2 -6'. The remaining carbinol protons of the sugar molety appeared in the range of $\delta 3.89 - 3.28$ as multiplets. A one-proton double doublet at δ 3.67 with coupling interactions of 5.0 and 10.5 Hz was accounted to α -oriented H–3 carbinol proton. Two three-proton doublets at $\delta 1.19$ (J = 6.65 Hz), 1.17 (J = 6.65 Hz) were attributed to C-21 and C-26 secondary protons, respectively. Five three-proton broad signals at δ 0.88, 0.93, 1.11, 1.25 and 1.31 were associated with C-18, C-19, C-28, C-29 and C-30 tertiary methyl protons, all attached to the saturated carbons. The remaining methylene and methine protons resonated in the range $\delta 2.98 - 1.31$. Two methoxy proton signals attached to C-10" and C-11" showed at δ_{11} 3.72 and 3.78 as broad singlets.

The ¹³C NMR spectrum of 1 exhibited important signals for carboxylic carbon at δ 180.21 (C–27), vinylic carbons at δ 143.25 (C–5), 122.63 (C–6), 137.25 (C–8) and 127.96 (C– 9), anomeric carbon at δ 109.72 (C-1), sugar carbons between δ 60.33–69.17 and 90.70, carbinol carbon at δ 73.93 (C-3), methoxy carbons at δ 57.05 and 53.42 and aromatic carbons between δ 168.73–106.88. The multiplicity of each carbon was determined by analysis of DEPT spectrum of 1. The ¹H-¹H COSY spectrum of 1 showed correlation of H-3 with H-1', H₂-2, H₂-1 and H₃-28; H-6 with H2-7, H3-28 and H3-29; H-2" with H-3" and H-4" and H-8" with H-9". The ¹H-¹³C HETCOR spectrum displayed interaction of C-27 with H₃-26, C-5 with H₃-28 and H₃-29, C-4' with H-5', H₂-6', H-3', and H-2", and C-10" with OMe. In HMBC spectra, H-4' showed a correlation with C-1" which confirmed the connecting position of anthracene moiety to the glucose unit. Acid hydrolysis of 1 yielded Dglucose as one of the component (TLC comparable). On the basis of spectral data analysis and chemical reactions the structure of 1 has been established as lanast-5, 8-dien-3 β -ol-27-oic acid-3 β -D-glucopyranosyl (4'-1")-10",11"-dimethoxy anthracene. This is a new lanastenenol glycoside isolated for the first time.

Compound **2**, was obtained as a dark yellow crystalline mass from EtOAc:MeOH (9.5:0.5) eluants. It responded positively to glycosidic tests. Its IR spectrum displayed characteristic absorption bands for hydroxyl groups (3423, 3380 cm⁻¹), carbonyl groups (1704 cm⁻¹) and long aliphatic chain (747, 722 cm⁻¹). Its positive FAB MS showed a molecular ion peak at m/z 819 corresponding to an

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Figure 3. Fragmentation pattern of compound 2.

anthracene glycoside with C-29 aliphatic side chain. The prominent ion peaks generated in the EI MS m/z 435 [M– anthracene glycoside]⁻, 383 [anthracene glycoside]⁻, 220 [383 – rhamnose]⁺, 205 [220 – Me]⁻, 191[205 – Me]⁺, 236 [383 – C₆H₁₁O₄]⁺, 221[236 – Me]⁻, 206 [221 – Me]⁻, 147 [C₆H₁₁O₄]⁺ and 163 [C₆H₁₁O₅]⁻ indicated that rhamnose sugar was attached to anthracene carbon framework containing a methyl, a methoxy and C₂₀-side chain. The ion peaks arose at m/z 379 [C_{4*} – C_{5*} fission]⁺, 351 [C_{5*} – C_{6*} fission]⁺ and 322 [C_{6*} – C_{7*} fission]⁺ indicated the location of the carbonyl groups at C-5" and C-6". The fragmentation pattern of compound **2** is shown in Figure 3.

The ¹H-NMR spectrum of **2** showed a meta-coupled doublet at δ 7.46 (J = 3.0 Hz), three meta-, ortho coupled double doublets at δ 7.31 (J = 7.5, 3.0 Hz), 7.08 (J = 7.5, 3.0 Hz) and 7.28 (J = 7.0, 2.5 Hz), one ortho-coupled doublet at δ 7.44 (J = 7.5 Hz) and an ortho-coupled double doublet at δ 7.16 (J = 7.0, 7.5 Hz), assigned correspondingly to H–1, H–3, H–8, H–10, H–4 and H–9 respectively. Two three-proton broad signals at δ 1.96 and 3.74 were attributed to C–15 methyl and C–2 methoxy protons, respectively. A one-proton doublet at δ 4.40 (J = 6.0 Hz) and a three proton doublet at δ 1.17 (J = 7.0 Hz) were accounted to anomeric H–1' and secondary methyl Me–6' protons. The remaining hydroxy-methine protons of the sugar moiety resonated between δ 3.94-3.19. Four one-proton doublets at δ 2.48 (J = 7.2 Hz), 2.43 (J = 7.2 Hz) and at δ 2.23 (J = 7.1 Hz) and δ

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Table 1. 1 H (500 MHz) and 13 C (125 MHz) NMR data for compound (1) in MeOD

position	δ _н (ррт)	$\delta_{ m C}$ (ppm)
$1\alpha, 1\beta$	1.98 m. 2.24 m	33.06 t
$2\alpha, 2\beta$	2,41 m, 1,31 m	27.42 t
3	3.67 dd (5.0, 10.5)	73.93 d
4	_	45.01 s
5	-	143.25 s
6	5.32 d (2.5)	122.63 d
7α , 7β	2.98 d (2.5), 2.96 d (2.5)	50.861
8	_	137,25 s
9	_	127.96 s
10	_	39.38 s
11α , 11β	2.75 dd (4.5, 9.1), 2.68 dd (4.5, 5.0)	50,19 t
12α , 12β	2.04 dd (4.5, 9.1), 1.72 dd (4.5, 5.0)	29.90 t
13	_	45.49 s
14	-	51.42 s
15a. 15B	1.36 m. 1.34 m	30.85 t
$16\alpha \cdot 16\beta$	1.94 m. 2.22 m	29.52 t
17	2 ?? m	51.26 d
18	0.88 br s	13.89 0
19	1.11 br s	19.07 q
20	2 70 m	41 14 d
21	1 19 d (6 6)	21.97 a
21 22 (7 22 B	241 m = 201 m	29.44 t
$23\alpha 23\beta$	136 m 164 m	29.32 1
2302.23p 2472.24B	1.42 m 2.01 m	23.68 t
25	2 70 m	23.68 d
26	117 d (6 6)	30.85 a
20	-	180.21 s
28	0.93 hr s	1431.0
29	1.25 br s	1516 a
30	1 31 hr s	32 13 a
1	4 42 4 (7 2)	109.72 d
2'	3.75 m	69.17.d
2	3 38 m	64.57 d
4'	3.28 m	63.88 d
	3.80 m	90 70 d
Sec 6B	3.12 d (10.5) - 3.08 d (10.5)	60.33 t
1"	- (10.5): 5.00 d (10.5)	167.68 %
2"	712 dd (7930)	154.90 4
3"	7 10 dd (7.9, 7.2)	111.07 d
<u>4</u> "	7.08 dd (7.2, 3.0)	118.74 d
		127.47 %
5" 6"	7.52 s	121.473
7"		106.88 s
2 Q11	7.28 d (7.5)	127.96.4
0"	7.46 d (7.5)	127.90 d
7 107		121.04 u 168 60 c
10	_	168 72 0
12"		108.75 8
12"	814 c	100.003
1.7 1.7 1.7 1.7 1.7 1.7 1.7 1.7 1.7 1.7	0.147.5	121.30 U
DMa	- 3 72 br s	57.05 -
OMC	3.72 bi s	53.42 n

J values (in Hz) are in parenthesis.

2.21 (J = 7.1 Hz) were attributed to methylene H₂-4" and H₂-7" protons adjacent to the carbonyl groups. A threeproton triplet at δ 0.86 (J = 7.0 Hz) was due to C-29" Identification of New Compounds from Catharanthus roseus Root Cultures – Bull. Korean Chem. Soc. 2007, Vol. 28, No. 8 – 1297

primary methyl protons. The remaining methylene protons of the aliphatic side chain resonated between $\delta 2.07-1.25$.

The ¹³C-NMR spectrum of **2** showed aromatic carbon signals between δ 167.48-106.39, anomeric carbon at δ 111.62 (C-1'), hydroxy-methine carbons of the sugar moiety between δ 76.98-65.64, carbonyl carbons at δ 207.13 (C-5") and 195.69 (C-6"), methoxy carbon at δ 51.41, methyl carbons at δ 15.32 (Me-15), 26.34 (Me-6') and 14.33 (Me-29") and the remaining methylene carbons between δ 35.22-18.61 (Table 2). The multiplicity of each carbon was determined by analysis of the DEPT spectrum of 2. In ${}^{1}H{}^{-1}H$ COSY spectrum of 2 correlations of Me-15 were observed with H-1: Me-6' with H-4', H-3' and H-2'; H-3 with H-1 and H-4; H₂-1" with H-4 and H-8 and H₃-29" with the adjacent methylene protons. In HMBC spectrum of 2, C-11 showed a correlation with H-1' which indicates the Drhamnose is connected at C-11 of the athracene moiety. The connecting position of the aliphatic group in 2 was decided from the HMBC correlation between H-1" and C-5/C-6/C-7 as well as H-2" and C-6.

Acid hydrolysis of 2 yielded a D-rhamnose as a glycone moiety (TLC comparable). On the basis of the foregoing account the structure of 2 has been established as 2-methoxy- $6-(n-nonacontan-5", 6"-dionyl)-11-hydroxy-13-methyl-11\beta-D-rhamnopyranoside anthracene. This is new anthracene derivative isolated for the first time.$

Experimental Section

Chemicals. All chemicals were of an analytical grade: hexane. ethyl acetate, methanol. ethanol, sulphuric acid and vanillin were purchased from Daejung Chemicals and Metals (Seoul. South Korea). Pre-coated TLC plates (layer thickness 0.5 mm), silica gel for column chromatography (70-230 mesh ASTM) and LiChroprep RP-18 (40-63 μ an) were from Merck (Darmstadt, Germany). Authentic standards of β -sitosterol. oleic acid, and D-glucose were purchased from Sigma-Aldrich (St. Louis, MO, USA).

Instrumentation. Melting points were determined using Electrochemical Engineering (Electrothermal, Seoul, South Korea) model IA9100 melting point apparatus. Specific rotation was measured with an instruments Ltd (Seoul, South Korea) model AA-10 polarimeter. ¹H- and ¹³C -NMR spectra were obtained at 500 and 125 MHz, respectively, using a Bruker Avance model DRX-500 spectrometer at the Seoul National University (SNU), Seoul, South Korea. NMR spectra were obtained in deuterated chloroform and methanol using tetramethylsilane (TMS) as an internal standard. EI/MS and FAB/MS were recorded on JEOL JMS-SX 102A and JEOL JMS-AX 505WA spectrophotometers. respectively, at the Seoul National University. IR spectra were recorded on a Thermo Mattson Infinity Gold FT-IR model 60-AR spectrophotometer at the Korea Institute of Science and Technology (KIST) Seoul, South Korea.

Culture conditions. The hairy root line used in this study was previously generated by infection of *C. roseus* seedling with *Agrobacterium rhizogenes* 15834.¹⁴ The culture media

Table 2. 1H (500 MHz) and ^{13}C (125 MHz) NMR data for compound (2) in MeOD

position	$\delta_{\rm H}({\rm ppm})$	$\delta_{\rm C}({ m ppm})$
1	7.46 d (3.0)	155.44 d
2	-	167.48 s
3	7.31 dd (7.5, 3.0)	122.56 d
4	7.44 d (7.5)	118.67 d
5	_	105.33 s
6	-	150.16 s
7	-	106.39 s
8	7.08 dd (7.5, 3.0)	118.67 d
9	7.16 dd (7.0, 7.5)	124.39 d
10	7.28 dd (7.0, 2.5)	120.26 d
11	-	167.48 s
12	_	126.71 s
13	_	136.49 s
14	-	122.56 s
15	1.96 br s	15. 32 q
OMe	3.74 br s	51.41 q
1'	4.44 d (6.0)	111.62 d
2'	3.94 m	76.98 d
3'	3.24 m	65.64 d
4'	3.19 m	71.03 d
5'	3.51 m	72.51 d
6'	1.17 d (7.0)	2 6. 34 q
$1^{"}\alpha, 1^{"}\beta$	2.87 br s, 2.85 br s	35.22 t
$2^{\circ}\alpha, 2^{\circ}\beta$	2.04 m, 2.07 m	33.41 t
$3^{"}\alpha, 3^{"}\beta$	1.96 m, 1.94 m	30.14 t
$4^{"}\alpha 4^{"}\beta$	2.48 d (7.2), 2.43 d (7.2)	32.15 t
5"	-	207.13 s
6"	_	195.69 s
$7^{\circ}\alpha, 7^{\circ}\beta$	2.23 d (7.1), 2.21 (7.1)	30.38 t
$8^{"}\alpha, 8^{"}\beta$	1.62 m, 1.60 m	30.14 t
$9^{\circ}\alpha, 9^{\circ}\beta$	1.35 m, 1.33 m	30.38 t
$10^{\circ}\alpha$, $10^{\circ}\beta$	1.28 br s, 1.28 br s	30.14 t
$\Pi^{*} \alpha, \Pi^{*} \beta$	1.28 br s, 1.28 br s	29.92 t
12" α , 12" β	1.28 br s, 1.28 br s	29.92 t
$13^{\circ}\alpha, 13^{\circ}\beta$	1.28 br s, 1.28 br s	29.92 t
$14^{\circ}\alpha, 14^{\circ}\beta$	1.28 br s, 1.28 br s	29.92 t
$15^{\circ}\alpha$, $15^{\circ}\beta$	1.28 br s, 1.28 br s	29.92 t
$16^{\circ}\alpha$, $16^{\circ}\beta$	1.28 br s, 1.28 br s	29.92 t
$17^{"}\alpha, 17^{"}\beta$	1.28 br s, 1.28 br s	29.92 t
$18^{"}\alpha, 18^{"}\beta$	1.28 br s, 1.28 br s	29.92 t
19" α , 19" β	1.28 br s, 1.28 br s	29.92 t
$20^{"}\alpha, 20^{"}\beta$	1.28 br s, 1.28 br s	29.92 t
$21^{"}\alpha, 21^{"}\beta$	1.28 br s, 1.28 br s	29.92 t
$22^{"}\alpha, 22^{"}\beta$	1.28 br s, 1.28 br s	29.92 t
$23^{"}\alpha$, $23^{"}\beta$	1.25 br s, 1.28 br s	29.58 t
24" α , 24" β	1.25 br s, 1.25 br s	29.58 t
$25^{\circ}\alpha, 25^{\circ}\beta$	1.25 br s, 1.25 br s	29.58 t
$26^{\circ}\alpha, 26^{\circ}\beta$	1.12 m, 1.11 m	22.91 t
$27''\alpha, 27''\beta$	1.25 br s, 1.25 br s	18.61 t
28"α, 28"β 20"	1.25 br s, 1.25 br s	18.61 t
74"	U X6 I (/ U)	14 33 0

J values (in Hz) are in parenthesis.

consisted of a filter-sterilized solution of 3% sucrose. halfstrength Gamborg's B5 salts and full-strength Gamborg's vitamins with the pH adjusted to 5.7. The 50-mL cultures were grown in 250-mL Erlenmeyer flasks to late exponential phase in the dark at 26 °C at 100 rpm.

Extraction of hairy roots. The powdered hairy roots of *C. roseus* (200 g) were immersed in methanol (1.5 litre) for three days at room temperature and then the supernatant was concentrated under vacuum to yield 22.5 g of the extract. This material was suspended in water and extracted with ethyl acetate and *n*-butanol successively to produce 11.2 g of ethyl acetate and 7.4 g of *n*-butanol extract.

Isolation of the compounds from ethyl acetate extract. The entire ethyl acetate extract was subjected to normal phase CC over silica gel (400 g) to yield each of 500 mL fractions using the eluent of n-hexane (fraction 1), the mixtures of *n*-hexane:ethyl acetate at 9:1 (fraction 2), 8:2 (fraction 3), 7:3 (fraction 4), 1:1 (fraction 5), and 3:7 (fraction 6) ratio (v/v), ethyl acetate (fraction 7), the mixtures of ethyl acetate:methanol at 9.5:0.5 (fraction 8), 9:1 (fraction 9), 7:3 (fraction 10), 1:1 (fraction 11), and 3:7 (fraction 12) ratio (v/v), and methanol (fraction 13), respectively. All fractions were examined by TLC. Fractions 1 and 2 was not further separated due to the low amount of substance. Fractions 3 (0.8 g) was crystallized after the purification by CC, and then yielded β -sitosterol (20 mg) whose identity was confirmed through the comparison of TLC with an authentic sample from Sigma and spectroscopic data (¹H NMR and 13 C NMR. EI MS). Fractions 4 (0.6 g) was further purified by CC over silica gel (100 g; each fraction of 100 mL) eluting with dichloromethane and chloroform:methanol mixtures (99:1, 98.5:1.5 [fraction ii], 98.2:1.8 [fraction iii], 97.5:2.5 [fraction iv], and 97:3 [fraction v], v/v) to afford 3epibetulinic acid (120 mg). Fraction 6 with hexane:ethyl acetate (3:7, v/v), after re-separation by CC over silica gel with chloroform methanol (99:1 [fraction i], 98:2 [fraction ii], 97:3 [fraction iii], 96:4 [fraction iv], and 95:5 [fraction v], v/v). afforded five fractions. Fraction iv (from the eluent of CHCl₃:MeOH (96:4, v/v)) was re-chromatographed over Lichroprep RP18 ODS (50 g; each fraction of 50 mL). The elution was sequentially performed with methanol containing 80, 60, 40 20, 10, and 0% of water to yield n-pentadecanyl octa-dec-19-en-oate (23 mg). Fraction 9 with EtOAc-MeOH (9:1) after rechromatography over Lichroprep RP18 ODS (50 g; each fraction of 50 mL). The elution was sequentially performed with methanol containing 80, 60, 40 20, 10, and 0% of water to yield two new compounds. (1) (17 mg) and (2) (27 mg), in the methanol fraction containing 60% of water.

Lanast-5,8-dien-3/2-ol-27-oic acid-3/2-D-glucopyranosyl-(4'-1'')-10'',11''-dimethoxy anthracene (1). Dark yellow solid; mp 223-225 °C; $[\alpha]_{D}^{22}$ + 38.1° (*c* 0.21, MeOH); IR (KBr) ν_{max} cm⁻¹: 3577, 3415, 3381, 3180, 2920, 2855, 1695, 1613, 1441, 1350, 1298, 1191, 1107, 744 cm⁻¹; EIMS *m*:*z* 455 [M-C₂₂H₃₃O₇]⁻ (1.8), 438 (1.1), 410 (1.0), 395 (2.8), 383 (1.4), 380 (2.2), 365 (3.5), 355 (5.5), 351 (100), 265 (7.1), 237 (8.0), 222 (7.4), 184 (28.0), 175 (5.0), 143 (7.1), 139 (5.3); positive ion FABMS *m*:*z* 455 [M+H]⁺ C₅₂H₇₁O₁₀ (1.6). Negative ion FABMS *m*:*z* 455 [M-C₂₂H₂₃O₇]⁻; HRFABMS m/2 855.5042 [M+H]⁺ (C₅₂H₇₁H₁₀, calculated for 855.5047); ¹H NMR (MeOD, 500 MHz) and ¹³C NMR (MeOD, 125 MHz) (Table 1).

Acid hydrolysis: Compound 1 (5 mg) was refluxed with 2 mL of 1 M hydrochloric acid:dioxane (1:1, v/v) in a water bath for 4 h. The reaction mixture was evaporated to dryness and partitioned with chloroform and water four times, and each extract was concentrated. The chloroform extract contained the aglycone portion, while the water extract contained D-glucose (co-chromatographed on TLC with authentic sample).

2-Methoxy-6-(*n*-nonacontan-5",6"-dionyl)-11-hydroxy-**13-methyl-11**/*β***-D**-rhamnopyranoside anthracene (2). Dark yellow solid: mp 228-229 °C: $[\alpha]_D^{22} + 27.1^\circ$ (*c* 0.12. MeOH); IR (KBr) ν_{max} cm⁻¹: 3423. 3380, 2924, 2853. 1704, 1617, 1459. 1301, 1290, 1192, 1096. 747, 722; FABMS *m*/*z* 819 [M+H]⁺ (C₅₁H₇₉O₈); EIMS *m*/*z* (rel.int.) 435 [M – C₂₂H₂₃O₆: C₂₉H₅₅O₂]⁺ (5.4). 379 (5.5). 351 (4.6). 322 (5.0). 236 (5.3). 221 (14.9). 220 (4.8), 205 (6.2), 206 (7.6). 195 (10.1). 191 (7.8). 183 (31.5). 155 (39.3), 139 (3.6), 125 (5.6), 111 (7.5), 97 (10.1), 85 (10.0). 83 (9.7). 71 (12.6), 57 (100): ⁻¹H NMR (MeOD. 500 MHz) and ¹³C NMR (MeOD. 125 MHz) (Table 2).

Acid hydrolysis: Compound 2 (5 mg) was refluxed with 2 mL of 1 M hydrochloric acid:dioxane (1:1, v/v) in a water bath for 4 h. The reaction mixture was evaporated to dryness and partitioned with chloroform and water four times. and each extract was concentrated. The chloroform extract contained the aglycone portion, while the water extract contained D-rhamnose (co-chromatographed on TLC with authentic sample).

Acknowledgments. This work was supported by Institute of Biomedical Science and Technology (IBST-2004-1) of Konkuk University.

References

- Taylor, W. L. Farnsworth, N. R. *The Catharanthus Alkaloids*, Marcel Dekker: New York, 1975.
- Chockalingam, S.: Nalina Sundari, M. S.: Thenmozhi, S. J. Environ. Biol. 1989, 10, 303.
- Luijendijk, T. J. C.; Van der, M. E.; Verpoorte, R. J. Chemical Ecology 1996, 22, 1355.
- 4. Leveque, D.: Wihlm, J.: Jehl, F. Bull, Cancer 1996, 83, 176.
- 5. Farnsworth, N. R. Journal of Natural Products 1961, 24, 105.
- 6. Daniel, M.; Sabnis, S. D. Indian J. Exp. Biology 1978, 16, 512.
- Nishibe, S.; Takenaka, T.; Fujikawa, T.; Yasukawa, K.; Takido, M.; Moritsu, Y.; Hirota, A.; Kawamura, K.; Noro, Y. *Natural Medicine* (Tokyo) **1996**, *50*, 378.
- Van der, H. R.; Verpoorte, R.; Ten Hoopen, H. J. G. Plant Cell Tiss. Org. Cult. 1989, 18, 231.
- Moreno, P. R. H.: Van der, H. R.; Verpoorte, R. Plant Cell Tiss. Org. Cult. 1995, 42, 1.
- Moreno, P. R. H.: Poulsen, C.: Heijden, R. V.; Verpoorte, R. Enzyme and Microbial Technology 1996, 18, 99.
- 11. Knobloch, K. H.; Bast, G.; Berlin, J. Phytochemistry 1982, 21, 591.
- Seitz, H. U.; Eilert, U.; De Luca, V.; Kurz, W. G. W. Plant Cell Tiss. Org. Cult. 1989, 18, 71.
- Chung, I. M.; Park, H. Y.; Ali, M.; San, K. Y.; Peebles, C. A. M.; Hong, S. B.; Ahmad, A. Bull. Korean Chem. Soc. 2007, 28, 229.
- 14. Bhadra, R.; Vani, S.; Shanks, J. Biotechnol. Bioeng. 1993, 41, 581.