하해혼성 충적층 유기물 연용 논토양의 화학성 및 양분 집적형태 변화

양창휴 · 유영석 · 류철현 · 정지호 · 김재덕 · 정광용

작물과학원 호남농업연구소, 1전북농업기술원 숙근약초시험장

Changes of Nutrient Accumulation Type and Chemical Property on **Annual Dressing Paddy Soil in Fluvio-marine Deposit**

Chang-Hyu Yang*, Young-Seok Yoo¹, Chul-Hyun Yoo, Ji-Ho Jung, Jae-Duk Kim and Kwang-Yong Jung

Honam Agricultural Rese arch Institute, NICS, RDA, Iksan 570-080, Korea ¹Jinan Medical Herbs Experiment Station, Jeollabuk-do ARES, Jinan 567-806, Korea

A long-term experiment was initiated in 1979 at Experiment Farm of Honam Agricultural Research Institute, to investigate the effects of continued use of organic matter (rice straw and compost) along with different levels of nitrogen fertilizer in rice cultivation. The soils of experimental plots is classified as Jeonbug series developed on Fluvio-marine deposits. The application rates of rice straw and compost were 5 Mg/ha/year and 10 Mg/ha/per year, respectively. The fertilizer N application rates per season were 0, 100, 150, 200, and 250 kg/ha. In 2002, after 24 years of experiment, the effect of different treatments on soil pH, characteristics soil organic matter, soil N and P were investigated. The results of the study is summarized as following. The continue use of organic matter tended to lower the soil pH, to increase organic matter and available phosphate contents. The rice straw tended to lower soil pH more than compost, while the effect of compost was greater in increasing soil organic matter and available soil P then rice straw. The application of organic matter resulted in the increase in total organic N in the soil. Such effect was greater in compost application than in rice straw application. In organic N, greatest was amino acid-N, followed by unidentifiable organic N. The least was amino sugar-N. The application of organic matter with and without nitrogen fertilizer affected the in organic fractions of P, particularly Fe-P and Al-P. The application of rice straw tended to increase Ca-P.

Key words: Nutrient accumulation type, Chemical property, Annual dressing Organic matter, Paddy soil, Fluviomarine deposit

서 언

우리나라 논토양 중 67%인 850천 ha(ASI, 1992)는 작물생산성이 낮은 저위생산 논이다. 생산저해의 주요 인으로는 토양생성 모재의 불량성이 크게 기여하지만 미곡중심의 작부체계, 양질의 유기물(볏짚, 보릿짚 등)과 토양개량제의 지속적인 시용에 의해 지력 증진 을 꾀할 기회가 없었고 과도한 화학비료 편중시비로 철, 망간 등 미량성분의 용탈 등이 중요한 요인으로 생각된다.

벼의 생산은 토양비옥도에 의해 크게 좌우되는데 화 학비료에만 의존하면 지력유지가 매우 어려워 유기물

접 수 : 2006, 8, 18 수 리 : 2006, 11, 8

*연락저자 : Phone: +82638402272, E-mail: yang1907@rda.go.kr 시용에 의한 비옥도 유지 및 증진이 매우 바람직하다. 유기물은 질소, 인산, 가리 등 여러 가지 염류와 미량 요소를 함유하고 있어 양분공급의 기능뿐만 아니라 토양 완충능력을 높여 지력을 좋게 하므로 유기물 함 량이 많을 경우 인산의 용해도가 증가된다(Lee et al., 1995).

토양유기물은 작물의 중요한 질소원으로 토양의 생 산력을 좌우하는 인자로 알려져 있다. 토양유기물은 동식물 및 미생물에서 유래되며 그 중 중요한 것은 탄수화물, 리그닌 및 단백질과 같은 유기질소화합물이 다. 특히 토양 중의 유기태질소는 작물생육 및 토양비 옥도를 결정하는 중요한 요인으로 많은 연구자들이 토양유기물의 질소형태를 분석하였으나, 유기태질소는 토양 중에서 복잡한 화합물로 존재하기 때문에 화학

```
nonacid Aeric Fluventic Haplaquepts)
           가
                                                                                                              0, 100,
                                                 (Ibin,
                                                             150, 200, 250 kg ha<sup>-1</sup>
                                                                          (5 Mg ha<sup>-1</sup>)
1968).
                                                                                                               5 10
                                                             cm
                                                                                               (10 Mg ha<sup>-1</sup>)
                         Chang and Jackson (1957)
AI-P, Fe-P, Ca-P, reductant soluble Fe-P, occluded
AI · Fe-P
                                                             Table 1
               가
                                                                                            Table 2
                                      AI-P, Fe-P, Ca-
         가
                                                             (NIAST, 2000)
                                                                                        рН
                                                             Tyurin ,
                                                                                                                 1N-
                                                                                      Lancaster,
                                                             NaOAc (pH4.0)
                                                                Stewart (Stewart et al., 1963)
                                                                                                          가
                                          (Aramy and
                                                                                                           (
Kemper, 1991).
                                                                  )
                                                                                                         0.5M NH<sub>4</sub>F,
                                                             0.1M NaOH, 0.25M H<sub>2</sub>SO<sub>4</sub>
                                                                                                               AI-P,
                                                             Fe-P, Ca-P
                                              가
                                                                                                     Table 3
                                                                                    рΗ
                                                                                                       가
                                                                                                                рΗ
                                                                                    (Kwun et al., 1984; Muneno et
                                                             al., 1970; Oono et al., 1970; Yamane and Matzura,
                                                             1970)
             1979
                         24
                     (Jeonbug series, fine silty, mixed
Table 1. Major chemical composition of organic matter used in the study.
```

Organic matter	T-C	T-N	P ₂ O ₅	K ₂ O	C/N ratio
			%		
Rice straw	40.3	0.6	0.24	2.32	67.2
Compost	30.5	1.5	2.50	1.30	20.3

Table 2. Chemical properties of soil in experimental plot.

»II	OM	OM Av. P ₂ O ₅	Av. SiO ₂	Exch. cations			CEC
pН	OM	AV. P2O5	Av. 51O2	K	Ca	Mg	
1:5	g kg ⁻¹	mg	kg ⁻¹		cmo	k kg ⁻¹	
6.4	23	100	109	0.12	4.0	2.0	12.0

.

pН

Table 3. Soil pH observed in different years under different treatments.

Organia mattar	pH						
Organic matter -	1984	1990		2000	2002		
			1:5				
Non application	5.60±0.06	5.86±0.10		6.72±0.19	6.96±0.14		
Rice straw	5.60±0.06	5.84±0.15		6.48±0.40	6.66±0.05		
Compost	5.68±0.04	5.96±0.10		6.60±0.25	6.78±0.15		
					(Yoon, 1983)		
Table 4							
		가					
,	가						
1		750 kg ha ⁻¹			Table 6		
	10%		>	>	가 .		
	7.5 M	g ha ⁻¹ yr ⁻¹					
7.5 Mg	ha ⁻¹				가		
가	0.43 0.45 g kg ⁻¹ y	r ⁻¹	>	(-N)> 가		
(Cho et al., 1992)		10 Mg ha ⁻¹			가		
yr ⁻¹		-		가		20	
, ,			Mg ha⁻¹	14	Yoo et al.(19	992)	
가			· ·		가 가	·	
			>	(-N) > 가		
				•	(Ichida, 1986)).	
			가		•	,	
			918 mg kg	r ⁻¹	52.7%		
Table 5 .		70 90 kg		,			
ha ⁻¹		89 mg kg ⁻¹		가	(1	chida,	
		3 3	1986).	·	,	,	
가			,		Amino acid-N		
•	가(Kwak et	al. 1990)	- N	가	Amino sugar-	N	
	가	,	Humin-N		52941	-	
	•		***************************************				

Table 4. Soil organic matter content observed in different years under different treatments.

0		0	M	
Organic matter —	1984	1990	2000	2002
		g l	kg ⁻¹	
Non application	18.3±1.34	19.2±0.78	19.8±0.92	20.0±0.63
Rice straw	22.7±0.42	23.0±1.42	23.0±2.92	26.0±0.30
Compost	23.9±0.41	25.0±0.14	25.6±2.09	26.2±0.66

Table 5. Available soil P content observed in different years under different treatments.

Organic matter —		Av.	P2O5		
Organic matter —	1984	1990	2000	2002	
	mg kg ⁻¹				
Non application	61±13.02	75±0.89	85±8.01	89±12.88	
Rice straw	94±17.05	96±25.27	99±13.17	105±0.89	
Compost	94±6.72	102±1.79	112±28.68	126±29.31	

Table 6. Effect of organic matter application on different fractions of organic N in soil.

Organic matter	Nitrogen level		Acid soluble nitrogen		Acid insoluble nitrogen
Organic matter	Nitrogen iever	Amino sugar	r-N	Amino acid-N	Acid hisolable hidogen
	kg ha ⁻¹			mg kg ⁻¹	
37 11 2	0	277		697	399
Non application	150	283		785	406
Rice straw	0	294		778	411
Nice suaw	150	238		918	449
Compost	0	308		869	445
	150	312		835	470
(Lee and	I Hwang, 1984).				(JSSP, 1984).
,	가 가	가	17		Fe-P
60 70%가	. 1	. ,	 가		
		(Ikeda,	Fe-P	가 AI-	P Ca-P
1997).	가	,	가	(Singh et al., 19	
,	,	7%		,	,
가		4		occluded-P	
•	(25 45%),	(12		가	(AI-P, Fe-P,
50%),	(12 25%),	(1 7%)	Ca-P)	·	(Robertson et al.,
	Lim and Moon, 1983).	(, , ,	1966).		(1.020.100.101.01.01.
(.	,,,.	Table 7	.000).		
> :	> 가	Fe-			
P > AI-P > Ca-F		-			
가		·		가	
가		Park et		가	
		가		가	
al.(2003)	•	71	(Taylor a		
	가		(Taylor a	nd Jonathan, 1981).	
	71				
		가			
			,		
, pH, Eh,			가		
Fe-P, AI-P가	(Iman	o and Hirohou,			

Acid soluble nitrogen

Table 7. Effect of organic matter application on different fractions of inorganic P in soil.

가

1970).

Organic matter	Nitrogen level	Al-P	Fe-P	Ca-P	Occluded-P	
	kg ha ⁻¹	mg kg ⁻¹				
Non application	0	181	291	99	49	
	150	250	309	86	50	
Rice straw	0	207	425	116	48	
	150	186	298	96	51	
Compost	0	198	366	113	45	
	150	228	375	110	48	

가 가 가 가 2 3 가 가 가 24 (0, 100, 150, 200, 250 kg ha⁻¹) 5 가 가 가 (NPK) 가 가 > > 가 () >) > (가 가 918 mg kg⁻¹ 가 52.7% 가 > Fe-P 425 mg kg⁻¹ 가 + Fe-P AI-P . Ca-P

ASI. 1992. General remarks of Korean soils. p. 308-610.

Agricultural Sciences Institute, Suwon, Korea.

Aramy, T.J., and W.D. Kemper. 1991. Support for long term agriculture research. J. Agronomy 83:62-65.

Chang, S.C., and Jackson. 1957. Fractionation of soil phosphorus. Soil Sci. 84:133-144.

Cho, S.J., C.S. Park, and D I. Uhm. 2003. Soil Science. p. 48-49.

Ibin, S. 1968. On nitrogen type of volcanic ash soils and nonvolcanic ash soils. Soil Sci. Plant Nutr. 39:478.

Ichida, A.H. 1986. Soil properties and yield potential of long-term compost application in paddy field. Soil Sci. Plant Nutr. 57:418-420.

Ikeda, A.H. 1997. Rational fertilization management of rice plant by nutrition and soil diagnosis. Agriculture & Hoticulture. 72:999.

Imano, R.H., and T.M. Hirohou. 1970. The changes of substance on percolation state in paddy soil. Soil Sci. Plant Nutr. 41:225-229.

JSSP. 1984. Phosphoric acid and paddy soil . Fertilizer efficiency and availability on phosphoric acid of paddy soil in temperate area.
p. 105-108. Japanese Society of Soil Science and Plant Nutrition, Tokyo, Japan.

Kwak, H.K., C.S. Lee, and S.K. Lim. 1990. Influence of soil amendments on phosphorus response and changes of available phosphate amount in paddy soil. Res. Rept. RDA. 32:52-56.

Kwun, K.C., J.K. Kim, I.Y. Kim, S.H. Park, and D.S. Jo. 1984. Effects of continuous fertilizations on physical and chemical properties of paddy soil and growth of rice plant. Res. Rept. RDA. 26:67-76.

Lee, K.B., J.G. Kang, T.Y. Uhm, J.G. Kim, S.K. Kim, and G.S. Lee. 1995. The effect of long-term organic matter application on N, P and K uptake by rice in paddy soil. RDA. J. Agri. Sci. 37:291-297.

Lee, S. K. and G. N. Hwang. 1984. Effects of compost and rice straw on immobilization and mineralization of nitrogen fertilizer added to coarse loamy and clay soil. Korean J. Soil Sci. Fert. 17:60-66.

Muneno, O.T., K. Naohara, S.H. Kusaka, and T.M. Imai. 1970. Change of soil physico-chemical properties by application barley and rice straw. Res. Rept. Chugoku National Agricultural Experiment Station. 5:62-64.

NIAST. 2000. Methods of soil and crop plant analysis. National Institute of Agricultural Science and Technology. Suwon.

Oono, T.R., H.O. Nishio, and Y.K. Ishigaki. 1970. Change of soil physico- chemical properties by application barley and rice straw. Res. Rept. Chugoku National Agricultural Experiment Station. 5:67-68.

Park, C.Y., W.T. Jeon, K.D. Park, Y.S. Cho, and U.G. Kang. 2003. Effects of long-term application of the same fertilizers to rice crop on a river alluvial paddy soil. 50th ceremony Symposium on annual dressing experiment of fertilizer. p. 97-144.

Robertson, W.K., L.G. Thompson, and C.E. Hutton. 1966. Availability and fractionation of residual phosphorus in soils high in aluminium and iron. Soil Sci. Soc. Amer. Proc. 30:446-450.

Singh, R.N., D.C. Martens, and S.S. Obenshain. 1966. Plant availability and form of residual phosphorus in Davidson clay loam. Soil Sci. Soc. Amer. Proc. 30:617-620.

Stewart, B.A., L.K. Porter, and D.D. Johnson. 1963. Immobilization

and mineralization of nitrogen in several organic fraction of soils. Soil Sci. Am. Proc. 27:302-304.

Taylor, R.W, and J. Woods. 1981. Inorganic phosphorus in calcareous rock land soils of the Bahamas. Soil Sci. Soc. Amer. Proc. 45:730-734.

Yoo, C.H., J.D. So, A. Ida, F. Tanaka, and M. Nishida. 1992. Effects

of long-term organic matter application on the fine textured paddy soils of double cropping system in temperate area. Korean J. Soil Sci. Fert. 25:325-333.

Yoon, J.H. 1983. Parameters of soil phosphorus availability factors in predicting yield response and fertilizer recommendation(Ph. D. Thesis). Dong-Guk Univ. Korea.