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Convergent Synthesis of Triazole Dendrimers via Click Chemistry Using Tripodal Core

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Azides and alkynes are among the least reactive functional groups and normally require prolonged heating to form the cycloaddition product. Recent investigation of the click chemistry, the Cu(I)-catalyzed Huisgen [2+3] dipolar cycloaddition reaction between an organic azide and an alkyne, has attracted attention.¹ The reaction, characterized by very high yields, mild and simple reaction conditions, oxygen and water tolerance, and simple ways to product isolations, is highly chemoselective affording only the desired 1,2,3-triazole even in the presence of a large variety of other functional groups. The methodology has found many applications in organic chemistry,² supramolecular chemistry,³ drug discovery,⁴ bioconjugations,⁵ and materials science.⁶

Dendrimers are highly branched macromolecules, which are prepared by repetition of a given set of reactions using either divergent or convergent strategies.⁷ Generally, the reactions employed are high yielding without any side reactions. This then allows one to obtain defined and uniform structures. Well known processes, such as the Michael reaction, Williamson ether synthesis, amidations and reductions have been used extensively in pioneering work by Vögtle, Tomalia, Fréchet and Newkome.8 There are few reports to synthesize the triazole-mediated dendritic materials using click chemistry with dendronized azide.9 The most attractive feature of the convergent synthesis is that it requires few transformations per molecule in the coupling step. The reaction can be driven to completion with equivalent or only a slight excess of reagent, providing greater structural control in preparing well-defined (un)symmetrical dendrimers. Due to our interest in developing new dendrimers, we became involved in exploring efficient click reaction that provides an easy access to dendrimers. Here we present the convergent synthesis of triazole dendrimers by trimerization of acetylenic dendrons via click chemistry using tripodal core, 1,3,5-triazidomethylbenzene.

Experimental Section

General procedure for the preparation of acetylenicdendrons 2-Dn. N,N'-Dicyclohexylcarbodiimide (0.3 mmol) was added to a solution of propiolic acid (0.3 mmol) and the corresponding dendritic benzyl alcohols (0.3 mmol) in CH₂Cl₂ (5 mL) at 5 °C. The reaction mixture was stirred for 2 h at the same temperature and a further 12 h at room temperature. The resulting suspension was filtered and the filtrate was concentrated and purified by column chromatography (EtOAc/Hex) to afford the pure product **2-Dn**.

2-D1: 90% yield; ¹H NMR (300 MHz, CDCl₃) δ 2.93 (s, 1 H), 3.79 (s, 6 H), 5.14 (s, 2 H), 6.43 (d, *J* = 2.3 Hz, 1 H), 6.51 (d, *J* = 2.3 Hz, 2 H); ¹³C NMR (75 MHz, CDCl₃) δ 161.0, 152.5, 136.7, 106.3, 100.7, 75.4, 74.6, 67.8, 55.4; MS (EI): *m*/*z* 220 [M⁺], 167, 151, 139; HRMS (EI) calcd for C₁₂H₁₂O₄: 220.0736. found: 220.0732.

2-D2: 75% yield; ¹H NMR (300 MHz, CDCl₃) δ 2.92 (s, 1 H), 3.78 (s, 12 H), 4.95 (s, 4 H), 5.12 (s, 2 H), 6.40 (m, 2 H), 6.56 (m, 5 H), 6.59 (m, 2 H); ¹³C NMR (75 MHz, CDCl₃) δ 161.1, 160.1, 152.5, 139.1, 136.8, 107.4, 105.3, 102.3, 100.0, 75.4, 74.6, 70.1, 67.7, 55.4; MS (EI): m/z 492 [M⁺], 413, 219, 195, 151, 113; HRMS (EI) calcd for C₂₈H₂₈O₈: 492.1784. found: 492.1786.

2-D3: 60% yield; ¹H NMR (500 MHz, CDCl₃) δ 2.89 (s, 1 H), 3.78 (s, 24 H), 4.97 (s, 12 H), 5.13 (s, 2 H), 6.40 (m, 4 H), 6.56 (m, 13 H), 6.66 (m, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 161.1, 160.2, 160.1, 152.6, 139.3, 139.2, 136.9, 107.5, 106.5, 105.4, 102.3, 101.8, 100.1, 75.4, 74.6, 70.2, 70.1, 67.8, 55.3; MS (FAB): *m*/*z* 1035.8 [M⁺], 663.1, 647.1 422.9, 391.0; HRMS (FAB) calcd for C₆₀H₆₀O₁₆: 1036.3881. found: 1059.3778 [M⁺ + Na].

General procedure for the preparation of 1,2,3-triazole dendrimers 3-Gn by reaction between tripodal azide 1 and acetylenic-dendrons 2-Dn. A mixture of 1,3,5triazidomethylbenzene¹⁰ 1 (0.05 mmol) and acetylenicdendrons 2-Dn (0.15 mmol) in toluene in the presence of CuI was stirred at 50-60 °C. The reaction was monitored by TLC regarding on the disappearance of 2-Dn. The mixture was cooled to room temperature and then normal workup was carried out. The desired product was isolated by column chromatography.

3-G1: ¹H NMR (500 MHz, CDCl₃) δ 3.78 (s, 18 H), 5.31 (s, 6 H), 5.53 (s, 6 H), 6.42 (s, 3 H), 6.58 (s, 6 H), 7.16 (s, 3 H), 8.03 (s, 3 H); ¹³C NMR (75 MHz, CDCl₃) δ 160.9, 160.2, 140.0, 137.6, 136.4, 128.3, 128.2, 106.3, 100.1, 66.7, 55.3, 53.3; MS (FAB): *m*/*z* 903.8 [M⁺]; HRMS (FAB) calcd for C₄₅H₄₅N₉O₁₂: 903.3188. found: 903.3184.

3-G2: ¹H NMR (300 MHz, CDCl₃) δ 3.75 (s, 36 H), 4.91 (s, 12 H), 5.24 (s, 6 H), 5.46 (s, 6 H), 6.37 (m, 6 H), 6.53 (m, 15 H), 6.61 (m, 6 H), 7.16 (s, 3 H), 8.02 (s, 3 H); ¹³C NMR

(75 MHz, CDCl₃) δ 161.1, 160.2, 160.1, 140.4, 139.1, 137.8, 136.6, 128.11, 128.05, 107.5, 105.4, 102.0, 100.0, 70.1, 66.8, 55.5, 53.5; MS (FAB): m/z 1720.8 [M⁺], 1743.1 [M⁺ + Na]; HRMS (FAB) calcd for C₉₃H₉₃N₉O₂₄: 1719.6334. found: 1719.6314.

3-G3: ¹H NMR (300 MHz, CDCl₃) δ 3.75 (s, 72 H), 4.92 (s, 36 H), 5.25 (s, 6 H), 5.35 (s, 6 H), 6.37 (m, 18 H), 6.49 (m, 6 H), 6.54 (m, 27 H), 6.62 (m, 12 H), 7.06 (s, 3 H), 7.90 (s, 3 H); ¹³C NMR (125 MHz, CDCl₃) δ 161.0, 160.1, 160.0, 140.3, 139.2, 139.1, 137.7, 136.4, 128.0, 127.8, 107.3, 106.4, 106.3, 105.3, 102.0, 101.6, 99.9, 70.0, 66.6, 55.5, 53.4; MS (FAB): *m*/*z* 3354.6 [M⁺]; HRMS (FAB) calcd for C₁₈₉H₁₈₉N₉O₄₈: 3352.2625. found: 3352.2604.

Results and Discussion

The synthetic strategy for triazole dendrimers **3-Gn** utilized a convergent method using tripodal azides **1** and Fréchet-type acetylenic-dendrons **2-Dn** (Scheme 1). The Fréchet-type acetylenic-dendrons **2-Dn** (n = 1, 2, and 3: generation of dendron) were synthesized by DCC coupling

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of propiolic acid and the corresponding dendritic benzyl alcohols (Figure 1).

To test the effectiveness of the dipolar cycloaddition reactions of tripodal azide 1 and acetylenic-dendrons 2-Dn, we have reacted 1 with 2-D1 in toluene. The cycloadduct 3-G1, which can be seen as first generation dendrimer was obtained in good yield contaminated with dendritic isomer having 1,5-disubstituted 1,2,3-triazole units. The reaction could be checked by IR spectroscopy by following the disappearance of the N₃ stretching band at around 2100 cm⁻¹. The generation and disappearance of the intermediates, which are mono- and/or di-triazole derivatives, were monitored by TLC runs of the reaction mixture. Both the 1,4- and 1,5-isomers of 1,2,3-triazole moiety, known to be generated from the azide and acetylene reaction,¹¹ were also formed in the reactions of 2-D1 with benzyl azide (Scheme 2). At 80 °C in toluene the reaction provided the 1,2,3-triazole product in 93% yield with the 1,4- to 1,5isomers in the ratio of $\sim 3 : 1.^{12}$ To reduce the generation of 1,5-disubstituted 1,2,3-triazole units in the dendrimers, we adapted a literature procedure, where the role of Cu(I)



Scheme 1. Synthetic strategy of triazole dendrimers by trimerization. Reaction condition: toluene, CuI, 50 °C.



Figure 1. Structures of acetylenic-dendrons 2-Dn.

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Scheme 2. Cycloaddition reaction of benzyl azide with acetylenicdendron 2-D1. Reaction condition: toluene, 80 °C, 30 h or toluene, CuI, 50 °C, 48 h.

species in Huisgen [2 + 3] dipolar cycloaddition reaction has been proved to influence the regioselective product distribution.¹³ The reaction of **2-D1** with benzyl azide in the presence of CuI at 50 °C in toluene provided the 1,2,3triazole product in 95% yield with ~8 : 1 ratio of 1,4- to 1,5isomer.

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To synthesize the triazole dendrimers, we investigated the reactions of tripodal azide 1 with acetylenic-dendrons 2-D1 in the presence of CuI at various temperature in solvents such as toluene, THF, DMF, and THF-toluene mixture. The reaction conducted at 50 °C in toluene provided the most satisfying result and led to the first generation triazole dendrimer. With this optimized reaction condition, we synthesized the triazole dendrimers by the trimerization reactions between tripodal azide 1 and acetylenic-dendrons 2-Dn. The reaction of 1 with 2-D1 in the presence of CuI at 50 °C in 0.05 M solution in toluene provided the triazole dendrimer **3-G1** having just 1,4-disubstituted 1,2,3-triazole units in 59% yield. Other isomeric products of the reaction are the triazole dendrimers containing two 1,4-disubstituted 1,2,3-triazole units and one 1,5-disubstituted 1,2,3-triazole unit (~18%), and a 1,4-disubstituted 1,2,3-triazole unit and two 1.5-disubstituted 1.2.3-triazole units (\sim 3%). The same reaction conducted in 0.015 M concentration afforded the desired product 3-G1 in 70% yield and reduced the generation of isomeric triazole dendrimers.14

Next, we conducted the reaction in diluted solution to suppress the generation 1,5-isomer unit(s) and remove the solubility problem of dendritic intermediate(s) in the construction of higher generation dendrimers. The reaction of 1 with 2-D2 in the presence of CuI at 50 °C in 0.008 M solution in toluene provided the triazole dendrimer 3-G2 in 79% yield having only 1,4-disubstituted 1,2,3-triazole units and tiny amount of the triazole dendrimers containing two



Figure 2. Structure of the third generation of triazole dendrimer 3-G3.

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1,4-disubstituted 1,2,3-triazole units and one 1,5-disubstituted 1,2,3-triazole unit. These observations suggested that the Cu(I) species, steric effect of the dendron, and kinetic factors controlled by concentration are crucial in the formation of the desired 1,4-disubstituted 1,2,3-triazole dendrimer. The reaction of 1 with the third generation dendron 2-D3 in the presence of CuI at 50 °C in 0.002 M toluene solution provided the third generation triazole dendrimer **3-G3** in 66% yield having only 1,4-disubstituted 1,2,3-triazole units without any isomeric dendritic components. This observation indicates that the steric hindrance of dendron is important factor in ruling out the generation of 1,5-isomeric unit in trimerization reaction between tripodal azide and dendrons. Thus, the formation of triazole between tripodal azide and acetylenic dendrons can be regarded as a new approach to construct various dendrimers.

All compounds were confirmed by ¹H and ¹³C NMR spectroscopy and exhibited very good correlation with the calculated molecular masses as evidenced by their FAB mass spectra. From their ¹H NMR spectra (CDCl₃), the peaks of the benzene protons of core and the triazole protons in dendrimers **3-Gn** were found at 7.16 and 8.03 ppm for **3-G1**, 7.16 and 8.02 ppm for **3-G2**, and 7.06 and 7.90 ppm for **3-G3**, respectively. As the dendrimer generation increased, the peaks of the benzene protons of core and the triazole protons showed up-field shift. In third generation dendrimer (Figure 2) it is observed that the benzene protons of core and the triazole protons are influenced by the larger dendritic effect changing their microenvironment.

In summary, we have demonstrated that the trimerization reactions between tripodal azide and acetylenic dendrons lead to the formation of 1,4-disubstituted triazole dendrimers in high yields. Furthermore, it is found that the presence of Cu(I) species, steric effect of the dendron, and kinetic factors controlled by concentration are crucial in higher yields of the 1,4-disubstituted 1,2,3-triazole dendrimer. This unique reaction may provide an insight into designing various symmetrical dendrimers. We are currently working towards various functional dendrimers using this strategy for various applications, such as the templated synthesis of nanomaterials.

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References and Notes

- Kolb, H. C.; Finn, M. G.; Sharpless, K. B. Angew. Chem. Int. Ed. 2001, 40, 2004.
- (a) Katritzky, A. R.; Zhang, Y.; Singh, S. K. *Heterocycles* 2003, 60, 1225. (b) Wang, Z.-X.; Qin, H.-L. *Chem. Commun.* 2003, 2450. (c) Jin, T.; Kamijo, S.; Yamamoto, Y. *Eur. J. Org. Chem.*

2004, 3789.

- (a) Badjiæ, J. D.; Balzani, V.; Credi, A.; Lowe, J. N.; Silvi, S.; Stoddart, J. F. *Chem. Eur. J.* **2004**, *10*, 1926. (b) Lee, J. K.; Chi, Y. S.; Choi, I. S. *Langmuir* **2004**, *20*, 3844. (c) Collman, J. P.; Devaraj, N. K.; Chidsey, C. E. D. *Langmuir* **2004**, *20*, 1051. (d) Chi, Y. S.; Lee, J. K.; Lee, K.-B.; Kim, D. J.; Choi, I. S. *Bull. Korean Chem. Soc.* **2005**, *26*, 361.
- (a) Lenda, F.; Guenoun, F.; Tazi, B.; Larbi, N. B.; Martinez, J.; Lamaty, F. *Tetrahedron Lett.* **2004**, *45*, 8905. (b) Manetsch, R.; Krasiñski, A.; Radiæ, Z.; Raushel, J.; Taylor, P.; Sharpless, K. B.; Kolb, H. C. *J. Am. Chem. Soc.* **2004**, *126*, 12809. (c) Kolb, H. C.; Sharpless, K. B. *Drug Discovery Today* **2003**, *8*, 1128.
- (a) Wang, Q.; Chan, T. R.; Hilgraf, R.; Fokin, V. V.; Sharpless, K. B.; Finn, M. G. J. Am. Chem. Soc. 2003, 125, 3192. (b) Meng, J.-C.; Siuzdak, G.; Finn, M. G. Chem. Commun. 2004, 2108. (c) Link, A. J.; Vink, M. K. S.; Tirrell, D. A. J. Am. Chem. Soc. 2004, 126, 10598. (d) Sivakumar, K.; Xie, F.; Cash, B. M.; Long, S.; Barnhill, H. N.; Wang, Q. Org. Lett. 2004, 6, 4603 and references therein.
- Díaz, D. D.; Punna, S.; Holzer, P.; McPherson, A. K.; Sharpless, K. B.; Fokin, V. V.; Finn, M. G. J. Polym. Sci., Part A 2004, 42, 4392.
- 7. Grayson, S. M.; Fréchet, J. M. J. Chem. Rev. 2001, 101, 3919.
- (a) Newkome, G. R.; Moorefield, C. N.; Vögtle, F. *Dendrimers* and Dendrons: Concepts, Synthesis, Applications; Wiley-VCH: Weinheim, 2001. (b) Choi, J. S.; Choi, Y. H.; Park, J. S. Bull. Korean Chem. Soc. 2004, 25, 1025. (c) Kim, T.-i.; Jang, H.-S.; Joo, D. K.; Choi, J. S.; Park, J.-S. Bull. Korean Chem. Soc. 2003, 24, 123.
- (a) Wu, P.; Feldman, A. K.; Nugent, A. K.; Hawker, C. J.; Scheel, A.; Voit, B.; Pyun, J.; Fréchet, J. M. J.; Sharpless, K. B.; Fokin, V. V. *Angew. Chem. Int. Ed.* **2004**, *43*, 3928. (b) Helms, B.; Mynar, J. L.; Hawker, C. J.; Fréchet, J. M. J. *J. Am. Chem. Soc.* **2004**, *126*, 15020. (c) Wuytswinkel, G. V.; Verheyde, B.; Compernolle, F.; Toppet, S.; Dehaen, W. *J. Chem. Soc., Perkin Trans. 1* **2000**, 1337. (d) Lee, J. W.; Kim, B. K. *Bull. Korean Chem. Soc.* **2005**, *26*, 658.
- 10. Heyer, H.; Lehn, J. M. Tetrahedron Lett. 1986, 27, 5869.
- Huisgen, R. In 1,3-Dipolar Cycloaddition Chemistry; Padwa, A., Ed.; Wiley: New York, 1984; Vol. 1, pp 1-176.
- 12. Each isomer can be separated by column chromatography and distinguished by their ¹H NMR spectra (CDCl₃) that the triazole proton of 1,4-regioisomer was found at 7.97 ppm and in the corresponding 1,5-isomer it was shifted downfield to 8.17 ppm. For 1,4-regioisomer: ¹H NMR (200 MHz, CDCl₃) δ 3.77 (s, 6 H), 5.29 (s, 2 H), 5.56 (s, 2 H), 6.41 (m, 1 H), 6.57 (m, 2 H), 7.27 (m, 2H), 7.38 (m, 3 H), 7.97 (s, 1 H); ¹³C NMR (50 MHz, CDCl₃) δ 160.9, 158.1, 138.5, 138.3, 136.7, 134.8, 128.7, 128.4, 128.0, 106.2, 100.3, 67.2, 55.3, 53.5. For 1,5-regioisomer: ¹H NMR (200 MHz, CDCl₃) δ 3.78 (s, 6 H), 5.23 (s, 2 H), 5.92 (s, 2 H), 6.45 (m, 1 H), 6.50 (m, 2 H), 7.30 (m, 5H), 8.17 (s, 1 H); ¹³C NMR (50 MHz, CDCl₃) δ 160.8, 160.3, 140.0, 137.5, 133.6, 129.1, 128.2, 127.5, 127.3, 106.2, 100.2, 66.6, 55.3, 54.3.
- (a) Rostovtsev, V. V.; Green, L. G.; Fokin, V. V.; Sharpless, K. B. Angew. Chem. Int. Ed. 2002, 41, 2596. (b) Tornøe, C. W.; Christensen, C.; Meldal, M. J. Org. Chem. 2002, 67, 3057. (c) Fazio, F.; Bryan, M. C.; Blixt, O.; Paulson, J. C.; Wong, C.-H. J. Am. Chem. Soc. 2002, 124, 14397.
- 14. In addition to the desired product, the triazole dendrimer containing two 1,4-disubstituted 1,2,3-triazole units and one 1,5disubstituted 1,2,3-triazole unit was obtained in ~9% yield wheares the triazole dendrimer having a 1,4-disubstituted 1,2,3triazole unit and two 1,5-disubstituted 1,2,3-triazole units was not found.