

A New 24-Nor-Lupane-Glycoside of Acanthopanax trifoliatus

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A new 24-nor-lupaneglycoside was isolated from the leaves of *Acanthopanax trifoliatus*. Based on spectroscopic data its chemical structure was determined as 24-nor-11 α -hydroxy-3-oxo-lup-20(29)-en-28-oic acid 28-O- α -L-rhamnopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosyl ester.

Key words: Acanthopanax trifoliatus, Araliaceae, 24-Nor-lupane-glycoside, Acantrifoside B

INTRODUCTION

Acanthopanax trifoliatus (L.) Merr., (Araliaceae) is a medicinal plant that mainly occurs in Asian subtropical countries, including China, Japan, Korea, Laos and Vietnam. In Vietnam, the species is concentrated in the border mountainous regions of the North, in Caobang, Langson, Laocai and Laichau provinces, where it is used in folk medicine (Chi, 1997; Loi, 2001) as a drug with ginsenglike activity. In China, decoctions of the leaves and young shoots are prescribed for the treatment of tuberculosis and lung hemorrhage, and as a tonic to improve general weakness (Perry, 1980). Previously, a number of lupanetriterpene compounds have been isolated from the leaves of this plant (Ty et al., 1984 and 1985; Lischewsky et al., 1985; Yook et al., 1998). During the course of our continuing work on Acanthopanax species, we isolated and determined the structure of a new 24-nor-lupan-triterpene glycoside from the leaves of A. trifoliatus. Based on spectroscopic data, its chemical structure was determined to be 24-nor-11α-hydroxy-3-oxo-lup-20(29)-en-28-oic acid 28-O- α -L-rhamnopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-(1 \rightarrow 6)- β -Dglucopyranosyl ester, which we named acantrifoside B (1).

MATERIALS AND METHODS

General experimental procedures

M.p.'s were determined using a Kofler micro-hotstage;

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IR spectra were obtained on a Hitachi 270-30 type spectrometer from KBr discs. Optical rotations were determined on a JASCO DIP-1000 KUY polarimeter. FAB-MS and HR FAB-MS were obtained using a JEOL JMS-DX 300 spectrometer. $^1\text{H-NMR}$ (300 MHz) and $^1\text{C-NMR}$ (75 MHz) were recorded on a Bruker DRX300 spectrometer and chemical shifts are referenced to δ using TMS as an internal standard. Column chromatography (CC) was performed on silica gel 60, YMC RP-18 resin or Dianion HP-20 resin.

Plant material

The leaves of *A. trifoliatus* were collected in Langson province, Vietnam in January 2001 and identified by Prof. Dr. Tran Minh Hoi, Institute of Ecology, Biological Resources, NCST of Vietnam. Two voucher specimens (No 2539) are deposited at the herbarium of the Institute of Natural Products Chemistry, NCST, Vietnam, and at the herbarium of the College of Pharmacy, Chungnam National University, Korea.

Extraction and isolation

Dried and powdered leaves (3.7 kg) were extracted three times with hot MeOH. The combined solutions were evaporated under reduced pressure to yield a MeOH extract (250 g), which was suspended in water and partitioned with dichloromethane. The water fraction was adsorbed on highly porous polymer resin (Dianion HP-20, Mitsubishi Chem. Ind. Co. Ltd, Tokyo, Japan) and eluted with water containing increasing concentrations of MeOH (100% H₂O, 20% MeOH, 40% MeOH, 60% MeOH, 80% MeOH, and 100% MeOH). The 20% MeOH and 40%

Table I ¹H and ¹³C-NMR spectral data of compound 1

Pcs	$\delta_{c}^{a,b}$	δ _H ^{a,c}
1	43.9 (t)	1.36 (m)
2	38.4 (t)	2.09 (dd, 8.7, 6.9 Hz)
3	217.4 (s)	_
4	46.2 (d)	2.46 (m)
5	54.5 (s)	1.61 (m)
6	23.2 (t)	1.46 (m)/ 1.60 (m)
7	35.0 (t)	1.38 (m)/ 1.46 (m)
8	43.5 (s)	_
9	55.8 (d)	1.20 (d, 7.7 Hz)
10	39.4 (s)	
11	70.8 (d)	4.0 (dt, 10.7, 5.5 Hz)
12	37.6 (t)	2.07 (m)
13	38.4 (d)	2.25 (m)
14	43.5 (s)	
15	32.9 (t)	1.19 (m)/ 1.55 (m)
16	32.9 (t)	2.44 (m)
17	58.0 (s)	
18	50.2 (d)	1.82 (m)
19		1.82 (m)
20	48.3 (d)	3.09 (m)
	151.4 (s)	_ 2.07 (m)
21	31.3 (t)	2.07 (m)
22	37.6 (t)	2.07 (m)
23	12.7 (q)	1.04 (d, 6.5 Hz)
24	44.4.(*)	1 30 (a)
25	14.4 (q)	1.30 (s)
26	17.7 (q)	1.10 (s)
27	14.9 (q)	1.11 (s)
28	176.4 (q)	- A FO (411 bo -)/ A 70 (411 bo -)
29	111.0 (q)	4.59 (1H, br s)/ 4.73 (1H, br s)
30	19.8 (q)	1.77 (s)
C-28 O-r	ıner glc	
1	95.5 (d)	5.55 (d, 6.0 Hz)
2	73.9 (d)	3.92 (m)
3	78.4 (d)	3.66 (m)
4	71.1 (d)	4.19 (m*)
5	78.2 (d)	3,54 (m*)
6	69.6 (t)	4.21 (1H, d, 10.2 Hz), 3.92 (1H, m)
Glc'(1→6	3)glc	
1	104.6 (d)	4.47 (d, 7.8 Hz)
2	75.4 (d)	3.35 (t, 8.5 Hz)
3	76.8 (d)	3.56 (m*)
4	79.7 (d)	3.63 (m*)
5	77.0 (d)	3.58 (m*)
6	63.0 (t)	3.75 (1H, m), 3.90 (1H, m)
Rha 1-		
1	103.1 (d)	4.86 (d, 4.1 Hz)
2	72.5 (d)	3.85 (m)
3		3.71 (dd, 9.3, 3.2 Hz)
ა 4	72.3 (d)	, , ,
4 5	74.1 (d)	3.56 (m) 3.92 (m)
6	71.2 (d) 18.0 (q)	1.36 (d, 3.8 Hz)
	10.0 (4)	1.30 (u, 3.0 mz)

MeOH fractions were combined and chromatographed on a silica gel column using CHCl₃-MeOH-H₂O (70:30:4) as eluant. This was followed by CC on a YMC RP-18 column using a MeOH-H₂O (7:3) eluant to yield $\bf{1}$ (1.9 g).

24-Nor-11α-hydroxy-3-oxo-lup-20(29)-en-28-oic acid 28-O-α-L-rhamnopyranosyl-(1 \rightarrow 4)-β-D- glucopyranosyl-(1 \rightarrow 6)-β-D-glucopyranosyl ester (1)

A white powder, m.p. 218-220°C, $[\alpha]_0^{25}$ 20.0° (c 2.00, MeOH); IR ^{KBr}v_{max} cm⁻¹: 3,414 (br, OH), 2,985 (C-H), 1,736 (>C=O), 1,693(-COO), 1,645 (>C=CH₂), 1,060 (C-O-C); FAB-MS (m/z): 925 [M-H]⁻, HR FAB-MS (m/z): 925.4771 [M-H]⁻ (Calcd. 925.4800 for C₄₇H₇₃O₁₈); ¹H- and ¹³C-NMR: see Table I.

RESULTS AND DISCUSSION

Compound 1 was obtained as white powder from water fraction of the leaves of the MeOH extract of A. trifoliatus. and showed hydroxyl absorbance bands at 3,414 cm⁻¹, carbonyl absorbance at 1,736 cm⁻¹, and carboxyl at 1,693 cm⁻¹ in the IR spectrum (KBr). The negative HR FAB-MS spectrum produced a molecular ion at m/z 925.4771[M-H]⁻ providing the formula C₄₇H₇₄O₁₈ (Calcd. for C₄₇H₇₃O₁₈: 925.4800). The negative FAB-mass spectrum also exhibited a molecular ion due to [M-H]⁻ at m/z 925. The ¹H-NMR spectrum of 1 (Table I) confirmed the presence of four tertiary methyl groups [δ 1.10 (s), 1.11 (s), 1.30 (s) and 1.77 (s)]; two secondary methyl groups [δ 1.04 (d, J =6.5 Hz) and 1.36 (d, J = 3.8 Hz)], three anomeric protons [δ 4.47 (1H, d, J = 7.8 Hz), 4.86 (1H, d, J = 4.1 Hz), and 5.55 (1H, d, J = 6.0 Hz)], two primary alcohol functions [3.92 (1H, m), 4.21 (1H, d, J = 10.2 Hz), H_{CH2OH} -glc; and 3.75 (1H, m), 3.90 (1H, m) H_{CH2OH}-glc'] and two olefinic protons [δ 4.59 (br s) and 4.73 (br s)]. The ¹³C-NMR and DEPT spectrum showed 47 carbon signals, including six methyl groups (δ 12.7, 14.4, 14.9, 17.7, 18.0 and 19.8), 12 methylene groups, 22 methine groups (16 oxygenated) and 7 quaternary carbon groups (2 oxygenated). In the ¹³C-NMR spectrum, chemical shifts confirmed the presence of one carbonyl group at δ 217.4 (>C=O, C-3), monosubstituted double bonds at δ 151.4 and δ 111.0 (>C=CH₂, C-20 and C-29) and one carboxylate group at δ 176.4 (>COO, C-28). The assignment of signals was made by comparing results with the ¹³C-NMR data of 24-nor-11α-hydroxy-3-oxo-lup-20(29)-en-28-oic acid (Lischewski et al., 1985). The shift values of the aglycone moiety of lupane skeleton were in a good agreement, except for the three-unit sugar moiety of 1. On the other hand, the shift values of δ 95.5, 104.6 and 103.1 were typical for the three anomeric carbons of the inner glucopyranosyl, outer glucopyranosyl and rhamnopyranosy (terminal) and agreed with that of acantrifoside A isolated from A. trifoliatus and

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Fig. 1. Chemical structure of compound 1

A. koreanum (Yook et al., 1998). Based on the above spectral data, the structure of 1 was determined as 24-nor-11α-hydroxy-3-oxo-lup-20(29)-en-28-oic acid 28-O-α-L-rhamnopyranosyl-(1 \rightarrow 4)- β -D-glucopyranosyl-(1 \rightarrow 6)- β -D-glucopyranosyl ester, which we named acantrifoside B (1).

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