# An Easy and Convenient Synthesis of Optically Active $\boldsymbol{\beta}$-Amino Alcohols and 1,2-Diamines. Applications in Enantioselective Deprotonation of Cyclohexene Oxide 

Sung Hye Shin, Sang Kyu Kang, and Byung Tae Cho*<br>Department of Chemistry. Hallym Unnersity. Chunchon, Kangwon-Do 200-702, Korea Received August 18, 2003

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Optically active $\beta$-dialkylamino alcohols ${ }^{1}$ and 1,2-diamines ${ }^{2}$ are not only a common structural component in a vast group of naturally occurring and synthetic molecules, but also can be widely used as versatile chiral building blocks and chiral catalysts or ligands in a variety of asymmetric synthesis. such as enantioselective dialkylzinc addition to aldehydes. ${ }^{3}$ enantioselective conjugated addition ${ }^{1}$ and enantioselective deprotonation of meso ketones and epoxides. ${ }^{5}$ Accordingly, many synthetic methods including aminolysis of chiral epoxides, ${ }^{\text {5ithb }} 5$ optical resolution of their racemic mixtures ${ }^{6}$ and reduction of non-racemic $O$-acetyl mandelamides ${ }^{7}$. reduction of chiral $\alpha$-amino carboxamides, ${ }^{\text {tha }} 8$ asymmetric reduction of $\alpha$-amino ketones ${ }^{9}$ and aminolysis of aziridinium salts ${ }^{4.5 .8}$ for those compounds have been reported. However, aminolysis of chiral epoxides is commonly accompanied by the formation of undesired regioisomers. The resolution method of racemic mixture suffers from providing intrinsic limitation where the maximum yield of one enantiomer from the starting material is only $50 \% .{ }^{6}$ In the case of reduction of $\alpha$-amino carboxamides, unnatural $\alpha$-amino acids are not economical viable to use them as starting materials since they are expensive, but also racemization can occur in the reaction of some $N$-protected amino acids with dialkylamines to give $\alpha$-amino carboxamides. For example,

O'Brien et al reported significant levels of racemization when $N$-protected phenylglycine was reacted with pyrrolidine in the presence of coupling reagent to form the corresponding $\alpha$-amino carboxamide. ${ }^{83.10}$ Recently, we reported the synthesis of nearly enantiopure $\beta$-adrenergic agonists ${ }^{13}$ and $\beta$-hydroxy nitriles ${ }^{12}$ from optically active $1,2-$ diol monotosylates 1 obtained from CBS-oxazaborolidinecatalyzed borane reduction of $\alpha$-sulfonyloxyketones. ${ }^{13}$ We wish to report here an easy and simple method for the synthesis of nearly enantiopure $\beta$-dialkylamino alcohols 2 and 1,2-diamines 4-10 starting from I and their applications in enantioselective deprotonation of cyclohexene oxide using chiral lithium amides.
The monotosylates $1^{13}$ were directly reacted with 3 equiv. of $N, N^{\prime}$-dialkylamines under solvent-free conditions at 40-50 ${ }^{\circ} \mathrm{C}$ for $2-96 \mathrm{~h}$ to give 2 in $80-93 \%$ yield (Scheme 1). The optical purities and absolute configurations of 2 were determined by HPL.C analysis using chiral columns and/or by comparing optical rotation values of the known compounds. As shown in Table 1, all the products 2 obtained have very high optical purities approaching $100 \%$ ee. To obtain chiral diamines 4-10, the reaction was carried out by treatment of 2.0 equiv. of methanesulfonyl chloride with each of the selected amino alcohols, such as $\mathbf{2 c}, \mathbf{2 h}, \mathbf{2 j}$, and

 ( 1.0 eq ). $\mathrm{R}_{3} \mathrm{NH}_{2}(3.0 \mathrm{cq})$, water $(10 \mathrm{eq})$. It. $24-72 \mathrm{~h}(76-93 \%)$.

Table 1. Synthesis of Nearly I:nantiopure $\beta$-dialkylamino alcohols 2 from 1.2-1)iol Monotosylates $1^{\prime \prime}$

| No | Cpd | Time <br> (h) | $\begin{aligned} & \text { Yield } \\ & (\%)^{\prime \prime} \end{aligned}$ | $\begin{gathered} \mathrm{Mp} \\ \left({ }^{\circ} \mathrm{C}\right) \end{gathered}$ | Optical rotation (c. solvent) |  | \% ぜ | Conlig. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | This study | Values reported |  |  |
| 1 | 2a | 2 | 85 | ${ }_{0}$ i1 | $\|\alpha\|_{i j}^{\prime \prime \prime} 49.6(1.10, \mathrm{MeOH})$ |  |  |  |
| 2 | 2 b | 24 | 91 | oil | $\left.1 \alpha\right\|_{i}{ }^{\prime \prime} 56.4(1.30, \mathrm{ELOH})$ | $\|\alpha\|_{0}^{13)}-49.2(2.3, \mathrm{ElOH}), 86 \%$ ee,$R^{\text {6/b }}$ | 99 | S |
| 3 | 2 c | 96 | 93 | oil | $1 \alpha_{1}^{2010} 76.3\left(1.44, \mathrm{CHCl}_{3}\right)$ | d | 99 | $S$ |
| 4 | 2 d | 3 | 82 | 69-70 ( $\mathrm{lit}^{\text {¹6 }}$ (69.5-70.5) |  | $\|\alpha\|_{0}^{\text {Bi }}-40.3(1.88, \mathrm{ElOH}), 95 \%$ ee. $\mathrm{R}^{\text {Kl }}$ | 99 | S |
| 5 | 2 e | 3 | 90 | 77.79 (1i1. ${ }^{\text {5b }}$ (69-71) | $\left.1 \alpha\right\|_{\nu 0} ^{-20} 54.7(0.53, \mathrm{ELOH})$ | $\|\alpha\|_{0}^{\text {Ei }}$ - $51.2(1.12, \mathrm{ELOH}), 97 \%$ ee, $R^{\text {flt }}$ | 99 | S |
| 6 | 2 f | 12 | 83 | 95-96(lit. $\left.{ }^{7} 96-98\right)$ | $\left.1 \alpha\right\|_{\nu} ^{-2010} 55.4(1.07, \mathrm{ELOH})$ | $\mid \alpha \times 10_{0}^{33}-43.6(1.2, \mathrm{ElOH}), 99 \%$ ee, $R^{+6}$ | $99{ }^{2}$ | S |
| 7 | 2 g | 3 | 80 | 97.99 | $1 \alpha_{0}^{20} 51.9\left(1.04, \mathrm{CHCl}_{3}^{20}\right)$ | d | 99 | S |
| 8 | 2 h | 3 | 92 | 116-118( liL. $^{\text {4b }} 107-108$ ) | $1 \alpha_{0}^{2010} 58.4\left(0.62, \mathrm{CHCl}_{3}\right)$ | $[\chi]_{\mathrm{D}}-57.8\left(0.53, \mathrm{CHCl}_{3}\right), 99 \% \mathrm{ee}, R^{\text {¢ }}$ | 99 | S |
| 9 | 2 i | 3 | 80 | 117-119 |  | d | 99 | S |
| 10 | 2 j | 3 | 89 | 111-113 ( $\mathrm{liL}^{\text {+6 }} 110-112$ ) | $\left.1 \alpha_{0}^{20}\right\|_{0} ^{20} 66.2\left(1.00, \mathrm{CHCl}_{3}\right)$ | $[x]_{0}-66.4\left(1.00, \mathrm{CHCl}_{3}\right), 99 \% \mathrm{ee}, R^{\text {b }}$ | 99 | S |
| 11 | 2k | 3 | 91 | oil | $\left.1 \alpha_{1}^{20}\right\|_{0} ^{20} 64.9\left(1.28, \mathrm{CHCl}_{3}\right)$ | $\left.1 \alpha\right\|_{0} ^{14}-72.4\left(18, \mathrm{CHCl}_{3}\right), 99 \%$ ee,$R^{3,}$ | 98 | S |
| 12 | 21 | 2 | 80 | oil | $1 \alpha_{10}^{20} 41.9\left(1.17, \mathrm{CHCl}_{3}\right)$ | d | $99^{\circ}$ | $S^{t}$ |

"1 was treated with 3.0 equiv. of"A, "-dialkylamines at $40-50^{\circ} \mathrm{C}$ under solvent-free conditions. "Isolated yield. "Compared by optical rotation values and absolute configurations reported. unless otherwise indicated. "Not reported. 'Compared by \% ee of the corresponding I andior optical rotation values and absolute configuration of diamines reported. 'Not reproted. but probably $S$ by analogy based on the ( + )-sign of optical rotation values. "Tetermined by HPIC analysis using Chiraleel OD-H (eluent : hexane/h-PrOH = $9 / 1$ ).

Table 2. Synthesis of Nearly Enantiopure 1.2-diamines 4-9 from 2"

"The reaction was carried oul with treatment of 2.0 equiv. of methanesulfonyl chloride with amino alcobols 2 in the presence of 3.0 equiv. of Et, in in cther folloved by direct treatment of 3,0 equiv of $\mathrm{MeNH}_{2}, ~ c-\mathrm{C}_{6} \mathrm{H}_{11}$ or $\mathrm{PhNH}_{2}$ with the resultitne apiridium salts 3 itn the presence of water at room temperature. ${ }^{h-2}$ See the corresponding footnotes in Table 1 .
21. in the presence of 3.0 equiv. of $\mathrm{Et}_{3} \mathrm{~N}$ in ether, followed by aminolysis of the resulting aziridium salts 3 with 3.0 equiv of $\mathrm{MeNH}_{2}, c-\mathrm{C}_{6} \mathrm{H}_{11}$ or $\mathrm{PhNH}_{2}$ in the presence of water at room temperature. ${ }^{4+10}$ The reaction produced nearly enantiopure diamines in $76-93 \%$ yields with no racemisation (Scheme 1 and lable 2).
In order to study the effect of substituents attached on the asymmetric center of chiral amines used as precursors of chiral lithium amides in the enantioselective deprotonation of meso-epoxide, we selected structurally different chiral diamines $4,7,8$ and 9 bearing phenyl, p-phenylphenyl, 2naphtyl and cyclohexyl groups at the steregenic center, respectively. We examined the enantioselective deprotonation of cyclohexene oxide using chiral lithium amides obtained from treatment of $n$-butyl lithium with these diamines by the known procedure. ${ }^{8 i}$ Of the chiral bases examined, $4-\mathbf{L i}$ afforded the best result to give ( $R$ )-2-cyclohexen-l-ol with $79 \%$ ee. $7-\mathbf{L i}$ and $9-\mathrm{Li}$ provided $60 \%$ ee and $70 \%$ ee, respectively. However the deprotonation using $8-\mathrm{Li}$ which is a bulkier lithium amide did not occur. The results are summarized in Scheme 2.
In summary, we have established an easy and convenient method for the synthesis of the optically active $\beta$-dialkyl-


4: $92 \%$ yield (24 h); 79\% ee
ligand 7: $35 \%$ yield ( 60 h ); $60 \%$ ee 8. no reaction 9: 60\% yield (48 h); 70\% ee

Scheme 2
amino alcohols 2 and 1,2-diamines 4-9 with near $100 \%$ ee that avoids the formation of undesired regioisomers. Enantioselective deprotonation of cyclohexene oxide using chiral lithium amides prepared from 4 and 9 provided ( $R$ )-2-cyclohexen-1-ol with $79 \%$ ee and $70 \%$ ee, respectively.

## Experimental Section

General. The reactions were monitored by TLC using silica gel plates and the products were purified by flash column chronatography on silica gel (Merck; 230-400 mesh). NMR spectra were recorded at 300 MHz for ${ }^{1} \mathrm{H}$ and

Notes
75 MHz for ${ }^{13} \mathrm{C}$ using $\mathrm{Me}_{4} \mathrm{Si}$ as the internal standard in $\mathrm{CDCl}_{3}$ unless otherwise noted. Optical rotations were measured with a high resolution digital polarimeter. Melting points were uncorrected. The starting materials 1 were prepared by ( $S$ )-CBS-oxazaborolidine-catalyzed borane reduction of a-tosy loxy ketones. ${ }^{13}$

Preparation of Chiral $\boldsymbol{\beta}$-Dialkylamino alcohols 2 from 1.
General procedure: The monotosylates 1 ( 2 mmol ) were treated with dialkylamines ( 6 mmol ) at $40-50{ }^{\circ} \mathrm{C}$ for appropriate times under solvent-free conditions. When 1 was disappeared from the reaction mixture by TLC examination. excess of amines were pumped off under reduced pressure. The residue was triturated with $1 \mathrm{~N} \mathrm{NaOH}(20 \mathrm{~mL})$ and extracted with ether. The extract was dried over anhydrous $\mathrm{MgSO}_{4}$, filtered and concentrated. The crude product 2 obtained was further purified by flash cluromatography on silica-gel using ethyl acetate/methanol ( $1 / 2$ ) or EtOAc/hexane ( $1 / 2$ ) as an eluent. The solid products were recrystallized from hexane. All the products $\mathbf{2}$ except $\mathbf{2 c}, \mathbf{2 g}, \mathbf{2 i}$ and $\mathbf{2 l}$ are the known compounds which have been reported in literatures. ${ }^{\text {3a.ab.c. } .7}$ All spectroscopic data (IR, ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR) of these compounds are good agreement with those data reported
(S)-2-N,N-Dibutylamino-1-phenylethanol 2c: $R_{f} 0.26$ (EtOAc/hexane 1 : 2): oil; $[\alpha]_{\mathrm{D}}^{19} 76.24$ (c 1.44. $\mathrm{CHCl}_{3}$ ); IR (KBr. neat, $\mathrm{cm}^{-1}$ ) $3438.3406 .2956,2931,2871.2861,2823$.
 $J=7.43 \mathrm{~Hz}), 1.26-1.52(\mathrm{~m}, 8 \mathrm{H}) .2 .39-2.67(\mathrm{~m}, 6 \mathrm{H}) .4 .33(\mathrm{br}$ $\mathrm{s}, \mathrm{IH}) .4 .62$ (dd. $1 \mathrm{H}, J=3.58 \& 10.45 \mathrm{~Hz}), 7.24-7.38$ (m. 5 H ): ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 14.53,20.98 .29 .69$. 53.97. 63.38. 69.50. 126.02, 127.53, 128.49. 142.73: Calcd. for $\mathrm{C}_{16} \mathrm{H}_{37} \mathrm{NO}: \mathrm{C} .77 .06 ; \mathrm{H} .10 .91$ : N, 5.62. Found: C. 77.00 ; H, 10.94: N. 5.45.
(S)-2-(1-Piperidino)-1-(+-chlorophenyl)ethanol 2g: $R_{\mathrm{f}}$ 0.21 (EtOAc/MeOH $1: 2$ ): mp $97.99{ }^{\circ} \mathrm{C}$ (hexane); $[\alpha]_{\mathrm{D}}^{20}$ 51.9 (c 1.04. $\mathrm{CHCl}_{3}$ ); IR ( $\mathrm{KBr}, \mathrm{cm}^{-1}$ ) 3110, 3095. 2959. 2932. 2877, 2808, 1486. 1150, 1090. 879. 823: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.81-1.83(\mathrm{~m} .6 \mathrm{H}) .2 .42-2.54(\mathrm{~m} .4 \mathrm{H})$. $2.71-2.77$ (m. 2 H ). 4.15 (br s. 1 H ). 4.67 (dd. $1 \mathrm{H} . J=3.30 \&$ $10.59 \mathrm{~Hz}) .7 .30-7.35(\mathrm{~m}, 4 \mathrm{H})$ : ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 23.91. 54.03. 64.17, 70.25. 127.47, 128.67. 133.21, 141.24: Calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{ClNO}:$ C. 65.13: H, 7.57: N. 5.84. Found: C. 65.16: H. 7.43: N. 5.88
(S)-2-(1-Pyrrolidino)-1-(2-naphthyl)ethanol 2i: $R_{\mathrm{f}} 0.16$ (EtOAc/MeOH 1:2): mp 117-119 ${ }^{\circ} \dot{\mathrm{C}}$ (hexane): $[\alpha]_{\mathrm{D}}^{20} 47.8$ (c $1.10 . \mathrm{CHCl}_{3}$ ): IR $\left(\mathrm{KBr} . \mathrm{cm}^{-1}\right) 3442,3406.3102 .1096$. 3056. 2968. 2938. 2816. 2701, 1125. 823. 891: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.81-1.85(\mathrm{~m}, 4 \mathrm{H}) .2 .54-2.59(\mathrm{~m}, 4 \mathrm{H})$. $2.76-2.90(\mathrm{~m} .2 \mathrm{H}) .4 .21$ (br s. 1 H ). 4.88 (dd. $1 \mathrm{H} . J=3.16 \&$ $10.59 \mathrm{~Hz}) .7 .45-7.52(\mathrm{~m} .3 \mathrm{H}) .7 .82-7.87(\mathrm{~m} .4 \mathrm{H}):{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 24.36 .54 .66 .64 .69 .71 .51,124.77,125.24$, 125.41. 126.76. 128.70, 134.10, 140.57. Calcd for $\mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}$ : C. 79.63: H. 7.94: N. 5.80 . Found: C. 79.57: H. 7.489: N. 5.82.
(S)-2-(1-Piperidino)-1-cyclohexylethanol 2l: $R_{\mathrm{f}} \quad 0.19$ ( $\mathrm{EtOAc} / \mathrm{MeOH} 1: 1$ ): oil: $[\alpha]_{\mathrm{D}}^{20} 41.9\left(c 1.10 . \mathrm{CHCl}_{3}\right)$ : IR (KBr. neat. $\mathrm{cm}^{-1}$ ) 3426, 2927, 2852, 2793, 1448: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz} . \mathrm{CDCl}_{3}\right) \delta 0.97-1.31(\mathrm{~m} .6 \mathrm{H}) .1 .38-1.76(\mathrm{~m} .11 \mathrm{H})$.
1.91-2.60 (m, 6H). $3.78(\mathrm{~m} . \mathrm{lH}), 4.32(\mathrm{br} \mathrm{s}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz} . \mathrm{CDCl}_{3}$ ) $\delta 24.66,26.45 .26 .52,26.96 .28 .81,29.44$. 42.83. 62.67, 70.04; Calcd for $\mathrm{C}_{13} \mathrm{H}_{2} \mathrm{NO}: \mathrm{C}, 73.88 ; \mathrm{H}$, 11.92; N. 6.63. Found: C, 73.85 : H, 11.86; N. 6.57.

## Preparation of Chiral Diamines $\mathbf{4 - 9}$.

General procedure ${ }^{4 b, 1 i}$ : To a stirred solution of amino alcohol $2(2 \mathrm{mmol})$ in dry ether $(10 \mathrm{~mL})$ were added $\mathrm{Et}_{3} \mathrm{~N}$ ( 6 mmol). It was cooled to $0^{\circ} \mathrm{C}$ and mathanesulfonyl chloride ( 4 mmol ) was added drovise. The resulting reaction misture became sticky. After $0.5 \mathrm{~h}_{1}, \mathrm{Et}_{3} \mathrm{~N}(4 \mathrm{mmol})$ and $\mathrm{MeNH}_{2}(2.5$ $\mathrm{mL}, 40 \%$ aqueous solution; 16 mmol ) was added and the reaction mixture was stirred at $0^{\circ} \mathrm{C}$ to room temperature for 20 h [Whenc $c-\mathrm{C}_{6} \mathrm{H}_{11} \mathrm{NH}_{2}$ and $\mathrm{PhNH}_{2}$ instead of $\mathrm{MeNH}_{2}$ were used, the same reaction was carried out with the amine (6 mmol) in the presence of water ( 10 mmol )]. The organic and aqueous layers were separated and the aqueous layer was extracted with ether ( $15 \mathrm{~mL} \times 3$ ). The combined ether extracts were washed with saturated $\mathrm{NaHCO}_{3}$ and brine. The organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated in vacuo. The crude product diamines were further purified by flash chromatography on silica-gel using ethyl acetate/methanol ( $1 / 4$ ) or hexane $/ i-\mathrm{PrOH}(1 / 1)$ as an eluent. All the products except 7 and 9 are the known compounds have been reported in literatures. ${ }^{4,7,7,8 a}$ All spectroscopic data (IR. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR) of these compounds are good agreement with those data reported.
(S)-N-Methyl-1-cyclohexyl-2-(1-piperidino)ethanamine 9: $R_{\mathrm{f}} 0.11$ ( $\mathrm{EtOAc} / \mathrm{MeOH} \mathrm{I}: 4$ ): oil; $[\alpha]_{\mathrm{D}}^{20}-15.9$ (c 1.22 , $\mathrm{CHCl}_{3}$ ); IR ( KBr , neat. $\mathrm{cm}^{-1}$ ) $3318,2924,2850.2789$. 1447. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz} . \mathrm{CDCl}_{3}$ ) $\delta 0.87-1.31(\mathrm{~m}, 10 \mathrm{H}) .1 .43-$ $1.58(\mathrm{~m} .6 \mathrm{H}), 1.59-1.7 \mathrm{I}(\mathrm{m} .4 \mathrm{H}), 2.43 \mathrm{~s} .3 \mathrm{H}), 2.39-2.53(\mathrm{~m}$. $3 \mathrm{H}) .2 .68(\mathrm{~m}, \mathrm{IH}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz} . \mathrm{CDCl}_{3}$ ) $\delta 25.38$, 26.83. 27.74, 30.61. 32.96. 26.78, 37.69. 50.12. 50.63, 69.51; Calcd. for $\mathrm{C}_{14} \mathrm{H}_{28} \mathrm{~N}_{2}$ : C. 74.94 : H. 12.58: N. 12.48 . Found: C , 75.16 ; H. 12.28; N. 12.33.

## Enantioselective Deprotonation of Cyclohexene Oxide. ${ }^{8}$

 The reaction using $9-\mathbf{L i}$ as a chiral base is representative. To a solution of $9(2 \mathrm{mmol})$ in THF ( 6 mL ) was added a solution of $n$-BuLi in hexane ( $1.6 \mathrm{M}, 2.2 \mathrm{~mL}: 3.5 \mathrm{mmol}$ ) dropwise at $0^{\circ} \mathrm{C}$. After stirring for 0.5 min , a solution of cyclohexane oxide ( 2 mmol ) in THF was added slowly and the mixture was stirred at $0^{\circ} \mathrm{C}$ to room temperature for 48 h . Most of THF was removed in vacuo at $0^{\circ} \mathrm{C}$. and the reaction mixture was extracted with ether ( $15 \mathrm{~mL} \times 3$ ). The combined extracts were washed with water and brine. The organic layer was dried over anhydrous $\mathrm{MgSO}_{4}$ and concentrated in wacho. The crude product was chromatographed to provide 2 -cycloheven- 1 -ol in $60 \%$ yield. Its optical purity determined by a capillary GC analysis using a $30 \mathrm{~m} \beta$-Dex 120 chiral column (Supelco) showed it to be $70 \%$ ee in $R$-enantiomer (oven temp: $100^{\circ} \mathrm{C}$ : isothermal: $\mathrm{t}_{R} 32.76 \mathrm{~min}$ for $S$-isomer and $\mathrm{t}_{R} 33.70 \mathrm{~min}$ for $R$-isomer).Acknowledgment. This work was supported by grant no. R05-2002-000-00003-0 from the Basic Research Program of the Korea Science and Engineering Foundation and the Research Grant from Hallym University. Korea.

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