

A Flavone Glycoside from *Angelica gigas* Roots

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Abstract – A flavone glycoside was isolated from the roots of *Angelica gigas* (Umbelliferae) and identified as diosmin [diosmetin-7-*O*- α -L-rhamnopyranosyl (1 \rightarrow 6)- β -D-glucopyranoside] by spectroscopic methods. This is the first report of a flavone glycoside from *Angelica* species.

Key words – *Angelica gigas*, Umbelliferae, flavone glycoside, diosmin.

Introduction

Angelica gigas is genus of the family Umbelliferae. *A. gigas* grows on moist soils of Korea. The roots of this plant were used under the Korean name Zam Dang Gui. *A. gigas* has been used as traditional medicine not only for treatment anemia but also as a sedative, an anodyne or a tonic agent (Yook, 1990).

A. gigas has been studied extensively and are shown to contain a variety of substances including coumarins (Chi, 1969; Jung *et al.*, 1991; Konoshima *et al.*, 1968; Pachaly *et al.*, 1996; Ryu *et al.*, 1990), essential oils (Chi and Kim, 1988) and polyacetylenes (Choi *et al.*, 2000).

In the previous paper, we reported the isolation of coumarins and uracil from *A. gigas* roots (Lee *et al.*, 2002).

Our continuing phytochemical investigation from the roots of *A. gigas* has resulted in the isolation of a flavone glycoside, which was isolated for the first time from this plant.

Experimental

Instruments and reagents – Positive FAB- and EI-MS spectra were measured with Jeol JMS-AX505WA mass spectrometer. ^1H -, ^{13}C -NMR and HMBC spectra were recorded with Bruker AVANCE 400 NMR spectrometer in DMSO- d_6 using TMS as an internal standard. IR spectrum was recorded with Jasco FT/IR-300E instrument on KBr disc. Other reagents were commercial grade without purification.

Plant materials – The roots of *Angelica gigas* Nakai were purchased from Kyung Dong market in March 2001, Korea and verified by Prof. Emeritus H. J. Chi, Seoul National

University, Korea. A voucher specimen has been deposited at the Herbarium of Natural Products Research Institute, Seoul National University, Korea.

Extraction and isolation – The air-dried powdered roots (5 kg) of *A. gigas* were extracted and subfractionated by same method described as earlier (Lee *et al.*, 2002). The portion of *n*-BuOH fraction (20 g) was chromatographed on silica gel eluting with a gradient of CHCl_3 -MeOH to afford compound **1** (15 mg).

Compound **1**; FAB-MS, m/z 609 $[\text{M} + \text{H}]^+$; ^1H -NMR, (400 MHz, DMSO- d_6) δ : 12.93 (1H, *s*, 5-OH), 7.57 (1H, *dd*, 8.5, 2.1, H-6'), 7.45 (1H, *d*, 2.1, H-2'), 7.14 (1H, *d*, 8.5, H-5'), 6.83 (1H, *s*, H-3), 6.77 (1H, *d*, 2.1, H-8), 6.46 (1H, *d*, 2.1, H-6), 3.88 (3H, *s*, 4'-OCH₃); ^{13}C -NMR, (100 MHz, DMSO- d_6) δ : 181.9 (C-4), 164.2 (C-2), 162.9 (C-7), 161.2 (C-5), 156.9 (C-9), 151.3 (C-4'), 146.8 (C-3'), 122.9 (C-1'), 118.9 (C-6'), 113.1 (C-2'), 112.2 (C-5'), 105.4 (C-10), 103.8 (C-3), 100.5 (C-1'''), 99.9 (C-1''), 99.5 (C-6), 94.8 (C-8), 76.2 (C-3''), 75.6 (C-5''), 73.1 (C-2''), 72.0 (C-4'''), 70.7 (C-2'''), 70.3 (C-3'''), 69.5 (C-4''), 68.3 (C-5'''), 66.0 (C-6''), 55.8 (4'-OCH₃), 17.8 (C-6'''); IR, ν_{max} (KBr) cm^{-1} : 3424 (OH), 1733 (CO).

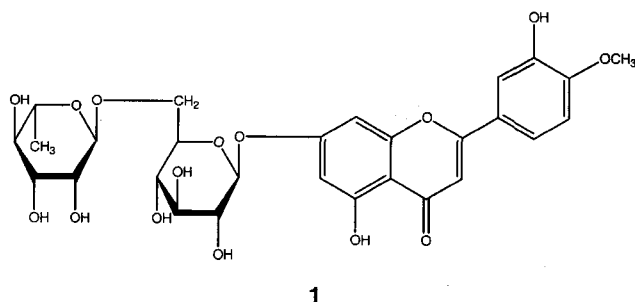
Acid hydrolysis of 1 – A MeOH soln. of **1** in 2 N HCl heated at 100° for 1 hr gave diosmetin, L-rhamnose and D-glucose identified by EI-MS, ^1H - and ^{13}C -NMR analysis (Han *et al.*, 1987; Son *et al.*, 1994).

Results and Discussion

The roots of *A. gigas* were extracted with MeOH. The MeOH extract was suspended in water, and then fractionated successively with equal volumes of Et₂O and *n*-BuOH. The *n*-BuOH extract was purified by chromatography on silica gel to afford compound **1**.

Compound **1** was obtained as white-yellow crystals from

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MeOH. The positive FAB-MS of **1** showed an ion $[M + H]^+$ at m/z 609. In the $^1\text{H-NMR}$ spectrum, the typical flavonoid signals were observed. The singlet at δ 12.62 assigned the aromatic 5-OH of (A) ring of a flavonoid. The proton signals at δ 8.02 (d, $J = 1.9$ Hz), δ 7.50 (dd, $J = 1.9, 8.4$ Hz) and δ 6.90 (d, $J = 8.4$ Hz) showed ABX splitting pattern of (B) ring of a flavonoid. The singlet signal at δ 3.85 was 4'-OCH₃ by HMBC assignments and the proton signals at δ 3.00-5.00 showed glycosides. Its $^{13}\text{C-NMR}$ spectrum of **1** showed C=O at δ 177.8 and OCH₃ at δ 56.4. The anomeric protons of glucose and rhamnose showed at δ 5.51 (d, $J = 7.7$ Hz) and δ 4.55 (s), respectively. According to the coupling constant of an anomeric proton, glucose attached to 7-OH of aglycone, diosmetin. And in the $^{13}\text{C-NMR}$ spectrum, the terminal sugar was determined as a rhamnose by lowfield chemical shift of glucose C-6 methylene (δ 66.0). The carbon signals at δ 99.9 and δ 100.5 showed glucosyl C-1" and rhamnosyl C-1"', respectively. The IR spectrum of **1** indicated the presence of hydroxy at 3424 cm^{-1} and carbonyl group at 1733 cm^{-1} . The acid hydrolysis of **1** yielded diosmetin, L-rhamnose and D-glucose identified by EI-MS, $^1\text{H-}$ and $^{13}\text{C-NMR}$ analysis. Consequently, the structure of **1** was established as diosmin [diosmetin-7-O- α -L-rhamnopyranosyl (1 \rightarrow) - β -D-glucopyranoside].

There have been no previous reports of flavone glycosides from *Angelica* species. To our knowledge, this is the first report of a flavone glycoside from *Angelica* species. It is reported the isolation of diosmin from *Evodia rutaecarpa* (Kang *et al.*, 1997).

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