

- 68.0, 67.3, 67.2, 63.2, 59.0.
17. **10**: IR (neat, cm^{-1}) 3300 (alkyne), $^1\text{H NMR } \delta$ (400 MHz, CDCl_3) 5.80 (ddd, 1H, $J_1=6.9$ Hz, $J_2=10.5$ Hz, $J_3=17.4$ Hz, 2-H), 5.30 (m, 2H, 1-H,2), 4.80 (m, 6H, 2x (OCH_2O) (MEM)), 4.23 (m, 1H, 4-H), 3.4-4.0 (m, 14H, 3x $(\text{OCH}_2\text{CH}_2\text{O})$ (MEM) and 3-11 and 5-11), 3.36 (s, 9H, 3x OCH_3 (MEM)), 2.60 (m, 2H, 6-H), 1.95 (t, 1H, $J=2.6$ Hz, 8-H); MS (FAB) 315 (M^{-1} , 100%).
18. Miyamoto, K.; Kubodera, N.; Ochi, K.; Matsunaga, I.; Murayama, E. *Eur. Pat. Appl.* **1985**, EP 184,206.

19. **12**: $[\alpha]_D^{25} = -92.5^\circ$ ($c=0.62$, CHCl_3); UV λ_{max} (EtOH) 265 nm; $^1\text{H NMR } \delta$ (400 MHz, CDCl_3) 6.36 (d, 1H, $J=11.2$ Hz, 6-11), 6.02 (d, 1H, $J=11.2$ Hz, 7-11), 5.42 (m, 1H, 19-11), 5.08 (m, 1H, 19-11), 4.22 (m, 1H, 1-11), 4.15 (m, 1H, 3-H), 3.50 (m, 1H, 2-H), 3.04 (s, 1H, OH), 2.80 (dd, $J_1=12.0$ Hz, $J_2=3.8$ Hz, 1H, 9-11), 2.58 (s, 1H, OH), 2.48 (m, 2H, 4-11), 2.24 (s, 1H, OH), 0.91 (d, 3H, $J=6.3$ Hz, 21- CH_3), 0.87 (d, $J=1.7$ Hz, 3H, 26- CH_3), 0.85 (d, $J=1.7$ Hz, 3H, 27- CH_3), 0.54 (s, 3H, 18- CH_3).

Efficient Synthesis of Hydroxyethylidene and (*E*)-Alkene Dipeptide Isosteres

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The development of novel dipeptide isosteres possesses a great value and importance in peptidomimetics. Among more than dozen peptide isosteres,¹ (*E*)-alkene dipeptide isostere² is a suitable amide bond surrogate in terms of mimicking the rigidity, bond angles, and bond length of the amide bond. We wish to report here general and efficient synthesis of hydroxyethylidene³ and (*E*)-alkene dipeptide isosteres, which would considerably increase their application to drug and development.

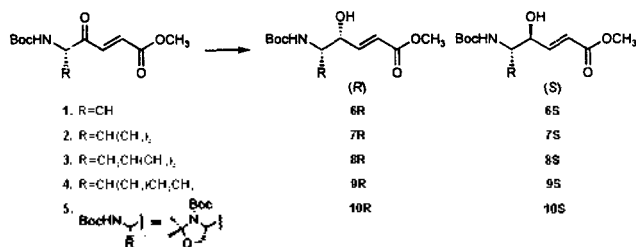
Hydroxyethylidene dipeptide isostere first reported by Hanson *et al.*^{3a} is an interesting dipeptide analog which combines conformational restriction, function as statine mimics, and the ability to undergo conjugate addition to enzyme nucleophiles such as cysteine thiol. Previous syntheses^{3b,c,d} of hydroxyethylidene dipeptide isosteres mainly resorted to the Hansons method,^{3a} which was hampered by the lack of stereoselectivity in the vinylmagnesium halide addition to amino aldehydes, long reaction steps, and low overall yields. As a solution to the synthetic problem in preparing hydroxyethylidene dipeptide isosteres, we have developed an efficient route from ketovinyl dipeptide isostere.⁴ Reduction of ketovinyl dipeptide isostere gives the corresponding hydroxyethylidene dipeptide isostere in one step (Scheme 1).

Various reducing agents including NaBH_4 , $\text{Zn}(\text{BH}_4)_2$, LiBF_4H , *L*-selectride, *S*-selectride, $\text{LiAl}(i\text{-Bu})(n\text{-Bu})_3$, and NaBH_4CN were used and additives such as Et_3BOMe ,

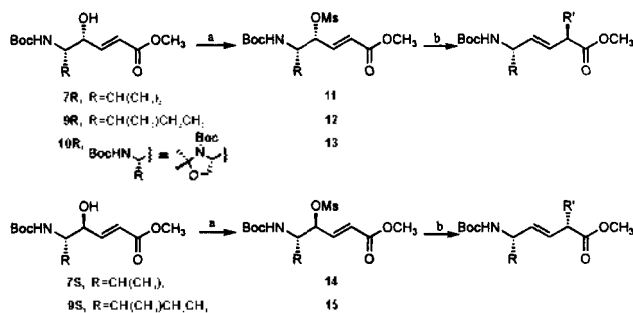
ZnCl_2 , CuCl_2 , and SmCl_3 were employed. Even though the stereoselectivity of reduction was moderate (product ratio, (*R*):(*S*)=88:12-18:82),⁵ combined isolated yields of (*R*)- and (*S*)-alcohols were good to excellent (55-99%). Furthermore, by employing some additives (CuCl_2 , SmCl_3) diastereoselectivity could be reversed and both diastereomers of hydroxyethylidene dipeptide isosteres could be prepared. Due to ease access of ketovinyl dipeptide isosteres from amino acids,⁴ this synthetic route constitutes an efficient and general pathway for hydroxyethylidene dipeptide isosteres. Conversion of hydroxyethylidene to (*E*)-alkene dipeptide isosteres through γ -mesyloxy (*E*)- α,β -enoate intermediates was completed by using Ibbuka's method.^{3b}

Anti-S_N2' displacement of γ -mesyloxy leaving group with organocopper. BF_3 complex provided (*E*)-alkene dipeptide isostere in a stereoselective manner (Scheme 2). Experimental results are summarized in Table 1.

The salient features of this synthetic route for (*E*)-alkene dipeptide isosteres include: (1) the relatively few number of steps required, (2) excellent chemical yields and stereoselectivity. Due to the simplicity and efficiency in preparation of scalemic γ -hydroxy α,β -enoates (hydroxyethylidene) and a



Scheme 1.



Scheme 2. Reagents and conditions: (a) MsCl , pyridine, CH_2Cl_2 , 0°C . (b) CuCN , $\text{R}'\text{MgCl}$, $\text{BF}_3 \cdot \text{Et}_2\text{O}$, THF , -78°C .

Table 1. Synthesis of (*E*)-alkene dipeptide isosteres

substrate	(<i>E</i>)-alkene dipeptide isostere	yield	diastereo-Selectivity (R) : (S) ^b	[α] _D ^c (c. CHCl ₃)
11				
	R-CH(CH ₃) ₂			
	R'-CH(CH ₃) ₂	89	99:1	-35.9(0.88)
	R'-CH ₂ CH(CH ₃) ₂	88	99:1	-40.3(0.88)
	R'-C(CH ₃) ₃	55	95:5	-26.5(0.92)
12	R-CH(CH ₃)CH ₂ CH ₃			
	R'-CH	69 ^d	86:14	-22.2(0.94)
	R'-CH(CH ₃)	89	99:1	-29.5(0.87)
	R'-CH ₂ CH(CH ₃) ₂	85	99:1	-42.4(1.02)
	R'-C(CH ₃) ₃	74 ^e	99:1	-2.3(1.07)
13				
	R'-CH(CH ₃) ₂	95	99:1	-56.2(1.42)
	R'-CH ₂ CH(CH ₃) ₂	93	99:1	-41.2(1.07)
	R'-C(CH ₃) ₃	93	99:1	-58.6(0.57)
14				
	R-CH(CH ₃) ₂			
	R'-CH	83 ^f	1:99	+18.9(1.31)
	R'-CH(CH ₃) ₂	92	1:99	+17.8(1.60)
	R'-CH ₂ CH(CH ₃) ₂	91	1:99	+26.5(1.09)
	R'-C(CH ₃) ₃	73 ^g	2:98	+8.02(1.09)
15	R-CH(CH ₃)CH ₂ CH ₃			
	R'-CH ₃	84	1:99	+30.7(1.18)
	R'-CH(CH ₃) ₂	80	5:95	+37.7(1.13)
	R'-CH ₂ CH(CH ₃) ₂	93	1:99	+43.1(0.98)
	R'-C(CH ₃) ₃	88 ^h	1:99	+20.0(1.01)

^aIsolated yield of the major product. Configuration at C2 center.

^bObtained along with 11% reductive elimination product.

^cObtained along with 14% reductive elimination product.

^dObtained along with 11% reductive elimination product.

^eObtained along with 6% reductive elimination product.

^fObtained along with 15% reductive elimination product.

^gObtained along with 10% reductive elimination product.

variety of available organocopper reagents, this synthetic route for (*E*)-alkene dipeptide isosteres is a valuable addition to the arsenal of reliable synthetic pathways and will be often applied to peptidomimetics.

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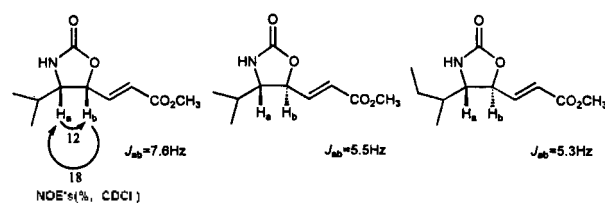
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5. The absolute configuration of **6R** and **6S** compounds was determined by the comparison of ¹H NMR chemical shift values of those compounds with the literature^{1b} values of compound **6S**. In the case of compounds **7R**, **7S**, and **9R** corresponding oxazolidone derivatives were prepared and the absolute configuration was determined by the comparison of coupling constants (*J*_{ab}) with the literature values and the NOE experiment.



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