

Biological Monitoring of Human Exposure to Volatile Halogenated Hydrocarbons Using Urinalysis with Capillary GC-ECD

Won Tae Jung and Dong Hun Sohn*

National Institute of Public Health, Shirokanedai 4-6-1, Minato, Tokyo 108, Japan

*College of Pharmacy, Chungang University, Seoul 156-070, Korea

(Received January 17, 1992)

Abstract □ For the risk assessment of human exposure to volatile halogenated hydrocarbons, a dynamic purge trap/on-column cryofocusing method using capillary gas chromatograph-⁶³Ni electron capture detector and thermal desorption unit was applied to analyze the free forms, metabolites of 1,1,2-trichloroethylene and 1,1,2,2-tetrachloroethylene. The urine sample was diluted with distilled water, hydrolyzed and sealed. Then the inert gas was infused to purge out free 1,1,2-trichloroethylene, free 1,1,2,2-tetrachloroethylene and trichloroethanol. These compounds were trapped to Tenax^R GC-gas trap device throughout clean up tube. Being undetectable to gas chromatograph directly, trichloroacetic acid was methyl esterified and trapped in the manner above mentioned. The optimal incubation time to get best recovery of methyl ester was 4 hours at 60°C. The concentrations of free volatile halogenated hydrocarbons and their metabolites in urine were obtained from 5 healthy volunteers. This analytical method is expected to make the biological monitoring, more precise and convenient.

Keywords □ Biological monitoring, volatile halogenated hydrocarbons, urinalysis, GC-ECD, purge trap/on-column cryofocusing method.

Volatile halogenated hydrocarbons (VHH) are widely used in high technology industry as a degreasing agent or solvent for dry cleaning. Most of them are discharged into ambient air due to their volatilities^{1,2}, and into ground water. Therefore, the main invasion route of VHH to human body is thought to be inhalation by breathing^{3,4}, and oral intake of drinkable water. Being VHH reported to pose grave problems of environmental pollution, there are many reports about the analytical method of VHH in ambient air⁵⁻⁷, and water⁸⁻¹⁰. For the analysis in biological sample from the industrial worker exposed to high levels of VHH, the spectrophotometric method using Fujiwara reaction¹¹ is usually adapted because of its simplicity of operation. Head space method is adapted for more accurate determination of VHH. However, this method needs incubation time and accurate pre-treatment for gas-liquid equilibrium and even large volume of sample (20-30 ml) for an analysis. So, few information for

analysis of urine containing trace level of VHH is available due to above problems.

To get over these problems, purge trap/on-column cryofocusing method¹² which has been developed for the trace analysis of VHH in aqueous sample such as rain water were modified and applied to the risk assessment of human exposure to VHH¹³. This method doesn't need precise treatment, incubation time or equilibrium temperature those are essential for head space method. Free form of 1,1,2-trichloroethylene (TRI), 1,1,2,2-tetrachloroethylene (PER) in human urine¹⁴ from airborne environment were analyzed with capillary gas chromatograph-⁶³Ni electron capture detector (ECD). The final metabolites of VHH are well known as trichloroethanol (TCE) and trichloroacetic acid (TCA)¹⁵. These compounds were also analyzed because they could be the indices of biological monitoring¹⁶. The urine samples were diluted with distilled water, sealed in vials. TRI, PER and TCE were purged out

by inert gas and trapped in Tenax^R GC tube through the clean up tube and analyzed. Being undetectable to GC directly, TCA was methyl esterified (TCA-M) according to Breimer's method¹⁷⁾ and analyzed identically. The optimal incubation time was investigated. The concentrations of TRI, PER, TCE and TCA obtained from 5 healthy volunteers are also described. The obtained results indicate that this analytical method could be successfully applied to monitor the VHH and their metabolites as the indices of human exposure to VHH.

EXPERIMENTAL METHODS

Materials & sample collection

Analytical grade reagents (sulfuric acid, dimethyl sulfuric acid, sodium carbonate and methanol) were purchased from the Wako Pure Chem. (Tokyo, Japan). The standards (1,1,2-trichloroethylene, 1,1,2,2-tetrachloroethylene, 2,2,2-trichloroethanol and trichloroacetic acid methyl ester) were purchased from the Tokyo Kasei (Tokyo, Japan). Distilled water was heated and purged out with nitrogen gas (purity: 99.9998%) to eliminate any possibility of VHH contamination. Possible contamination of the equipments used (Tenax^R GC, trap tube, test tube, vial, septa) was eliminated by washing with distilled water, heating and drying in oven.

Five healthy male volunteers, aged 22-56 years, working for Japanese National Institute of Public Health were selected for urine specimen collection. They had no experience to be exposed to VHH at least for 6 months. The volume of urine excretion for 24 hours was not considered because this method was examined for possibility of analysis. Five or 10 ml of human urine was taken in test tube and no sooner air tightly capped. These test tubes were kept below 4°C prior to pre-treatment.

Pre-treatment of sample

Urine specimen, 1 ml was taken into a vial and diluted with 1 ml of distilled water. Then, 1 ml of H₂SO₄ was added to hydrolyze the conjugated compounds and sealed air tightly with Teflon^R faced silicone septa. The sample vial was purged out and trapped to gas trap device for TRI, PER and TCE analysis. For TCA analysis, diluted urine sample with 1 ml of H₂SO₄ was sealed and incubated at 60°C for 3 hours. And 0.1 ml of (CH₃O)₂SO₂ was

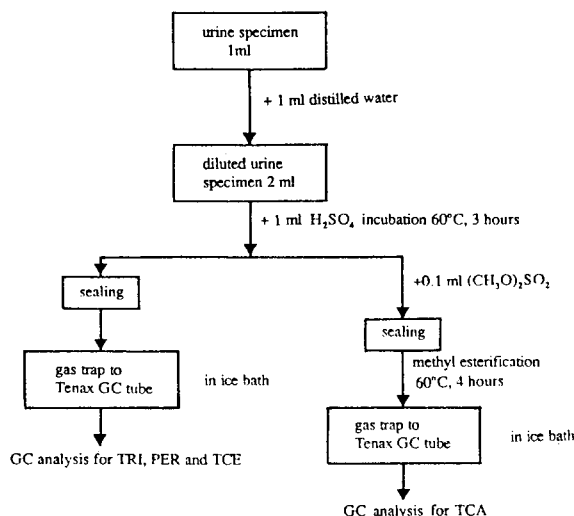


Fig. 1. Pre-treatment of urine specimen prior to analysis.

spiked and incubated again at 60°C for 4 hours. Then the TCA-M was purged out and trapped to gas trap device, as shown in Fig. 1.

Gas trap device

Fig. 2 shows the diagram of gas trap device. The clean up tube containing sodium carbonate was used to eliminate the mist of sulfuric acid and dimethyl sulfate. Being kept at 0°C, the vial was purged by nitrogen gas. The purging flow rate was 20 ml/min for 4 min as our previous report¹²⁾.

Analytical method

The apparatus used were a Shimadzu GC-15A gas chromatograph with ⁶³Ni-ECD and Shimadzu C-R4A integrator. DKK GAS-20 with DKK-OMRON thermal desorption unit was used as purge trap/on-column cryofocusing system for urinalysis as shown in Fig. 3. Purge gas was inserted to gas trap tube heating at 200°C for thermal desorption. Flow rate was 20 ml/min for 4 min. Following analytical procedures were identical with those of our previous report¹⁸⁾.

Tenax^R GC used was beads of polymeric 2,6-diphenyl-*p*-phenylene oxide (35/60 mesh). The capillary column used was Shimadzu CBP-1 (midbore: 0.25 mm i.d. × 25 m L.) fused silica column chemically bonded with 0.25 μm layer of methyl silicone. The same one with 60 cm length was used as cryo-

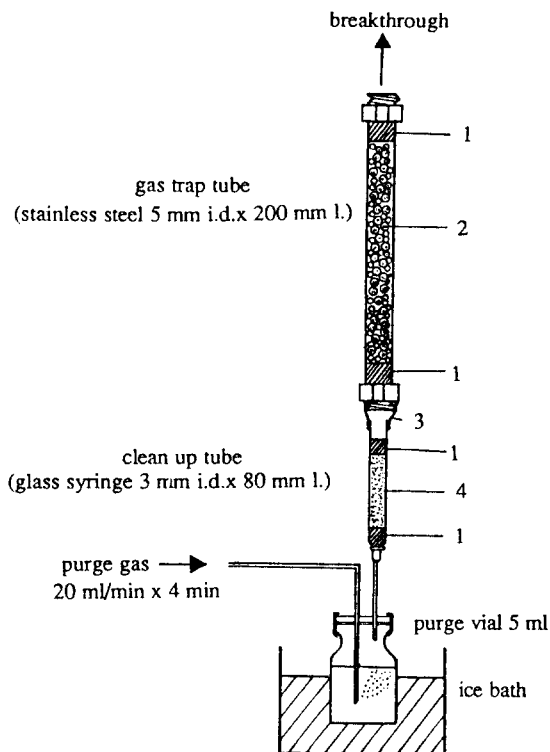


Fig. 2. Diagram of gas trap device.
 1. glass wool; 2. Tenax^R GC (35/60 mesh beads of polymeric 2,6-diphenyl-*p*-phenylene oxide); 3. polyethylene tube; 4. sodium carbonate (Na₂CO₃).

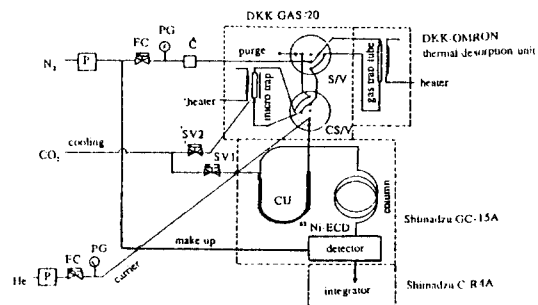


Fig. 3. Schematic diagram of analytical apparatus for urinalysis.
 p: purifier; FC: flow controller; PG: pressure gauge; SV: solenoid vail; C: charcoal filter S/V, CS/V: 6 port chromatographic valves; CU: on-column cryofocusing unit.

focusing column. The initial column oven temperature was 45°C and raised to 145°C at the end of

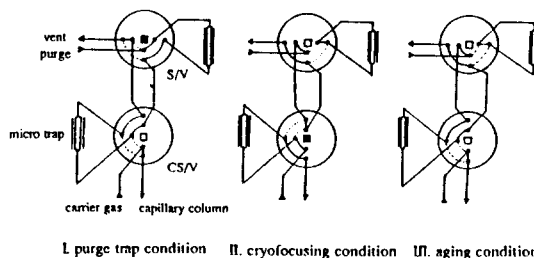


Fig. 4. Operation of 6 port chromatographic valves to switch the pathway for gas trapping.
 purge gas: N₂ (99.9998%); carrier gas: He (99.9999%); ■ valve on; □ valve off

Table I. Detection limits of volatile halogenated hydrocarbons in aqueous sample by purge trap/on-column cryofocusing method

Compound	\bar{D} . able	D. limit	
		ng/l level	ppb level
1,1,2-trichloroethylene	0.445	1.336	0.230
1,1,2,2-tetrachloroethylene	0.018	0.055	0.004

\bar{D} . able; detection ability calculated as $D_1 + D_2 + D_3 + \dots + D_n/n$
 D. limit; detection limit calculated as \bar{D} . able $\times 3$

50 min (2°C/min). The operation of 6 chromatographic valves is shown in Fig. 4.

Calibration curves were constructed by spiking 5 μ l of standard solution (Each ml contains; CICH-CCl₂ 1 μ g, CCl₂: CCl₂ 1 μ g, CCl₃CH₂OH 1 μ g and CCl₃COOCH₃ 1 μ g). The procedures for pre-treatment and analysis were similar to those of sample preparation. The detection limits and reproducibilities of VHH by this method were also verified.

RESULTS AND DISCUSSION

Detection limits and reproducibilities of VHH

The response to amount sampled from a standard solution was observed to be linear in the range required. To verify the detection limits of VHH by this method, the responses to amount sampled from aqueous solution were calculated in according to following equation referred to JEMCA method¹⁹.

$$\bar{D} = t(n - 1, 0.005) \times \delta R/n \times dC/dR$$

where \bar{D} = detection ability

Table II. Reproducibilities of volatile halogenated hydrocarbons in aqueous sample by purge trap/on-column cryofocusing method

Compound	Amount sampled(ng/ml)	Response (Peak area, mV)					Coefficient of variance
		1	2	3	4	5	
1,1,2-trichloroethylene	0.23	692,017	766,825	656,023	764,290	601,886	9.1
1,1,2,2-tetrachloroethylene	0.11	1,697,605	1,784,913	1,818,917	1,525,977	1,871,707	4.8
						mean	6.9%

Table III. Concentrations of volatile halogenated hydrocarbons and their metabolites in urine specimen (unit: $\mu\text{g/l}$)

Volunteers	Sex	Age	Body weight(kg)	n ^a	Free form		Metabolites	
					TRI	PER	TCE	TCA ^b
FM	Male	56	60	3	1.24± 0.08	0.82± 0.02	0.74± 0.08	0.69± 0.09
HK	Male	28	71	4	3.26± 0.12	2.90± 0.10	1.08± 0.07	0.78± 0.08
TF	Male	22	74	3	1.03± 0.07	0.94± 0.06	0.92± 0.03	0.95± 0.04
WT	Male	29	88	3	1.80± 0.07	0.95± 0.04	0.99± 0.01	0.63± 0.05
NH	Male	38	65	3	2.64± 0.30	1.33± 0.05	1.42± 0.11	1.78± 0.06
				Mean	1.99	1.22	1.03	0.96

^aNumber of analysis, (The data are given as mean± SD)

^bTCA-M was expressed in term of TCA.

$t(n-1, 0.005)$

= "t" value with 95% confidence

δR = standard deviation

n = number of analysis

dC = amount sampled

dR = average response

As shown in Table I, detection limits by this method were much lower than 1-10 $\mu\text{g/l}$ for TRI, 0.04-4 $\mu\text{g/l}$ for PER provided for head space method.²⁰⁾ Many reports about the determination of VHH using head space method have been reported,^{14-17,21)} but sensitivity with approximate 1-5 $\mu\text{g/l}$ of detection limit is not enough to determinate using small volume of aqueous sample, because this method uses only a part of target compound which are in gaseous phase equilibrated from aqueous phase. Reproducibilities of VHH by this method were also inspected (Table II) by repeated analysis with spiking the standard solution. This method showed average 6.9% coefficient of variance which is more reproducible than 10-20% provided for head space method. In a matter of fact, VHH determination using head space method has been reported to have approximate 10.7% coefficient of variance²²⁾.

The optimal time for hydrolysis and methyl esterification

Vials each of which containing 1 ml of distilled water and 1 ml of urine divided from the same specimen were prepared and added H_2SO_4 1 ml. Then each 3 of the vials were incubated at 60°C for 1, 2, 3, 4, 5 and 6 hours, respectively. The best recoveries of TRI, PER and TCE were obtained by 3 hours' incubation. After 3 hours' hydrolysis, each 3 of them were methyl esterified by adding 0.1 ml of $(\text{CH}_3\text{O})_2\text{SO}_2$ and incubating at 60°C for 1, 2, 3, 4, 5 and 6 hours with mechanical shaking. In methyl esterification, the best recoveries were obtained at 4 hours' incubation at 60°C as shown in Fig. 5. Therefore, this results were applied for the analysis of all specimen.

VHH concentrations in human urine

Table III shows the concentrations of free VHH and their metabolites in human urine. The concentration of TCA-M was expressed in term of TCA. The minimum and maximum ranges were 0.63 $\mu\text{g/l}$ -3.26 $\mu\text{g/l}$, respectively. The average urine concentration from 5 healthy volunteers were 1.99 $\mu\text{g/l}$ for

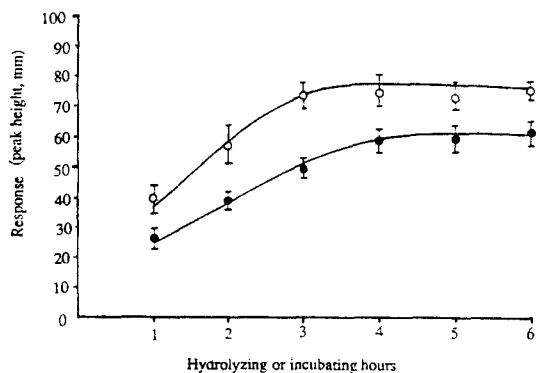


Fig. 5. The optimal conditions for hydrolysis and methyl esterification.

○ trichloroethanol (TCE); ● trichloroacetic acid methyl ester (TCA-M); The data are given as mean \pm SD by 3 determinations.

TRI, 1.22 $\mu\text{g/l}$ for PER, 1.03 $\mu\text{g/l}$ for TCE and 0.98 $\mu\text{g/l}$ for TCA. The concentrations were relatively lower than those reported by Hajimiragha²³. As a conclusion, our biological monitoring method using dynamic purge trap system could be applied to determine the trace level of VHH in the urine from the human who exposed at low level in domestic place. This method is expected to be used risk assessment of VHH at very trace levels.

ACKNOWLEDGEMENTS

We are grateful to Ms Hisako Yaki, Ms Hiroko Tatematsu and Fumiko Matsuzawa for their devoted assistance for this work.

LITERATURE CITED

- Singh, H. B., Salas, L. J. and Stiles, R. E.: Distribution of selected gaseous organic mutagens and suspected carcinogens in ambient air. *Environ. Sci. Technol.*, **16**, 872 (1982).
- Bozzelli, J. W. and Kebbekus, B. B.: A study of some aromatic and halocarbon vapors in the ambient atmosphere of New Jersey. *J. Environ. Sci. Health*, **17**, 693 (1982).
- Cothorn, C. R., Conglio, W. A. and Marcus, W. L.: Estimating risk to human health. *Environ. Sci. Technol.*, **20**, 111 (1986).
- Lahl, U., Cetinkaya, M., Von Duszeln, J., Stachel, B. and Tiemann, W.: Health risk from volatile halogenated hydrocarbons. *Sci. Total Environ.*, **20**, 171 (1976).
- Ohta, T., Morita, M. and Mizoguti, I.: Local distribution of chlorinated hydrocarbons in the environmental air in Tokyo. *Atmos. Environ.*, **10**, 557 (1976).
- McClenny, W. A.: Automated cryogenic preconcentration and gas chromatographic determination of volatile organic compounds in air. *Anal. Chem.*, **56**, 2947 (1984).
- Noy, T., Fabian, P., Borchers, R., Janssen, F., Cramers, C. and Rijks, J.: Trace analysis of halogenated hydrocarbons in gaseous samples by on-line enrichment in air adsorption trap, on-column cold trapping and gas chromatography. *J. Chromatogr.*, **393**, 3436 (1987).
- Oston, R. and William, D. T.: Head space chromatographic determination of water pollutants. *Anal. Chem.*, **54**, 942 (1982).
- Grob, K. and Zurcher, F.: Stripping of trace organic substance from water equipment and procedure. *J. Chromatogr.*, **117**, 285 (1976).
- Warner, J. M. and Beasley, R. K.: Purge and trap chromatographic method for the determination of acrylonitrile, chlorobenzene, 1,2-dichloroethane and ethylbenzene in aqueous samples. *Anal. Chem.*, **56**, 1953 (1984).
- Ikeda, M. and Ohtuji, H.: A comparative study of the excretion of Fujiwara reaction-positive substance in urine of humans and rodents given trichloro-, tetrachloro-derivatives of ethane and ethylene. *Brit. J. Industr. Med.*, **29**, 99 (1972).
- Jung, W. T. and Fujita, M.: Optimal conditions of purge trap/on-column cryofocusing method with capillary gas chromatography for determination of volatile halogenated hydrocarbons in aqueous samples. *Eiseikagaku*, **37**, 395 (1991).
- Triebig, G. and Schaller, K. H.: Air monitoring of solvent exposed workers with passive samplers in comparison to "biological monitoring (BM)". *Toxicol. Environ. Chem.*, **12**, 285 (1986).
- Ikeda, M., Ohtuji, H., Immamura, T. and Komoike, Y.: Urinary excretion of total trichloro compounds, trichloroethanol and trichloroacetic acid as a measure of exposure to trichloroethylene and tetrachloroethylene. *Brit. J. Industr. Med.*, **29**, 328 (1972).
- Ogata, M., Takatsuka, Y. and Tomokuni, K.:

- Excretion of organic chloride compounds in the urine of the persons exposed to vapours of trichloroethylene and tetrachloroethylene, *Brit. J. Industr. Med.* **25**, 103 (1984).
16. Ogata, M., Shimada, Y. and Taguchi, T.: A new micro-determination method used in an analysis of the excretion of trichloro-compounds in the urine of workers exposed to trichloroethylene vapour, *Industr. Health*, **25**, 103 (1987).
 17. Breimer, D. D., Ketelaars, H. C. J. and VanRossum, J. M.: Gas chromatographic determination of chloral hydrate, trichloroethanol and trichloroacetic acid in blood and in urine employing head space analysis. *J. Chromatogr.*, **88**, 55 (1974).
 18. Fujita, M., Jung, W. T., Tatematsu, H., Sohn, D. H. and Maeda, T.: Automated analysis of volatile halogenated hydrocarbons in rain water and ambient air by purge trap capillary gas chromatography, *HRC*, **14**, 83 (1991).
 19. Japanese Environmental Monitoring Chemical Association, "Analytical Methods of Environmental Monitoring with Commentary", vol I ed. by JEMCA, Marugen Press, Tokyo, 1985. p.209
 20. JIS K 0125 Testing method for low molecular weight halogenated hydrocarbons in industrial water, JIS committee (1987).
 21. Nolan, R. J., Fresher, N. L., Rick, D. L., McCarty, L. P. and Saunders, J. H.: Kinetics and metabolism of inhaled methyl chloroform (1,1,1-trichloroethane) in male volunteers, *Fundam. Appl. Toxicol.*, **4**, 654 (1984).
 22. McNally M. E. and Grobe R. L.: Head space determination of solubility limits of the base neutral and volatile components from the environmental protection agents list of priority pollutants, *J. Chromatogr.*, **284**, 105 (1984).
 23. Hajimiragha, H., Ewers, U., Janssen-Rosseck, R. and Brockhaus, B.: Human exposure to volatile halogenated hydrocarbons from the general environment, *Int. Arch. Occup. Environ. Health.*, **58**, 141 (1986).