

Preparation and Properties of Frame Retardant Microencapsulated with polyurea

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1. INTRODUCTION

Entrapment can control the release of active ingredients, permit liquids to be handled as solids, protect reactive components until time of use, allow the safe handling of toxic materials, extend the shelflife of delicate materials, and overcome product incompatibilities.

Polyurea microcapsules are completely insoluble in water and other common solvents and can incorporate both hydrophobic and hydrophilic active materials. We studied the microencapsulation of fragrant materials with various polymer matrices.

Frame retardants, as a core material, can be protected from the microcapsules by entrapping in the wall. The productive rate of the functional material from the microcapsules can be controlled by the chemical structure of the capsule wall, its thickness, and the particle size of the microcapsules. Especially, the chemical structure of the monomer in the encapsulation by interfacial polycondensation is one of the important parameters determining the physical properties of the microcapsules.

In the present study, the physical properties of polyurea microcapsules was investigated by varying the type of monomeric diisocyanates.

2. EXPERIMENTAL

Polyurea microcapsules were formed by carrying out an interfacial polycondensation reaction in emulsion globules. An organic phase with the various diisocyanates as wall-forming material and various frame retardants as the core substance. The O/W emulsion was formed by adding the organic solution into the aqueous solution by stirring vigorously at ambient temperature. EDA solution was added into the O/W emulsion after stirring for 10 min to prevent an agglomeration among the resultant emulsion globules. Reaction for more than 120 min gave the formation of polyurea microcapsules.

The microcapsules were decanted, filtered out,

washed and dried in a vacuum oven at 25°C for at least 48hr.

3. RESULT AND DISCUSSION

Fig.1 shows FT-IR spectra of prepared polyurea microcapsule containing phosphorous flame retardant with the various diisocyanates. This figure shows that the disappearance of absorption pick at 2250 cm^{-1} by free-nco in diisocyanates.

At 1675 cm^{-1} by H-bonded urea carbonyl, at 1705 cm^{-1} by free carbonyl, and at 3300 cm^{-1} by H-bonded N-H absorption picks were observed.

And the absorption pick at 1240 cm^{-1} by phosphorous-oxygen double bond in TPP was also observed. From these results confirm that polyurea microcapsule containing TPP were successfully prepared.

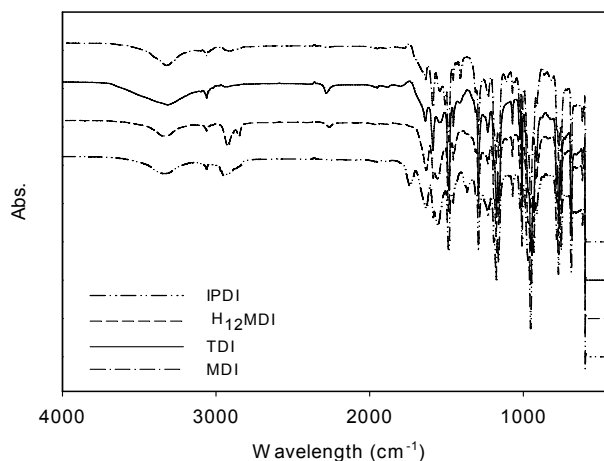


Fig. 1. FT-IR spectra of polyurea microcapsule containing TPP with the various diisocyanates.

Surface morphologies of the microcapsule wall from the different diisocyanates are shown in Fig. 2.

In the case of TDI, it is confirmed that a rapid random reaction occurred between TDI in the organic phase and EDA in the water-soluble phase on the emulsion globules, producing a very rough wall membrane. The different of diisocyanates reactivity induced from the chemical structure brings

about the various membrane morphologies, which can significantly determine the permeability, crystallinity, and thickness of the resultant microcapsules.

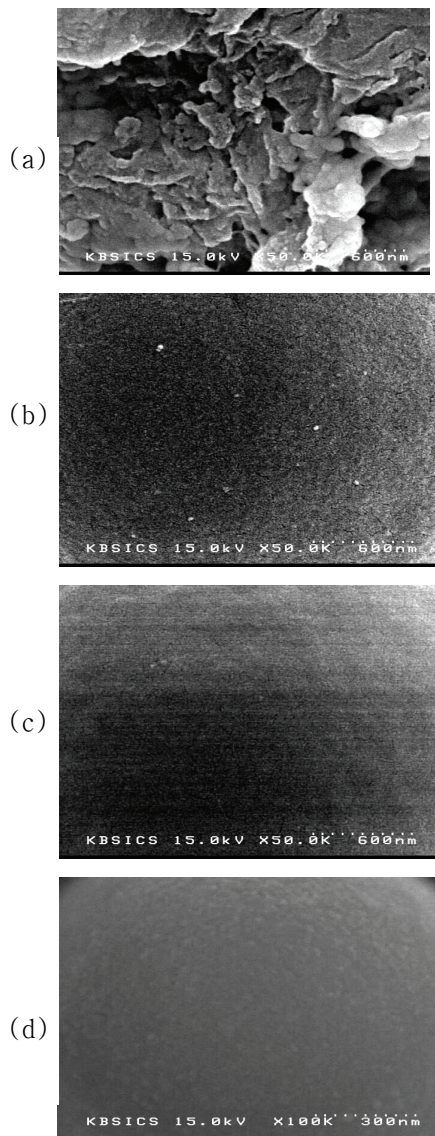


Fig. 2. SEM photograph of polyurea microcapsule from different diisocyanate : (a) TDI; (b) IPDI; (c) H12MDI; (d) MDI

Polyester sheets were dyed and finished with various flame retardant compounds. (commercial A,B and M/C)

The durable flame retardancy was obtained when the phosphorous content in PET fabrics was about more than 4.0%. The impartment of flame retardancy (SE grade and LOI 28 up) for polyester sheets was possible by treating them with mixed dispersions of M/C containing organic phosphorous compounds.

Table 1. Flame retardant properties of polyester sheets finished with the microcapsule containing phosphorous compounds

Items	Length of burnt position (mm)	Burning time (s)	Burning rate (mm/min)	LOI
Control	70.0	49.0	85.7	19.0
F-A *1	35.0	32.0	65.6	21.2
F-B *2	10.0	8.0	SENBR	24.0
F-C *3	-	-	SE	28.1
F-D *4	-	-	SE	29.2

Note *1 Dyed/finished with commercial A
 *2 " " with " B
 *3 Dyed/finished with TCP M/C
 *4 " " with TPP M/C

4. REFERENCE

- [1] Nixon, J. R. Microcapsulation; Marcel Dekker: New York, 1976
- [2] Hong, K.; Park, S. Master Chem Phys 1999, 58, 128.
- [3] Hong, K.; Park, S. React Funct Polym, in press.
- [4] Hong, K.; Park, S. Polymer, accepted.
- [5] Fuyama, H.; Shinjo, G.; Tsuji, K. J Pest Sci 1984, 9, 511
- [6] Yadav, S. K.; Khilar, K. C.; Suresh A. K. J Membr Sci 1997, 125, 213