

AP02

Fabrication and Mössbauer Characterization of MgFe₂O₄/ Fe₃O₄ Composite Nano-particles

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In recent years, nano-crystalline spinel ferrites have been investigated intensively due to their potential applications. It is found that magnesium ferrite, MgFe₂O₄, can be used as n-type semi-conducting gas-sensing materials [1]. In the mean time, H. Zhang and coworkers speculated that MgFe₂O₄ could be used to replace Fe₃O₄ for magnetic targeted drug delivery [2]. In addition, synthesis and study of nano-sized spinel ferrite particles is of great fundamental interest.

In the present study, MgFe₂O₄/ Fe₃O₄ composite nanoparticles with pure spinel structure are successfully fabricated at 90 °C by Chemical Oxidation in Aqueous Solution. Fig.1 shows the TEM image and selected area electron diffraction(SAED) pattern of the sample. Fig.2 shows a Mössbauer spectrum taken at room temperature. The composition is determined as (MgFe₂O₄)_{0.33}(Fe₃O₄)_{0.67} by Atomic absorption spectrometry and Mössbauer spectroscopy. Magnetization hysteresis loop indicates that the saturation magnetization is much larger than that of the normal MgFe₂O₄ nanoparticles. The Mössbauer spectra collected at different temperatures (87 K-293 K) show that the hyperfine field is smaller than that of the bulk and increase with decrease of measuring temperature. This can be understood by the collective magnetic excitation theory [3].

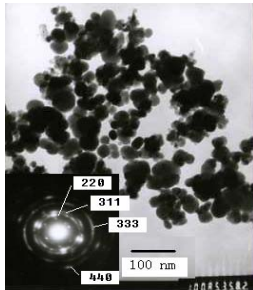


Fig. 1. TEM image and SAED pattern of MgFe₂O₄/Fe₃O₄ nanoparticles.

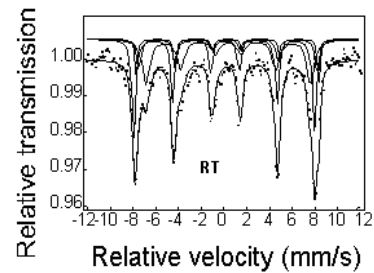


Fig. 2. Mössbauer spectrum for MgFe₂O₄/Fe₃O₄ nanoparticles collected at room temperature.

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AP03

Withdrawn