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## Discovery of Novel Nuclear Factor- $\kappa$ B (NF- $\kappa$ B) Inhibitors using chroman-2-carboximide derivatives

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Nuclear factor (NF)- $\kappa$ B, an inducible transcription factor, regulates the immune response and plays critical roles in the pathogenesis of chronic inflammatory diseases and a variety of human cancers. It has been suggested that NF- $\kappa$ B function inhibitors may be useful as both anti-inflammatory agents and antitumor agents. During the search for novel compounds that can inhibit NF- $\kappa$ B activation, 6-hydroxy-7-methoxy-chroman-2-carboxylic acid phenyl amide (**KL-1156**) was identified as a good inhibitor of NF- $\kappa$ B activation. NF- $\kappa$ B inhibitory activity of **KL-1156** with an  $IC_{50}$  value of 40.4  $\mu$ M was comparable to that of pyrrolidine dithiocarbamate (PDTC) with an  $IC_{50}$  value of 37.2  $\mu$ M. PDTC acts as an antioxidant and has been found to be a potent inhibitor of NF- $\kappa$ B activation. In the present study, we report the synthesis and inhibitory effect of chroman-2-carboxylic acid *N*-(substituted)phenyl amide. Hydroxy and methoxy substituents of compound **KL-1156** were removed in the target compounds. Substituents on *N*-phenyl ring of the target compounds were selected considering their electronic and hydrophobic character. Their NF- $\kappa$ B inhibitory activities were evaluated on lipopolysaccharide (LPS)-stimulated macrophage RAW 264.7. The NF- $\kappa$ B inhibitory activities of the compounds were compared with that of **KL-1156**.

The target compounds contained various substituents (H, Cl, OMe, CH<sub>3</sub>, CF<sub>3</sub>, and NO<sub>2</sub>) on phenyl ring. The positional effects of the substituents were also explored by examining the compounds with substituents at various position (2-, 3-, 4-, 2,5-, 3,4-, 3,5-). Compounds with H, CF<sub>3</sub> and NO<sub>2</sub> substituents on phenyl ring were inactive ( $IC_{50}$ : > 100 $\mu$ M). In compounds bearing substituent at 4-position, the activity decreases in the order of Cl ( $IC_{50}$ : 18.2 $\mu$ M), OH ( $IC_{50}$ : 44.5 $\mu$ M), and Me ( $IC_{50}$ : 95.8 $\mu$ M). In compounds bearing substituent at 3-position, compound with Cl ( $IC_{50}$ : 74.7 $\mu$ M) and OMe ( $IC_{50}$ : 76.8 $\mu$ M) were equipotent. In compounds bearing substituent at 2-position, only compound with 2-OH substituent showed good activity ( $IC_{50}$ : 22.1 $\mu$ M). The positional effects of substituents on inhibitory activities are contradictory. Compound with 2-OH ( $IC_{50}$ : 22.1 $\mu$ M) showed more potent activity than compound with 4-OH ( $IC_{50}$ : 44.5 $\mu$ M) while compound with 4-Cl substituent were 4 times more potent than compound with 3-Cl ( $IC_{50}$ : 74.7 $\mu$ M) substituent. Direct correlation was not observed based on the electronic and hydrophobic character of the substituents.