부양증발가스응축장치로 제조한 페란이트 나노분말의 특성연구

The study of properties for Fe₃O₄ and NiFe₂O₄ nano powders synthesized by levitational gas condensation (LGC) method

Nuclear Nano Materials Development Lab., Korea Atomic Energy Research
Institute (KAERI)

Y. R. Uhm, B. S. Han, M. K. Lee, W. W. Kim and C. K. Rhee

High purity Fe₃O₄, NiFe₂O₄ powders were synthesized by a LGC method using micron powder feeding system, and its characterization was carried out by using XRD, TEM, XPS and Mössbauer spectroscopy. The TEM images for Fe₃O₄ and NiFe₂O₄ showed spherical particles. The range of particle size for Fe₃O₄ is from 14 to 26 nm. In case of NiFe₂O₄, it is from 8 to 22 nm. Mössbauer spectra for Fe₃O₄ consist of two sets of six Lorentzian lines corresponding to octahedral and tetrahedral sites. Two sets of sextet at Mössbauer spectrum for Ni-ferrite correspond to trivalent iron with size ranges from 12 to 25 nm at the Td and the Oh sites. Two doublets correspond to Fe^{3+} ion and Fe^{2+} ion at Oh site. The isomer shift of 0.54 (mm/s) for the second doublet exhibits that the sample contains Fe²⁺ ionic state. It means there are Ni³⁺ ions in the samples to have shown optimum charge valance and compensation. The analysis of Fe 2p3/2 photoelectron peak indicates different ionic state of Fe. The 2p3/2 peak at 709.5 and 711.2 eV is associated with $\mathrm{Fe^{2+}}$ and $\mathrm{Fe^{3+}}$. The sample shows two peaks at 854.43 due to $\mathrm{Ni^{2+}}$ and 855.96 eV due to Ni³⁺. From the results of Mössbauer spectrum and XPS, Ni-ferrite synthesized by LGC include unusual ionic state and it can be looked forward to attracting hole effect on the surface layer of particles.