

## High Temperature Stability of Bulk $Ti_3SiC_2$ Materials in Air

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The ternary carbides,  $Ti_3SiC_2$ , have attracted an enormous attention because of their unique combination of both metallic and ceramic properties [1,2]. Like metals, they have excellent electrical and thermal conductivities, high toughness, high fatigue-crack growth threshold, low hardness, good machinability, and high-thermal shock resistance. Like ceramics, they display excellent chemical resistance, high Young's modulus, high temperature strength, and high melting point.

$Ti_3SiC_2$  has been synthesized from various starting powders such as Ti/SiC/graphite [1-3], Ti/Si/graphite [4-6], Ti/Si/TiC [7], TiC/Si [8], etc. The employed processing methods were hot pressing, hot isostatic pressing [1-3,7,8], *in situ* hot pressing/solid-liquid process [4-6], etc. However, it has been known that the synthesis of bulk  $Ti_3SiC_2$  with high purity is not easy because relatively a long reaction time at high temperatures is required to synthesize via powder metallurgical routes. Recently, we have successfully synthesized dense  $Ti_3SiC_2$  with a relatively high purity by hot pressing newly developed  $Ti_x/Si$  powder mixtures.

The starting powders of  $TiC_x$  ( $x=0.6$ ) and Si (<45  $\mu m\phi$ , 99.7% purity) were weighed in a molar ratio of 3 : 1.1 using a microbalance, mixed in a SPEX™ shaker mill apparatus for 10 min, and hot pressed at 1360°C to 19 mm $\phi$  x 10 mm bulk specimens under 25 MPa Ar pressure for 90 min. During hot pressing,  $Ti_3SiC_2$  was synthesized by a direct reaction between  $TiC_x$  and Si without forming any intermediate phases. Bulk  $Ti_3SiC_2$  was cut into small coupons with a size of 10x8x0.5 mm<sup>3</sup>, ground to a 1000 grit finish, and then ultrasonically cleaned in acetone and methanol. Oxidation tests were performed in the temperature range between 25°C~1200°C in atmospheric air for up to 100 hr. The  $Ti_3SiC_2$  samples were etched using a HF-HNO<sub>3</sub>-H<sub>2</sub>O solution for microstructural evaluation. Furthermore, they were investigated by a thermogravimetric-differential thermal analyzer (TG-DTA), a scanning electron microscope (SEM) equipped with an energy-dispersive spectroscopy (EDS), an Auger electron spectroscopy (AES), and an X-ray diffractometer (XRD) with Cu-K $\alpha$  radiation.

The ternary layered carbide,  $Ti_3SiC_2$ , began to oxidize above 850°C appreciably. The oxidation resulted in the formation of an outer, coarse  $TiO_2$  grains, and an inner, fine ( $TiO_2$ +  $SiO_2$ )-mixed grains. During the initial stage of oxidation, oxygen diffused inward seriously, and carbon escaped from  $Ti_3SiC_2$ . Si tended to diffuse outward more than Ti. The oxide layer was softer and less stiff than the  $Ti_3SiC_2$  matrix.

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