

TiO₂ NANOSTRUCTURES AND APPLICATION TO DSSC AND ECD

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TiO₂ nanoparticles were controlled to 7~30 nm by a solvothermal synthetic method. Also, the TiO₂ nanotubes were prepared by reflux of titania in a concentrated NaOH, and the mesoporous films were fabricated by evaporation-induced self-assembly (EISA) process. These TiO₂ nanostructures were applied as electrode materials for dye-sensitized solar cell (DSSC) and/or electrochromic device (ECD). First, the DSSC derived from smaller-sized TiO₂ showed relatively higher J_{sc} in overall, since they have larger surface area to adsorb more N3 dye molecules. The J_{sc} of DSSC from 7 nm-sized TiO₂ was slightly lower than that of DSSC from 15 nm-sized particles, even though the adsorbed amount of dye was as high as 1.3 times. This suggests that the pore structure of TiO₂ electrode is an important factor in determining J_{sc}, because it influence on the ionic diffusion of electrolytes. Second, the phosphonated viologen was anchored on the surfaces TiO₂ nanostructures for the construction of a typical ECD. The effect of particle size and pore structure has been investigated with regard to the switching time and contrast ratio for the fabricated ECD. The ECD derived from 7 nm-sized TiO₂ demonstrated the highest contrast ratio with high optical transparency, whereas it showed the slowest switching response. We have also found that the mesoporous TiO₂ films can also be a possible candidate for the cathode material of ECD.