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## Charge-Discharge Properties of Amorphous Vanadium Polyanionic Borate as Anode Material for Lithium Secondary Battery

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Polyanionic compounds have engendered much interest not only as rare-metal-free cathode materials, but also as anode materials such as Fe<sub>3</sub>BO<sub>6</sub>, FeBO<sub>3</sub>, and VBO<sub>3</sub> for next-generation lithium secondary batteries. VBO3 is attractive as anode material than FeBO3, due to its low redox potential. However, VBO3 showed a large irreversible capacity at 1st cycle. In this study, we tried to reduce the irreversible capacity of VBO3 at 1st cycle, and to improve the synthetic process into a cheap and simple process by synthesizing amorphous VBO3 containing Li. To synthesis amorphous VBO<sub>3</sub> containing Li, the mixtures of V<sub>2</sub>O<sub>3</sub>, B<sub>2</sub>O<sub>3</sub>, and Li<sub>2</sub>CO<sub>3</sub> with the molar ratio of  $V_2O_3$ : $B_2O_3$ : $Li_2CO_3$ :AB = 1:1:x:y (x = 1,2 with y = 0 and y = 0,1,2,5 with x = 1) were melt at 1300 ?for about 10 min under atmospheric condition. The molten mixtures were then quenched at a cooling rate of about 1000 ?/s. Two The electrochemical anode performances were evaluated in coin-type cells with 1 M LiPF6/EC:DMC = 1 : 1 in volume and a polypropylene separator against a Li metal counter electrode. As Li content in synthetic process of water-quenched amorphous VBO3 increase, irreversible capacity of VBO3 decrease. This seems to come from structural modification of VBO3 into open structure by a glass modifier of Li2O. Both the reversible and irreversible capacities of air-quenched VBO3 increase with the content of AB added in synthetic process. This suggest that a number of electrochemically active site in air-quenched VBO3 increased by the addition of AB in synthetic process.

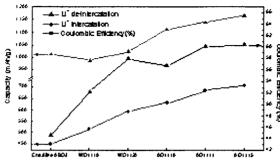


Fig. 1 Effects of Li and Acetheylene Black contents in synthetic process on capacity and coulombic efficiency of VBO3 at 1<sup>st</sup> cycle. (In Fig. 1, WQ and AQ indicate synthesized products by water quenching and by air quenching, respectively. Four numbers after WQ or AQ stand for molar ratio of V2O3, B2O3, Li2CO3, and Acetylene Black in sequence.)