

Synthesis of magnetic Fe₃N and Fe₃C nano-powders by chemical vapor condensation

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1. Introduction

Iron based magnetic nanopowders have been actively developed because of the potentially attractive application such as high density recording media or magnetic fluids, exploiting their magnetic properties superior compared to commercialized nanostructured magnetite (Fe₃O₄) powders. In this study, it was tried to synthesize the nanostructured Fe₃C and Fe₃N powders by chemical vapor condensation technique with various process parameters and perform the powder characterization such as microstructures, chemical compositions and magnetic properties.

2. Experiment procedure

Iron pentacarbonyl, Fe(CO)₅, was vaporized in the isolated bubbler container(50mm□x100mmH) at 160°C and transferred with the carrier gas of NH₃ for Fe₃N and CO for Fe₃C, respectively, into the hot-wall reactor tube (50mm□x400mmL) held at 500~850°C. The synthesized iron nano-powders were then generated into the chamber vacuumed mechanically and protected with argon gas. The produced powders were collected carefully after they were air-passivated passivation for 2 hours.

3. Results and discussion

It was discovered by X-ray diffraction patterns and chemical analysis results that the phases of produced powders were varied with different reaction temperatures and chamber pressures. That is, the pure Fe₃C and Fe₃N powders were synthesized in the limited conditions as shown in Figure. It was found here that Fe₃N powders were more effectively synthesized than Fe₃C because of the different kinetic behaviors of carbon and nitrogen gas in a vacuum condition. The mean sizes of powders produced at vacuum conditions were about 10~20 nm much lower than those at 1 atm. The coercivity and maximum magnetization of produced powders were measured and the relationship between the magnetic property and particle morphology is being studied with Moessbauer and XPS analysis.