## Structural Stability of Mn Based Oxide for Lithium Secondary Battery

리튬 2차전지용 망간계 산화물의 구조 안정성

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Recently, solid solution series between  $\text{Li}_2\text{MnO}_3$  ( $\text{Li}[\text{Li}_{1/3}\text{Mn}_{2/3}]O_2$ ) and  $\text{LiMO}_2$  (M=Cr, Ni, Co...) have received significant attention as an alternative cathode material for  $\text{LiCoO}_2$  due to their high discharge capacity.  $\text{Li}_2\text{MnO}_3$  was considered electrochemically inactive for  $\text{Mn}^{4^+}$  in  $\text{Li}_2\text{MnO}_3$  normally could not be oxidized beyond 4+ oxidation state in order to extract Li from its lattice. However, it is interesting that substantial charge can be removed from, and, to some extent, reinserted into the  $\text{Li}_2\text{MnO}_3$  structure. Moreover, the solid solution series between  $\text{Li}_2\text{MnO}_3$  and  $\text{LiMO}_2$  (M=Cr, Ni, Co..) showed very high discharge capacity over 200mAh/g when they cycled with an upper cut-off voltage of about 4.6  $^{\sim}$  4.8 V.

We investigated the electrochemical properties have Li[Ni<sub>x</sub>Li<sub>(1/3-2x/3)</sub>Mn<sub>(2/3-x/3)</sub>]O<sub>2</sub> and Li[Co<sub>x</sub> Li<sub>(1/3-x/3)</sub>Mn<sub>(2/3-2x/3)</sub>]O<sub>2</sub> compound prepared by a simple combustion method. The  $Li[Ni_xLi_{(1/3-2x/3)}Mn_{(2/3-x/3)}]O_2$  compound with low Ni content (X<0.25) showed stable cycle performance and sustained high discharge capacity of about 200 mAh/g after several cycles. On the contrary, the Li[Co<sub>x</sub>Li<sub>(1/3-x/3)</sub>Mn<sub>(2/3-2x/3)</sub>]O<sub>2</sub> compound displayed phase conversion to spinel-like phase during cycling. In this study, we characterized the electrochemical properties of Li[Ni<sub>x</sub> Li<sub>(1/3-2x/3)</sub>Mn<sub>(2/3-x/3)</sub>]O<sub>2</sub> and Li[Co<sub>x</sub>Li<sub>(1/3-x/3)</sub> Mn(2/3-2x/3)]O<sub>2</sub> compound. And, the cycling performance is compared by XRD and XANES. Specially, we tried to understand the origin of the difference of phase stability for two kinds of compound during cycling.