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Enzymatic and Energetic Properties of NADH:Ubiquinone
Oxidoreductase in the Marine Bacterium
Pseudomonas nautica

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Abstract Each oxidoreductase activity of the aerobic respiratory chain-linked NADH oxidase system in the marine bacterium *Pseudomonas nautica* was stimulated by monovalent cations including Na⁺, Li⁺, and K⁺. In the presence of NADH or deamino-NADH as electron donors, QH₂ formation was approximately 1.3 fold higher in Na⁺ than K⁺ at a concentration of 0.08 M, whereas the other reductase activities were not significantly higher in Na⁺ than K⁺. The optimal pH of NADH (or deamino-NADH):ubiquinone-1 oxidoreductase was 9.0 in the presence of 0.08 M NaCl. The activity of NADH (or deamino-NADH): ubiquinone-1 oxidoreductase was about 33% inhibited by 60 μM 2-heptyl-4-hydroxyquinoline-*N*-oxide (HQNO). The activity of NADH (deamino-NADH): ubiquinone-1 oxidoreductase was about 32 to 38% inhibited by 80 μM rotenone, whereas the activity was highly resistant to capsaicin. On the other hand, electron transfer from NADH or deamino-NADH to ubiquinone-1 generated a membrane potential ($\Delta\psi$) which was larger in the presence of Na⁺ than that observed in the absence of Na⁺. The $\Delta\psi$ was almost completely collapsed by 5 μM carbonylcyanide *m*-chlorophenylhydrazone (CCCP), and approximately 50% inhibited by 100 μM rotenone, or 60 μM 2-heptyl-4-hydroxyquinoline (HQNO). HQNO, also, made the $\Delta\psi$ very unstable. On the basis of the results, we suggest that the NADH: ubiquinone oxidoreductase of *P. nautica* is quite different in enzymatic and energetic properties compared to that of other marine halophilic bacteria.