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Metabolism of 1-Substituted Benzenes by *Pseudomonas putida* F1 and Its Mutants

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We have examined the metabolism of n-alkylbenzenes (C3-C7), biphenyl, styrene and cumene by the tod pathway from Pseudomonas putida F1 in terms of their catabolism by the pathway enzymes and their inducibility of the tod operon. F1 cells grown on toluene exhibited oxygen consumption in the presence of the chemicals examined. Toluene dioxygenase catalyzed the formation of monol, cis-dihydrodiol and triol metabolites from the n-alkylbenzenes tested and the triol formed from n-propylbenzene was metabolized to the derivative of 2-hydro-6-oxohexa-2,4-dienoate (HOHD) by the subsequent enzymes in the tod pathway. All the chemicals tested were catabolized by F1 cells grown on toluene to ring-cleavage HOHD derivatives, indicating that 6-methyl-HOHD hydrolase encoded by todF was a determinant for the further degradation of the selected 1-substituted benzenes. The results obtained from the enzyme activity assay and RT-PCR showed that not only the growth-supporting substrates but also n-propylbenzene, styrene and cumene act as inducers of the tod We have also isolated F1 mutants that can grow on benzene, toluene, operon. ethylbenzene, n-propylbenzene, cumene and biphenyl. The genetic backgrounds of the mutants will be discussed.

Introduction. Pseudomonas putida F1 utilizes benzene, toluene and ethylbenzene as carbon and energy sources for growth [7,8]. The enzymatic pathway for conversion of these aromatic hydrocarbons to the TCA cycle intermediates is called the toluene degradation (tod) pathway [5,9,18]. The pathway consists of a total of seven enzymatic reactions [13]. Toluene dioxygenase (TDO), the first enzyme of the pathway, has been shown to catalyze the formation of respective cis-1,2-dihydroxy-1,2-dihydrodiol from each growth-supporting substrate [7-9]. The cis-dihydrodiol then undergoes to the TCA cycle intermediates such as pyruvate and acetyl-CoA through the tod pathway enzymes as shown in Fig. 1.

In this tod pathway the first four reaction steps (upper pathway) are critical for the 1-substituted benzenes to be catabolized to the TCA cycle intermediates since the rest three enzymatic reaction steps (lower pathway) uses the same intermediate regardless of the substitutions. Many 1-substituted benzenes such as monoalkylbenzenes [4,10,14,16], biphenyl [3,6], styrene [17] microorganisms using the enzyme systems similar to the tod pathway. However, the substrate specificity of the isofunctional enzymes in the pathways is sometimes different from each other and this confers discrete degradation pathways for specific chemicals. For instance, the tod pathway enzymes degrade biphenyl to the ring fission product of 6-phenyl-HOHD as a dead-end metabolite. When a bphD gene (equivalent to todF) from the biphenyl-growing P. pseudoalcaligenes KF707 was introduced to P. putida F1, the resultant strain was reported to grow on biphenyl [6]. This result showed that TodF functions as a critical bottleneck blocking further degradation of biphenyl by the tod pathway enzymes. Although extensive studies on substrate specificity of TDO have been carried out [9,11], much is not known about the substrate specificities of the next three enzymes in the tod upper pathway.

In this study, we have examined the possibility if the *tod* pathway enzymes could cometabolize other 1-substituted benzenes. Our primary interest was to determine how the structural changes in the side chain on the benzene ring affect their catabolism by the *tod* pathway enzymes and their inducibility of the *tod* operon. For this purpose we have chosen *n*-alkylbenzenes to change the chain length on the benzene ring and other chemicals such as biphenyl, cumene and styrene to change the functional group on the benzene ring. Our results show that the first three enzymes in the upper *tod* pathway have relaxed substrate specificities and the *tod* operon also has a relaxed effector specificity mechanism. The results obtained from this study could provide the basis for metabolic engineering of the *tod* pathway to degrade the non-growth supporting 1-substituted aromatic hydrocarbons to the TCA cycle intermediates.

Oxygen consumption by P. putida F1 cells grown on toluene. In order to determine whether n-alkylbenzenes (C3-C7), biphenyl, styrene and cumene could be degraded by the tod pathway enzymes, we have examined oxygen consumption by P. putida F1 cells grown on toluene in the presence of the chemicals. The resulting initial rate of the oxygen consumption could be determined primarily by TDO and the first reaction step is crucial for overall metabolism of the substrates by the rest tod pathway

enzymes. In addition to growth-supporting substrates (e.g. benzene, toluene and ethylbenzene), n-propylbenzene, n-butylbenzene, biphenyl, styrene and cumene showed high oxygen consumption. Furthermore, n-pentyl-, n-hexyl-, and n-heptylbenzene also showed low levels of oxygen consumption. However, n-octyl-, n-nonyl- and n-decylbenzene did not exhibit oxygen consumption.

Accumulation of dead-end metabolites from non-growth substrates by the tod To determine the metabolic barrier that prevents complete pathway enzymes. mineralization of the non-growth supporting 1-substituted benzenes, the accumulation of dead-end product from the chemicals was examined with P. putida F1 cells grown on Cell suspensions incubated with n-propylbenzene, n-butylbenzene were very toluene. yellow. However, cell suspensions incubated with n-alkylbenzenes (C5-C7) did not yielded visible color during the incubation. In addition, cell suspensions incubated with biphenyl, styrene and cumene showed deep yellow color. GC-MS analysis was conducted with organic extracts to identify the metabolites accumulated besides the The organic extracts ring-fission product. from the reaction mixtures of n-alkylbenzenes (C3-C7), biphenyl, styrene and cumene did not yielded GC peaks with molecular ions of M(OH)2+ of methane boronic acid and MO2+ which were cis-dihydrodiol and catechol products, respectively. However, organic extracts from n-alkylbenzenes (C5-C7) were yellow and yielded GC peaks at low levels (5 - 20%) with a molecular ion of MO⁺, which was benzylic monooxygenated products by showing the fragment peaks at m/z 79 and 107 from the mass analysis and which were confirmed by ¹H-NMR analysis. No products were identified from the same incubations with P. putida F1 cells grown on succinate during the incubation period. These results showed that the tod pathway enzymes recognized n-alkylbenzenes (C3-C7), biphenyl, styrene and cumene, and the substrates were metabolized to the meta-ring cleavage product, derivative of HOHD.

Biotransformation of n-alkylbenzenes by E. coli JM109(pDTG01A) expressing TDO. Although cis-dihydrodiol products from the growth-supporting n-alkylbenzenes (toluene and ethylbenzene) have been reported [7,8], detailed information on the reaction of TDO with n-alkylbenzenes with an alkyl chain longer than C3 is not available. E. coli JM109(pDTG601A) cells expressing TDO were incubated with n-alkylbenzenes for biotransformation. From the TLC, GC/MS and 1 H-NMR analyses, TDO from P. putida F1 was shown to catalyze the oxidation of n-alkylbenzenes (C3-C7) to

cis-2,3-dihydrodiol (major) and benzylic monooxygenated (minor) products with relative yields of 5-8 to 1 by integration of total ion current peak areas. In addition, TLC analysis showed that n-alkylbenzenes tested were oxidized to triols which might be formed by the dioxygenation reaction of monol compounds not by the monooxygenation reaction of cis-diols.

Inducibility of the *tod* operon by various chemicals. It has been previously shown that 2,3-dihydroxybiphenyl was a substrate for TodE and the yellow-colored *meta* ring fission product, 6-phenyl-HOHD, was not further metabolized by the subsequent enzyme, TodF [6]. This result suggested that to determine the expression level of the *tod* operon TodE could be a reporter enzyme, whose activity could be determined easily by monitoring accumulation of a yellow ring fission product. Resting cells grown on succinate supplemented with chemical inducers were subjected to the spectrophotometric assays. As expected, the growth-supporting substrates (e.g. benzene, toluene and ethylbenzene) has been proved to be strong inducers. Interestingly, non growth-supporting substrates such as *n*-propylbenzene, styrene and cumene also act as inducers for the *tod* operon even though they are not as efficient as the growth-supporting substrates.

To further verify the results obtained from the report enzyme assays, RT-PCR has been carried out. The results obtained from biotransformation described above indicates TodF appears to be critical for the catabolism of 1-substituted benzenes. For this reason, RT-PCR was carried out to detect transcription of the *tod* operon with amplification of *todF* gene. The RT-PCR result confirmed the role of each chemical in the induction of *tod* operon as determined by the TodE activity assays. The PCR products yielding ca 900 bp DNA fragments were observed in the cells grown on succinate supplemented with benzene, toluene, ethylbenzene, *n*-propylbenzene, styrene and cumene (Fig. 2). In addition, a very weak DNA product band was also observed in the cells grown on succinate supplemented with biphenyl. The RT-PCR DNA product formed from the cells supplemented with toluene was cloned in the pGEM-T easy vector and the product identity was confirmed by DNA sequencing.

Discussing conclusions. TDO has been shown to catalyze the formation of enantiomerically pure *cis*-1,2-dihydroxydihydrodiols from a variety of benzene derivatives [2,9,11]. The metabolites are an important class of compounds used as chiral synthons for the synthesis of value-added compounds and biologically active compounds such as

conduritols, inositols, pinitols, deoxysugar analogs and alkaloids [11]. This indicates that TDO is an industrially important enzyme and the studies on the substrate specificity of TDO are worth to continuing.

It has been reported early by Gibson and colleagues that *P. putida* 39/D, a *todD*-mutant of F1, oxidized ethylbenzene to *cis*-3-ethyl-3,5-cyclohexadiene-1,2-diol and (+)-1-phenylethanol and *cis*-3-(1'-hydroxyethyl)-3,5-cyclohexadiene-1,2-diol [7]. Our results shown here expanded the early result of the substrate specificity of TDO, that is TDO could recognize *n*-alkylbenzenes with an alkyl side chain of C3-C7 and catalyzes the formation of derivatives of monol, *cis*-diol and triols. *cis*-Dihydodiols from benzene and *n*-alkylbenzenes (C1-C5) by *P. putida* UV4 expressing TDO have been reported in a review article and the *cis*-dihydrodiol metabolites of higher homologs were not reported formed by the UV4 strain [2], indicating that there might be a size limitation on substrate acceptability between two TDOs.

It has been also examined in this study the metabolism of 1-substituted benzenes by the *tod* pathway of *P. putida* F1. The results show that the first three enzymes in the upper *tod* pathway has relaxed substrate specificities that accomodate the degradation of *n*-alkylbenzenes (C3-C7), styrene, cumene and biphenyl as well as the growth-supporting substrates. But the further degradation of the non-growth chemicals was blocked by TodF. This result is consistent with kinetic data obtained with purified TodF in which TodF has no detectable activity with 6-phenyl-HOHD and a negligible activity with 6-isopropyl-HOHD exhibiting 105-fold less activity compared with the growth substrate, 6-methyl-HOHD [15].

In addition, n-propylbenzene, styrene, cumene acted as inducers for the tod operon. For the inducer compounds to be growth substrates, TodF should be able to recognize their HOHD derivatives. In other report, the tod operon was induced by various other aromatic hydrocarbons including m-, p-xylene and water-soluble JP-4 jet fuel components [1]. Although TodS and TodT encoded by todST genes located at the end of the tod operon have been proposed to be involved in the transcriptional activation of the tod operon as two-component signal transduction pathway enzymes [12], the exact protein component for inducer binding has not been reported yet.

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$$\begin{array}{c} \text{CH}_3 \\ \text{TDO} \\ \text{NADH} \\ \text{O}_2 \\ \end{array} \begin{array}{c} \text{CH}_3 \\ \text{OH} \\ \text{TodD} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{CH}_3 \\ \text{OH} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{CH}_3 \\ \text{OH} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{CH}_3 \\ \text{COOH} \\ \text{OH} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{COOH} \\ \text{OH} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{COOH} \\ \text{OH} \\ \text{OH} \\ \end{array} \begin{array}{c} \text{COOH} \\ \text{OH} \\ \text{OH} \\ \end{array}$$

Fig. 1. The *tod* pathway in *P. putida* F1. Abbreviation for enzymes: TDO, toluene dioxygenase; TodD, *cis*-toluene dihydrodiol dehydrogenase; TodE, 3-methylcatechol dioxygenase; TodF, 6-methyl-HOHD hydrolase.



Fig. 2. RT-PCR amplification of todF mRNA transcripts purified from P. putida F1 grown in 5 mM succinate supplemented with each chemical on the abscissa. Abbreviations for chemicals: B0, benzene; B1, toluene; B2, ethylbenzene; B3, n-propylbenzene; B4, n-butylbenzene; B5, n-pentylbenzene; B6, n-hexylbenzene; B7, n-heptylbenzene; B8, n-octylbenzene; Bp, biphenyl; St, styrene; Cm, cumene. Chr represents a positive control in which todF gene was amplified by PCR from chromosomal DNA.

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