## Hydrogeochemical and environmental isotopic studies of ground waters in Pocheon area: application of M3 modeling to understand the hydrogeologic setting of regional ground water system for sustainable exploitation

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In Pocheon area, two distinct groups of bedrock ground water is recognized based on hydrogeochemical and environmental isotopic data. These are: (1) deep (170 to 1,004 m below the surface), 'sodium-bicarbonate' type water; and (2) shallow (20 to 250 m), 'calcium-bicarbonate' type water. The Na-HCO<sub>3</sub> type water has been exploited as H<sub>2</sub>S-bearing (<5 mg/l) thermal springs (up to 4 4°C) and shows hydrochemically the general attainment to equilibrium with host rock (granite). The Ca-HCO<sub>3</sub> type water varies seasonally in water quality and is high in NO<sub>3</sub> concentration. Hydrogeologic conditions constructed based on oxygen and hydrogen isotope data are distinct between the two groups: the deeper, Na-HCO3 type waters have the lower  $\delta^{18}O$  and  $\delta D$  values than Ca-HCO<sub>3</sub> type water, suggesting both their recharge in adjacent high altitudes and their longer and deeper flow paths. During the hydrogeochemical evolution from Ca-HCO<sub>3</sub> type toward Na-HCO<sub>3</sub> type waters, Na, Li, Cl, F, HCO<sub>3</sub> and H<sub>2</sub>S are progressively enriched, whereas Ca and SO<sub>4</sub> are decreased. Sulfur isotope data of dissolved sulfates suggest that the generation of H<sub>2</sub>S probably occurs by bacterial sulfate reduction at depths. Carbon isotope data of dissolved carbonates and chemical mass balance modeling indicate that hydrochemistry of Na-HCO<sub>3</sub> type water is largely controlled by the silicate hydrolysis and calcite dissolution followed by ion exchange with secondary silicates.

We performed M3 (Multivariate Mixing and Mass balance calculation) modeling (Laaksoharju et al., 1999) in order to explain the recognized spatial variation of water chemistry on the basis of hydrologic mixing among water masses as well as hydrogeochemical reactions. An ideal mixing model among three water masses is suggested, in which each water mass is represented by surface water (sample PCs-2), shallow ground water (PCw-1), and deep ground water (PCd-3). The resultant net chemical reactions also agree well with the results of chemical and isotopic mass balance modeling. We now propose regional hydrogeologic sections showing probable hydrogeologic setting in Pocheon area, which may be used for sustainable development of ground water.

**Keywords:** Pocheon, bedrock ground water, hydrogeochemistry, environmental isotopes, sulfur system, M3 modeling