

# Cubic and non-cubic 상태방정식의 LNG 물성 예측 비교

손주영, 최은주, 이태종  
경북대학교 화학공학과

## LNG thermodynamic properties prediction by use of non-cubic & cubic equation of state

Son Juyoung, Choi EunJoo, Lee TaeJong  
Dep. of Chem. Eng. Kyungpook National University

### 1. 서론

본 연구에서는 Hydrocarbon Systems의 물성 예측에 cubic 형태의 Peng Robin -son 상태방정식(PREOS)과 Non-cubic 형태의 Benedict-Webb-Rubin-Starling (BWRS)상태 방정식을 사용하여 그 결과를 비교하였다.

### 2. 이론

#### 1) PR-EOS

$$P = \frac{RT}{V-b} - \frac{\theta_{PR}}{V^2 + 2bV - b^2}$$

$$\theta_{PR} = a''[1 + (0.37464 + 1.54226\omega - 0.26992\omega^2)(1 - T_r^{1/2})]^2$$

$$a'' = \frac{0.45724R^2T_c^2}{P_c}, \quad b = \frac{0.07780R^2T_c^2}{P_c}$$

#### 2) BWRS-EOS

$$\begin{aligned} P = & \rho RT + (B_o RT - A_o - \frac{C_o}{T^2} + \frac{D_o}{T^3} - \frac{E_o}{T^4})\rho^2 + (bRT - a - \frac{d}{T})\rho^3 \\ & + \alpha(a + \frac{d}{T})\rho^6 + \frac{c\rho^3}{T^2}(1 + \gamma\rho^2)\exp(-\gamma\rho^2) \end{aligned}$$

11개의 순수물질 parameter는 다음 식으로 계산된다. (Han&Starling Correlation)

$$\begin{aligned}
\rho_{ci} B_{oi} &= A_1 + B_1 \omega_i & (\rho_{ci} A_o)/(R T_{ci}) &= A_2 + B_2 \omega_i \\
(\rho_{ci}^2 C_{oi})/(R T_{ci}^3) &= A_3 + B_3 \omega_i & \rho_{ci}^2 \gamma_i &= A_4 + B_4 \omega_i \\
\rho_{ci}^2 b_i &= A_5 + B_5 \omega_i & (\rho_{ci}^2 a_i)/(R T_{ci}^3) &= A_6 + B_6 \omega_i \\
\rho_{ci}^3 a_i &= A_7 + B_7 \omega_i & (\rho_{ci}^2 c_i)/(R T_{ci}^3) &= A_8 + B_8 \omega_i \\
(\rho_{ci} D_{oi})/(R T_{ci}^4) &= A_9 + B_9 \omega_i & (\rho_{ci}^2 d_i)/(R T_{ci}^2) &= A_{10} + B_{10} \omega_i \\
(\rho_{ci} E_{oi})/(R T_{ci}^5) &= A_{11} + B_{11} \omega_i \exp(-3.8 \omega_i)
\end{aligned}$$

The Bishnoi-Robinson Mixing Rules(BR MR)

$$\begin{aligned}
B_{oij} &= \sum_{i=1}^n x_i B_{oi} \\
A_o &= \sum_{i=1}^n \sum_{j=1}^n x_i x_j A_{oi}^{\frac{1}{2}} A_{oj}^{\frac{1}{2}} (1 - k_{ij}) \\
C_o &= \sum_{i=1}^n \sum_{j=1}^n x_i x_j C_{oi}^{\frac{1}{2}} C_{oj}^{\frac{1}{2}} (1 - k_{ij})^3 & \gamma &= \left[ \sum_{i=1}^n x_i \gamma_i^{\frac{1}{2}} \right]^2 \\
b &= \left[ \sum_{i=1}^n x_i b_i^{\frac{1}{3}} \right]^3 & a &= \left[ \sum_{i=1}^n x_i a_i^{\frac{1}{3}} \right]^3 \\
a &= \left[ \sum_{i=1}^n x_i a_i^{\frac{1}{3}} \right]^3 & c &= \left[ \sum_{i=1}^n x_i c_i^{\frac{1}{3}} \right]^3 \\
D_o &= \sum_{i=1}^n \sum_{j=1}^n x_i x_j D_{oi}^{\frac{1}{2}} D_{oj}^{\frac{1}{2}} (1 - k_{ij})^4 & d &= \left[ \sum_{i=1}^n x_i d_i^{\frac{1}{3}} \right]^3 \\
E_o &= \sum_{i=1}^n \sum_{j=1}^n x_i x_j E_{oi}^{\frac{1}{2}} E_{oj}^{\frac{1}{2}} (1 - k_{ij})^4
\end{aligned}$$

Modified van der Waals one fluid Mixing Rules(CST MR)

$$\begin{aligned}
\sigma_x^3 &= \sum_{\alpha} \sum_{\beta} x_{\alpha} x_{\beta} \sigma_{\alpha\beta}^3 \\
\epsilon_x \sigma_x^3 &= \sum_{\alpha} \sum_{\beta} x_{\alpha} x_{\beta} \epsilon_{\alpha\beta} \sigma_{\alpha\beta}^3 \\
\gamma_x \epsilon_x^2 \sigma_x^3 &= \sum_{\alpha} \sum_{\beta} x_{\alpha} x_{\beta} \gamma_{\alpha\beta} \epsilon_{\alpha\beta}^2 \sigma_{\alpha\beta}^3 \\
\sigma_{\alpha\beta} &= \xi_{\alpha\beta} \sqrt{\sigma_{\alpha\alpha} \sigma_{\beta\beta}} \\
\epsilon_{\alpha\beta} &= \zeta_{\alpha\beta} \sqrt{\epsilon_{\alpha\alpha} \epsilon_{\beta\beta}} \\
\gamma_{\alpha\beta} &= \frac{1}{2} (\gamma_{\alpha\alpha} + \gamma_{\beta\beta})
\end{aligned}$$

BWRS 상태방정식에 두가지 다른 Mixing Rules를 사용하여 밀도와 액상조성에 대한 예측성을 비교하여 Table 1에 나타내었다.

Table 1. Comparison of prediction for hydrocarbon systems.  
systems.

Binary Systems	EOS	A.A.D.(%) Density	No. of Data points	A.A.D.(%) VLE ( $x_i$ )	No. of Data points
$\text{CH}_4/n\text{-C}_4\text{H}_{10}$	CST MR	1.6	6	9.2	14
	BR MR	4.0	6	11.8	14
$\text{C}_2\text{H}_8/n\text{-C}_5\text{H}_{12}$	CST MR	2.4	12	8.49	12
	BR MR	2.5	12	8.51	12
$\text{C}_3\text{H}_{12}/n\text{-C}_6\text{H}_{14}$	CST MR			16.3	10
	BR MR			12.7	10
$\text{C}_4\text{H}_{15}/n\text{-C}_7\text{H}_{16}$	CST MR	3.4	12	23.5	10
	BR MR	2.6	12	31.5	10

Non-cubic 형태의 BWRS 상태방정식으로 hydrocarbon 이성분계의 물성을 예측한 결과, cubic식인 PREOS에 비해 전반적으로 예측성이 뛰어남을 알 수 있었다. 다성분계에 대한 연구가 더 진행되어야 할 것이다.

### 3. 참고 문헌

1. P. Y. Tan and K. D. Luks, "Evaluation of the conformal solution theory of mixing.", J. Chem. Soc., Vol. 52, No. 6, Pg. 3091-3096 (1970)
2. Marcla L. Huber and James F. Ely, "Improved conformal solution theory for Mixtures with large size ratios.", Fluid Phase Equilibria, Vol. 37, Pg. 105-121 (1987)
3. K. E. Starling, "Fluid thermodynamic properties for light petroleum system", Gulf. Pub. Co.(1973)
4. W. R. Smith, "Perturbation theory and one-fluid corresponding state theories for fluid mixtures", Can. J. Chem. Eng., 50, pp. 271, (1972).
5. K. E. Starling, "Fluid thermodynamic properties for light petroleum system", Gulf Pub. Co., (1973).
6. M. R. Brule, C. T. Lin, L. L. Lee and K, E, Starling, "Multi parameter corresponding-state correlation of coal-fluid thermodynamic properties", AIChE J., 28, 4, pp. 616, (1982).